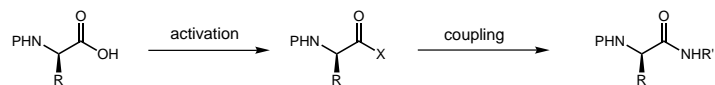


Advances in Peptide Coupling



Lead References:

K. Devries, Evans group Friday seminar, 4/9/91

L. Carpino *Methods in Enzymology* **1997**, 289, 104

Evans Group Seminar

Jeff Katz

12/15/98

Overview

1. The role of DMAP

2. Amide formations

- i. Carbodiimides
- ii. Mixed anhydrides of carbon
- iii. Acyl azides & cyanides
- iv. Mixed anhydrides of phosphorus
- v. Pentafluorophenyl esters
- vi. Phosphonium & aminium reagents
- vii. Acid fluorides
- viii. Obscurities

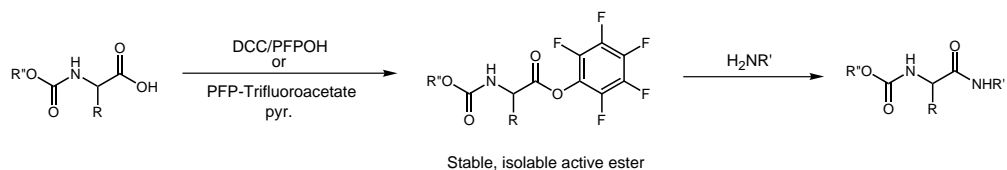
3. The coupling of hindered amino acids

4. Complex peptide synthesis:

- i. Didemnin
- ii. Microcystin LA

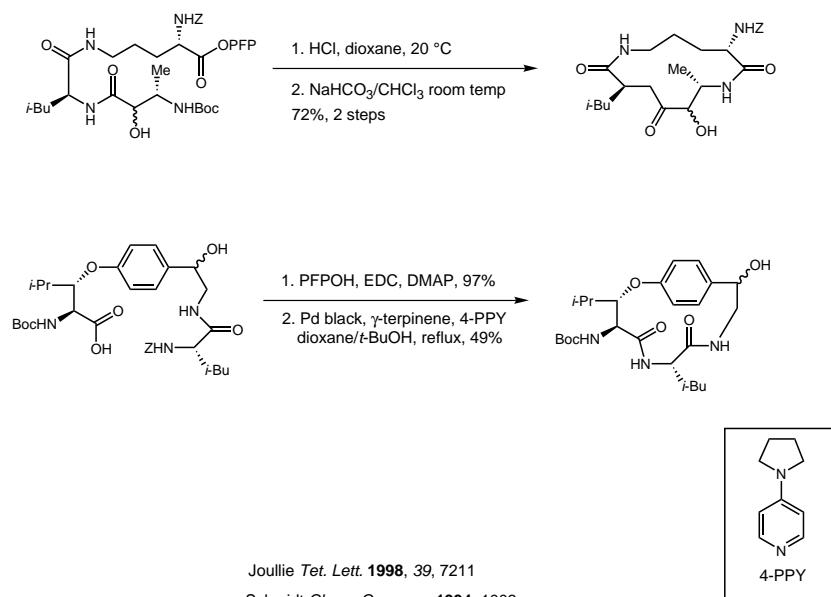
Pentafluorophenyl Esters

Preactivation method:



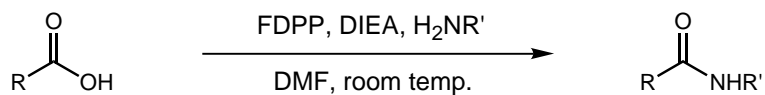
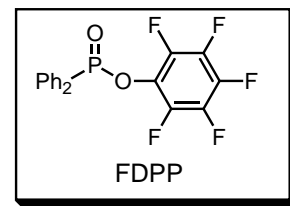
Green *Tet. Lett.* **1990**, 31, 5851

Pentafluorophenyl Esters

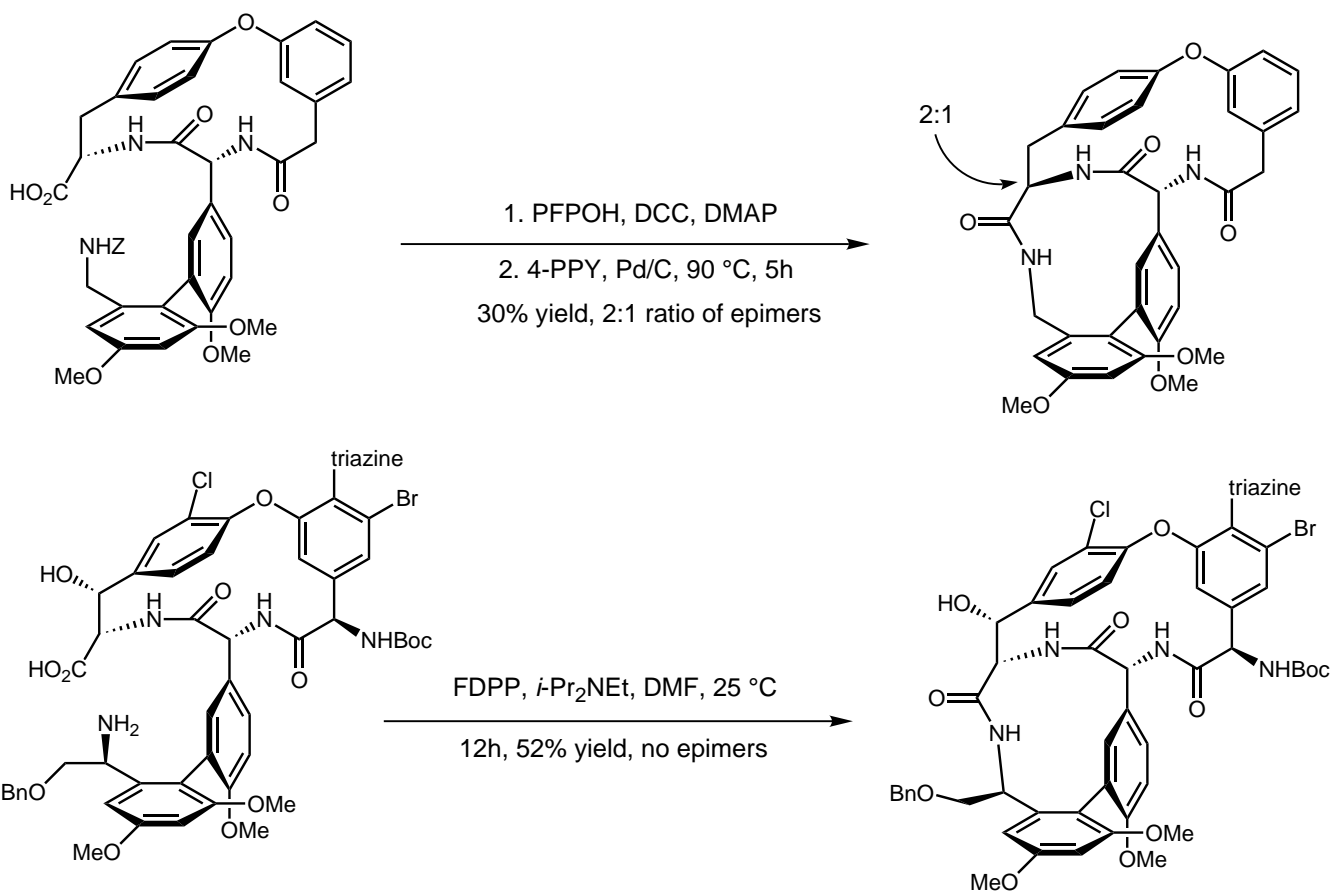


Joullie *Tet. Lett.* **1998**, 39, 7211
Schmidt *Chem. Commun.* **1994**, 1003

Pentafluorophenyl Esters



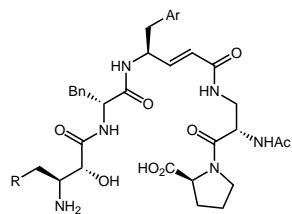
Xu *Tet. Lett.* **1991**, 32, 6711



K. C. Nicolaou *Chem. Commun.* **1997**, 1899

K. C. Nicolaou *Angew. Chem. Int. Ed.* **1998**, 37, 2708

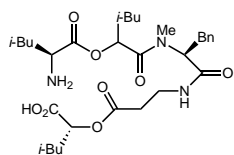
Pentafluorophenyl Esters



Shioiri *Tet. Lett.* **1996**, 37, 2261

Coupling agent	% Yield
EDC/HOBt	16
DPPA	8
DEPC	12
PFP ester	8
FDPP [*]	57

^{*}FDPP, *i*-Pr₂NEt, DMF, r.t. 14h



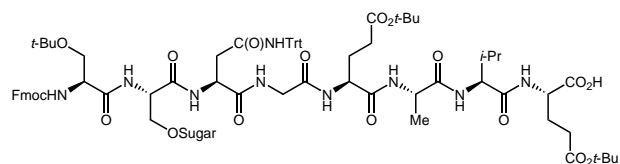
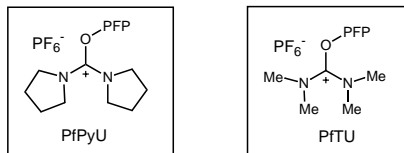
Ulrich Schmidt *Chem. Commun.* **1994**, 2381

Coupling agent	% Yield
DPPA	low
FDPP	60
HATU	50
PFP ester [*]	85

^{*}CHCl₃-NaHCO₃, room temp. 5h

Pentafluorophenyl Esters

The "first" *In situ* activation method:



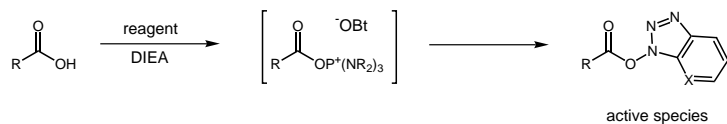
Synthesized in 13 steps on solid phase (tentagel) in an overall yield of 76% (98%/step):
(Hunig's base, collidine, PfpPyU, NMP)

Habermann *J. Prakt. Chem.* **1998**, 340, 233

Phosphonium Reagents



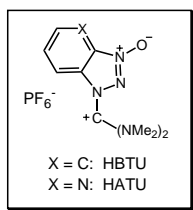
Highly active coupling reagents with "built-in" additive:



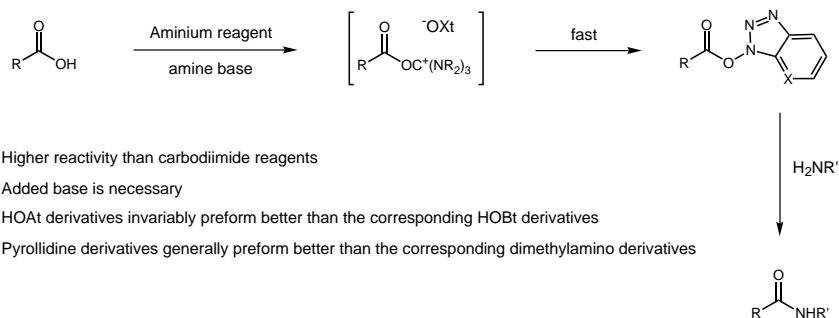
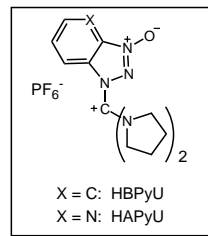
- HOAt derivatives invariably perform better than the corresponding HOBt derivatives
- Pyrrolidine derivatives generally perform better than the corresponding dimethylamino derivatives

Castro *Tet. Lett.* **1975**, 1219; *Tet. Lett.* **1990**, 31, 205

Carpino *Chem. Commun.* **1994**, 201



Aminium Reagents



Knorr *Tet. Lett.* **1989**, 1927
Carpino *J. Am. Chem. Soc.* **1993**, 115, 4397

Aminium Salt Couplings — Effects of the Amine Base

Aminium Salt	DIEA	NMM	Pyridine	collidine	lutidine	DMAP
HBPYU	10.7	5.7	7.7	0.6		
HATU		2.7		0.1		
HAPYU	2.3	1.7	6.2	0.1	0.6	10.4

% Racemization for 2+1 segment coupling of Z-Phe-Val-Ala-OMe in DMF

HBPYU	26.6			13.7		
HATU	6.5			1.7		
HAPYU	6.3	5.9		0.1	2.3	

% Racemization for 2+1 segment coupling of Z-Phe-Val-Pro-OtBu in DMF

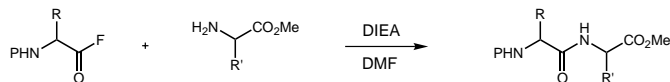
Aminium Salt	Additive	NMM (2 equiv)
HAPYU	none	12.2
HAPYU	HOAt	3.3

% Racemization for 2+1 segment coupling of Z-Phe-Val-Pro-NH₂ in DMF

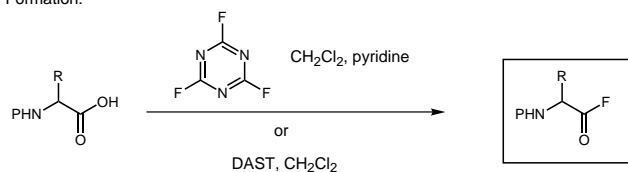
- Added HOAt/HOBt is shown to have a deleterious effect for solid phase coupling reactions

Carpino *J. Org. Chem.* **1994**, 59, 695
Carpino *J. Org. Chem.* **1996**, 61, 2460

Amino Acid Fluorides

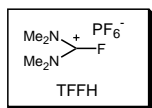


Acid Fluoride Formation:

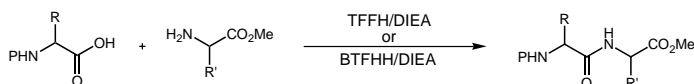
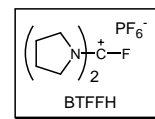


- Bench stable compounds for most Fmoc-protected amino acids
- Do not form oxazolone intermediates in the presence of tertiary amines
- Relatively unreactive to neutral oxygen nucleophiles
- React readily with anionic oxygen nucleophiles and neutral amines
- Silicon protecting groups are problematic

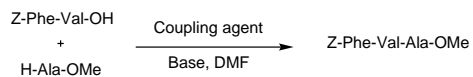
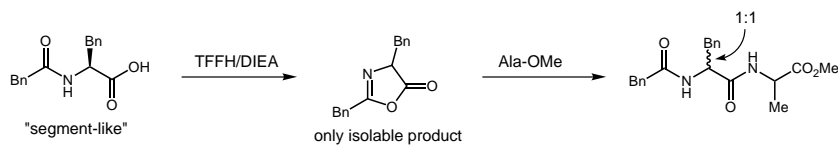
Carpino *Acc. Chem. Res.* **1996**, 29, 268



Amino Acid Fluorides — *In Situ* Generation



For segment condensations, HOAt is needed to suppress racemization:

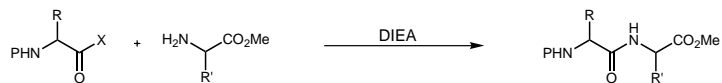


% Racemization:

	<i>i</i> -Pr ₂ NEt	NMM	collidine
TFFH	25	23	6
TFFH/HOAt	2		<0.1

Carpino *J. Am. Chem. Soc.* **1995**, 117, 540
 Carpino *Chem. Lett.* **1998**, 671

Amino Acid Fluorides



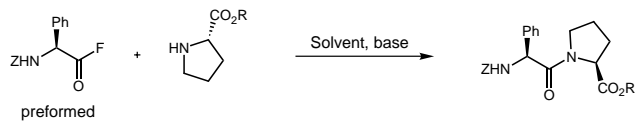
X	equiv of DIEA	% Conversion
F	0	100*
F	1	100
Cl	0	50
Cl	1	100

* Reaction rate is only slightly retarded

- No-base approach has been shown to be general for both solid- and solution-phase couplings
- No epimerization is observed

Carpino *Chem. Commun.* **1995**, 669

Acid Fluorides — Racemization



Solvent	base	% epimer
DMF	DIEA	19.0
DMF	collidine	11.6
CH ₂ Cl ₂	collidine	<0.1

- Highly racemization-prone amino acids still require carefully controlled conditions

Wenschuh, Carpino *J. Org. Chem.* **1995**, 60, 405

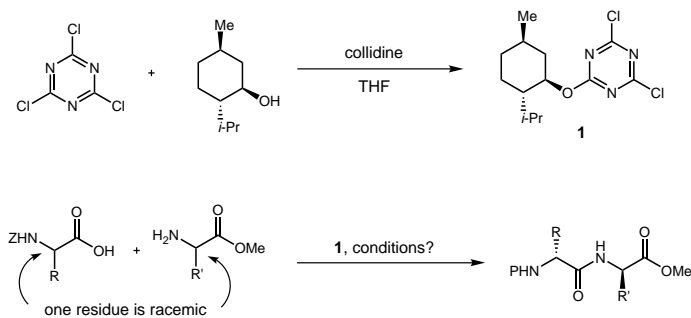
Segment Couplings

Z-Phe-Val-Pro-NH₂ via 2+1 segment coupling in DMF

Reagent	Base (equiv)	% Yield	% epimer	Reagent	Base (equiv)	% Yield	% epimer
EDC/HOAt		85	4.7	HATU/HOAt	DIEA (2)	76	10.9
EDC/HOBt		87	18.9	HATU/HOAt	DIEA (3)	88	15.8
EDC/HODhbt		89	7.3	HATU/HOAt	TMP (2)	72	2.4
EDC•HCl/HOAt	TMP (1)	89	5.3	HATU/HOAt	TMP (3)	87	4.5
EDC•HCl/HOBt	TMP (1)	87	19.9	HAPyU	TMP (2)	87	3.5
HATU	TMP (2)	83	5.3	HAPyU/HOAt	TMP (2)	76	1.6
HATU	DIEA (2)	86	13.9	HAPyU/HOAt	TMP (3)	89	2.3
HBTU	TMP (2)	81	14.2	HAPyU/HOAt	DIEA (2)	77	3.2
BOP	TMP (2)	81	13.9	HAPyU/HOAt	DIEA (3)	90	12.1

Carpino *J. Org. Chem.* **1995**, *60*, 3561

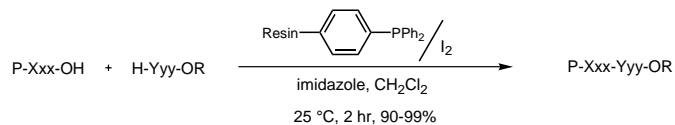
Obscurities — Couplings with kinetic resolution



Coupling partners (substrate ratio)	% Yield	ee or de
Z-Gly-OH + D,L-Ala-OEt (1:2)	87	93 (L)
Z-D,L-Ala-OH + Gly-OEt (2:1)	86	100 (L)
Z-D,L-Ala-OH + L-Leu-OMe (2:1)	72	100 (L,L)
Z-D,L-Ala-OH + L-Phe-OEt (2:1)	75	100 (L,L)

• %ee measured by rotation only

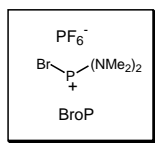
Obscurities — Polymer-bound coupling reagents



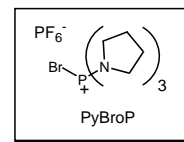
P	Xxx	Yyy	R	% Yield
Fmoc	Ala	Leu	Allyl	99
Cbz	Met	Ala	Me	98
Boc	Leu	Phe	<i>t</i> -Bu	95
Fmoc	Val	Val	Allyl	99
Fmoc	Cys(Trt)	Cys(Trt)	Me	95
Boc	Trp	Leu	Me	94

• Authors propose an acyl iodide as the active species

Caputo et. al. *Synthesis* **1995**, 141



Secondary Amine Coupling



More Powerful activation is needed for Secondary Amines:

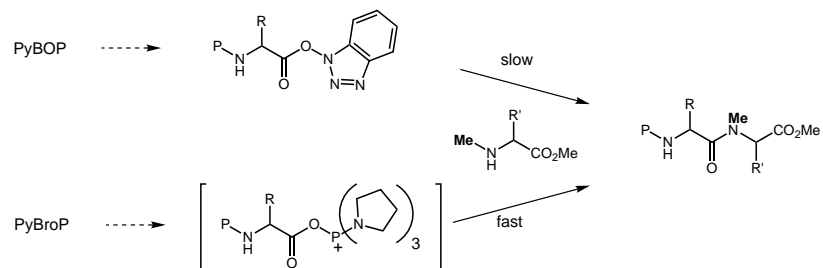


Sequence	% Yield (hours)	
	PyBOP	PyBroP
Z-MeVal-Val-OMe	90 (1)	85 (1)
Z-Val-MeVal-OMe	11 (1)	70 (1)
Fmoc-Val-MeVal-OMe	30 (24)	84 (3)
Boc-Val-MeVal-OMe	45 (24)	44 (3)

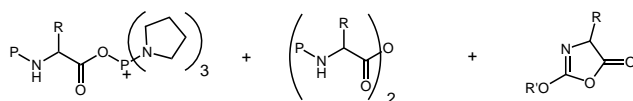
CH₂Cl₂, DIEA, room temp.

Coste *J. Org. Chem.* **1994**, *59*, 2437

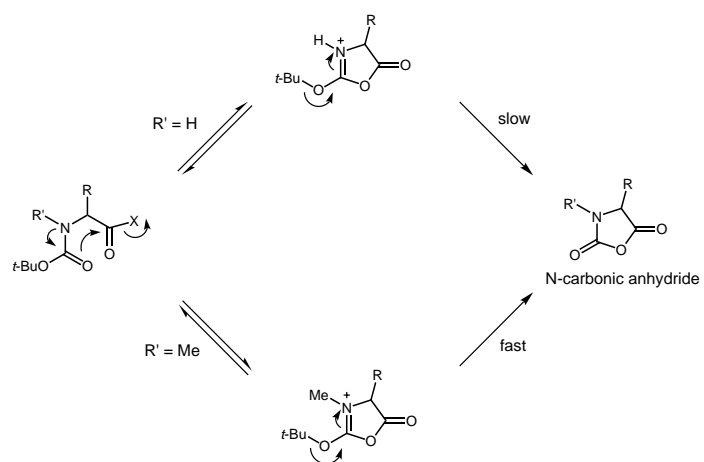
Phosphonium Reagents — Active Coupling Species



For PyBroP, The "active ester" is a mixture of species:

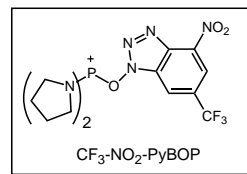


Secondary Amine Coupling: Side-product Pathways



Secondary Amine Couplings

An electron-poor HOBt additive can increase reactivity:

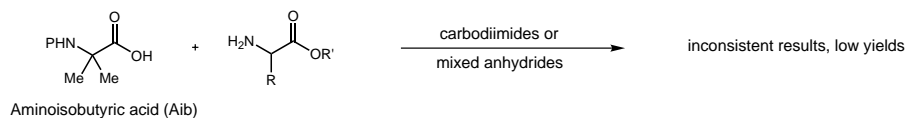


Sequence	% Yield	
	PyBroP	CF ₃ -NO ₂ -PyBOP
Z-Val-Val-OMe	89	98
Z-Val-MeVal-OMe	49	76
Fmoc-Val-MeVal-OMe	57	85
Boc-Val-MeVal-OMe	25	62
Boc-MeLeu-MeLeu-OMe	47	87
Z-MeVal-MeVal-OMe	22	71

CH₂Cl₂, DIEA, room temp., 1 hr

Bloemhoff *Tet. Lett.*, **1995**, 36, 4643

α,α-Dialkylamino Acids



% Yield for Phosphonium-based Aib couplings (CH₂Cl₂, room temp.)

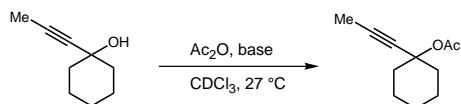
Sequence	BOP	PyBOP	BroP	PyBroP
Z-Aib-Gly-OEt	92	87	89	87
Z-Aib-Val-OMe	88		87	84
Z-Aib-Pro-Ot-Bu	82		84	95
Z-Aib-Aib-OMe		89		77
Boc-Aib-Aib-OMe	80	86	25 76*	25 77*

* With added DMAP

- Aib-Aib coupling is still sluggish — reaction times are 16 to 24 hours
- While good yields can be obtained, sluggish and varied reaction conditions make this unsuitable for solid phase

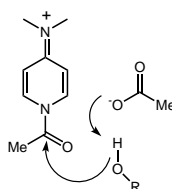
Coste *Tetrahedron* **1991**, 47, 259

DMAP Acylation Catalysis



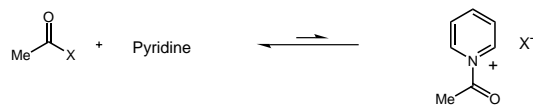
Base	Acylation Reagent	t 1/2 (min)
DMAP	Ac_2O	7
DMAP	AcCl	20
pyridine	Ac_2O	1000
1:1 pyridine: NEt_3	Ac_2O	1000

General Base Acceleration:



Steglich *AC/EE* 1978, 17, 569

DMAP — Equilibrium Concentrations



• Undetectable in non-polar solvents, but has been detected in H_2O by UV



X	% Pyridinium in solution
Cl	>95
OAc	5-10

CDCl_3 , room temperature

α,α -Dialkylamino Acids

Aminium salts and acid fluorides for solid-phase α,α -dialkyl and *N*-methylamino acid synthesis:

H-Tyr-**Aib-Aib**-Phe-Leu-NH₂

Reagent	sequence purity
HATU	94%
TFFH	92%
HBTU	43%

7 min. activation, 30 min. coupling in DMF

H-D-Ala-**MeLeu-MeLeu-MeVal**-Phe-Val-OH

Reagent	sequence purity
HATU	85%
HBTU	8%

7 min. activation, 30 min. coupling in DMF

Carpino *Chem. Commun.* **1994**, 201

Peptaibols

- Bioactive linear peptide sequences
- Sequences are approx. 20 residues and contain unusually high amounts of α,α -dialkylamino acids
- Aib-Pro linkages are acid-sensitive (cleaved by TFA)

Ac-**Aib-Pro-Aib-Ala-Aib-Ala-Gln-Aib-Val-Aib-Gly-Leu-Aib-Pro-Val-Aib-Aib-Glu-Gln-Pheol**
alamethicin F-30

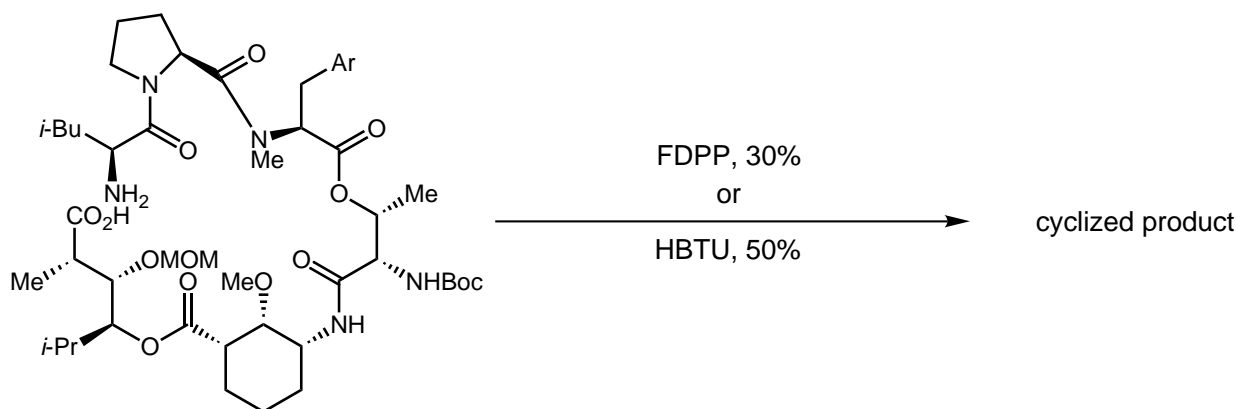
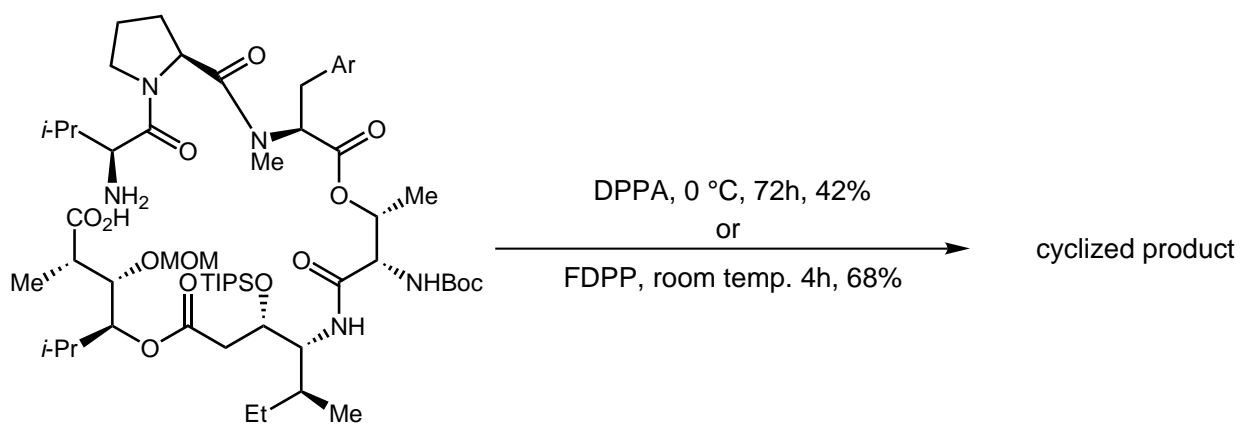
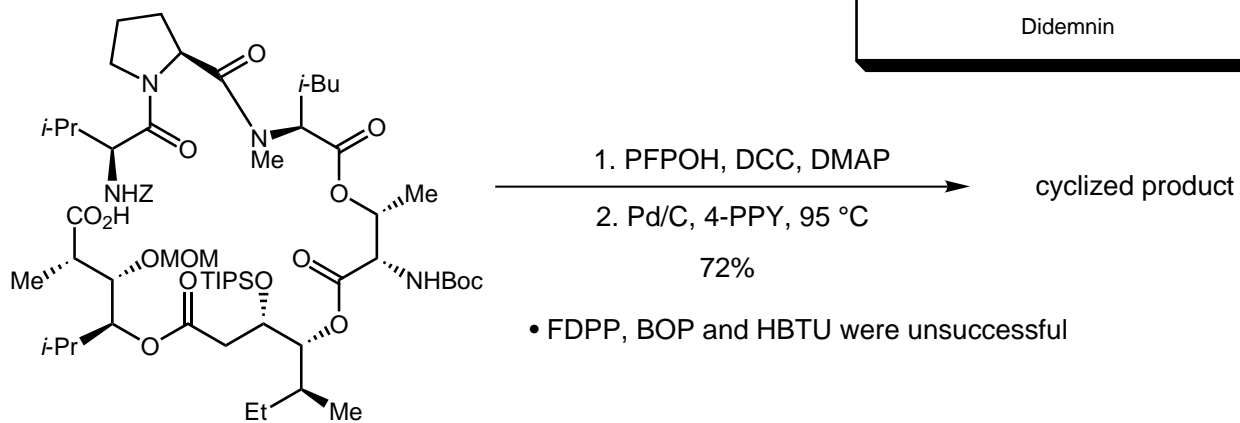
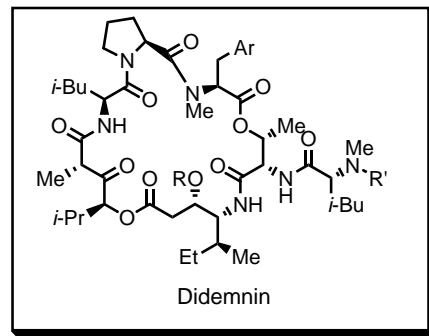
Synthesized on solid phase *via* Fmoc-amino acid fluorides in 78% purity

Ac-**Aib-Ala-Aib-Leu-Aib-Gln-Aib-Aib-Aib-Ala-Aib-Aib-Pro-Leu-Aib-Iva-Gln-Valol**
trichotoxin A-50

Synthesized on solid phase *via* Fmoc-amino acid fluorides in 60% purity

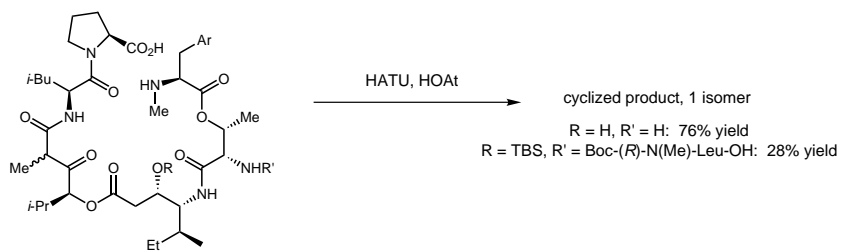
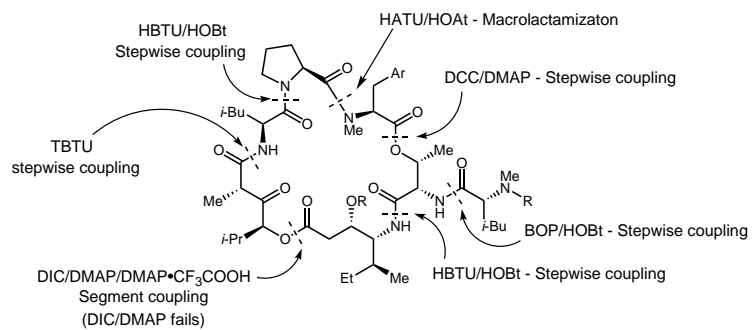
Wenschuh, Carpino *J. Org. Chem.* **1995**, 60, 405

Didemnin

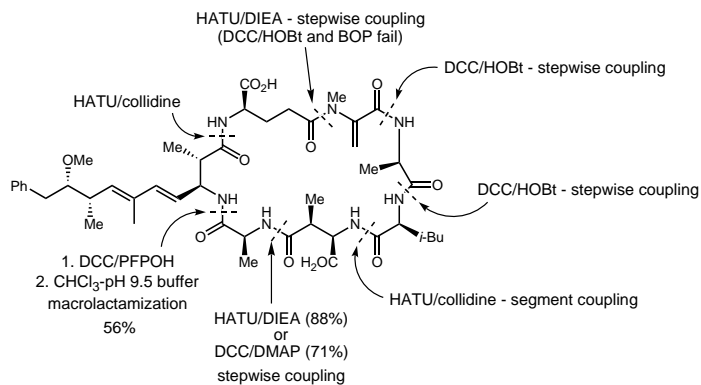


Joullie *J. Org. Chem.* **1994**, *59*, 5192
Joullie *Bioorg. Med. Chem. Lett.* **1996**, *6*, 2713
Joullie *J. Org. Chem.* **1996**, *61*, 1655

Didemnin



Microcystin LA



Richard Chamberlin *J. Am. Chem. Soc.* **1996**, *118*, 11759

Overview

Stepwise couplings:

Standard amino acids — EDC/HOBt. DEPC, or BOP are also good choices

Racemization-prone amino acids — If EDC/HOBt is insufficient, try EDC/HOAt or HATU/collidine

Segment couplings:

Phosphonium or aminium salts are good choices, HOAt/pyrrolidine derivatives for the difficult cases

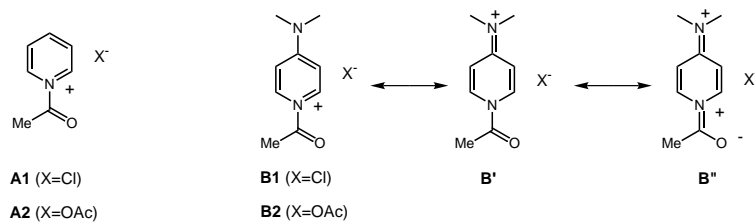
Macrolactamizations:

FDPP/PFP esters, DPPA and HAPyU/collidine are good bets

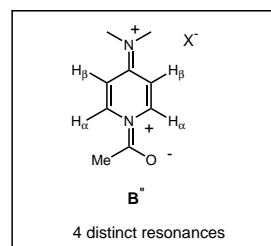
Hindered couplings:

Acid fluorides and HAPyU/collidine are good choices. PyBroP and BOP-Cl work for solution phase.

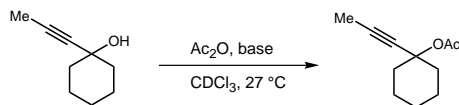
DMAP Acylation Catalysis



B1 and **B2** are observable by ^1H NMR in CDCl_3 or CD_2Cl_2 at low temperature



DMAP Acylation Catalysis



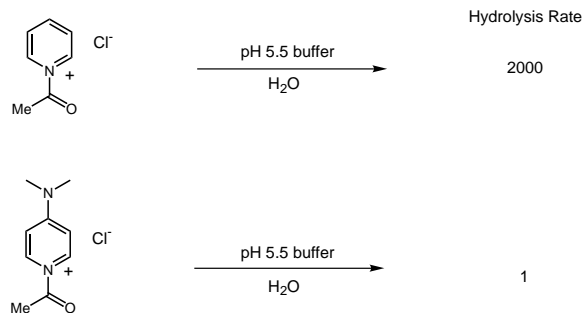
Base	Acylation Reagent	t 1/2 (min)
DMAP	Ac_2O	7
DMAP	AcCl	20
pyridine	Ac_2O	1000
1:1 pyridine: NEt_3	Ac_2O	1000

• Rate of pyridinium-acylations is tied to:

- Concentration/solubility
- Identity of counterion
- Tightness of ion pair

DMAP Acylation Catalysis

A Reactivity Reversal:



• Rate of pyridinium-acylations is tied to:

- Tightness of ion pair
- Concentration/solubility
- Identity of counterion

Overview

1. The role of DMAP

2. Amide formations

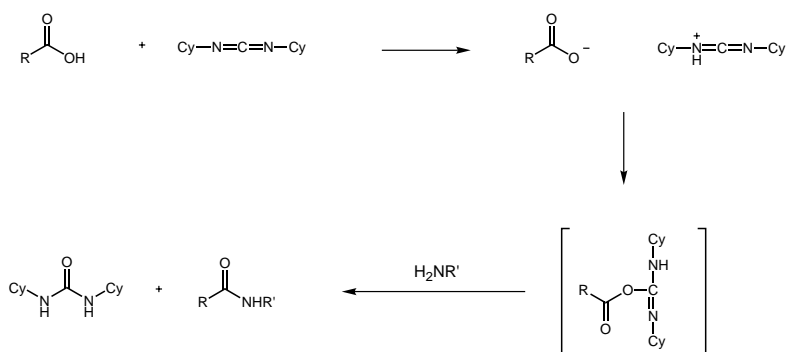
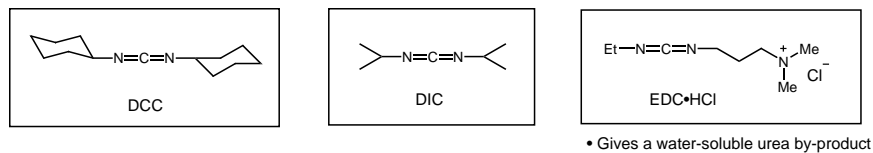
- Carbodiimides
- Mixed anhydrides of carbon
- Acyl azides & cyanides
- Mixed anhydrides of phosphorus
- Pentafluorophenyl esters
- Phosphonium & aminium reagents
- Acid fluorides
- Obscurities

3. The coupling of hindered amino acids

4. Complex peptide synthesis:

- Didemnin
- Microcystin LA

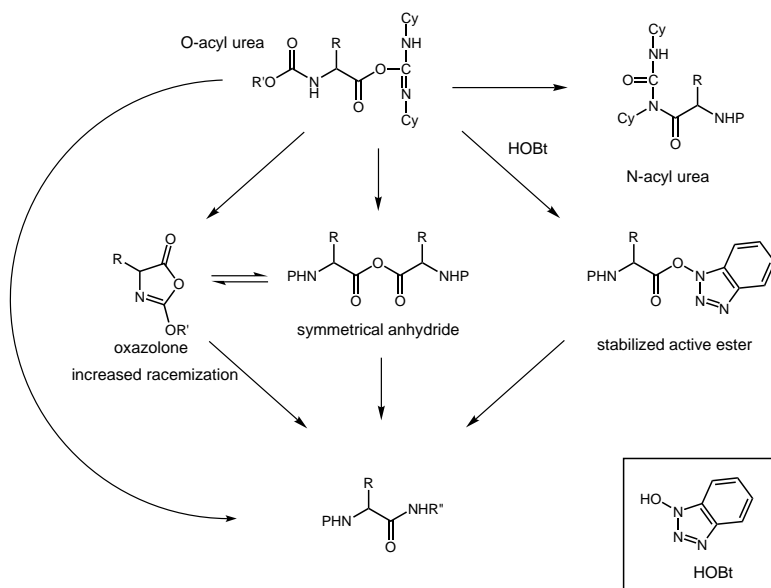
Carbodiimide Activation



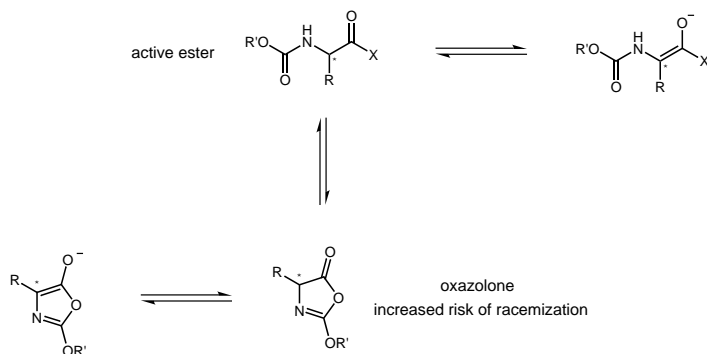
Sheehan *J. Am. Chem. Soc.* **1955**, *77*, 1067

Rebek *J. Am. Chem. Soc.* **1973**, *95*, 4052

Carbodiimides — Reaction Pathways

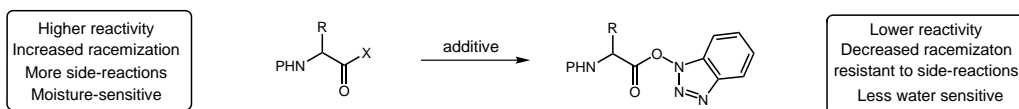


Racemization Pathways

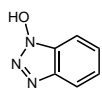


- Direct deprotonation of active esters does occur, but oxazolone formation is the major racemization pathway
- Carbamate-protected amines help suppress oxazolone formation and deprotonation
- Amide "amino-protection" is the main reason for increased racemization in segment couplings

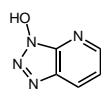
Modern Advances in Coupling Additives



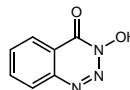
Most common coupling additives:



HOBt



HOAt

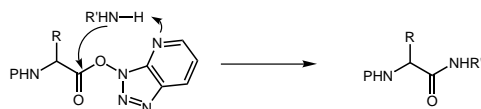


HODhbt

- Decreased racemization vs. HOBt but more side-reactions

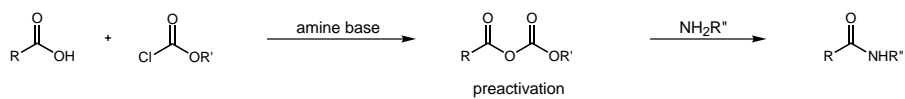
Z-Phg-Pro-NH₂ formation in DMF:

Reagent	% Yield	% epimer
EDC/HOAt	81	7
EDC/HOBt	80	25
EDC/HODhbt	82	10

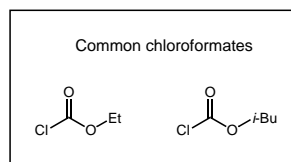


- HOAt has the advantage of internal base acceleration

Mixed Anhydrides of Carbon



Amine	equiv	% Yield	% epimer
NMe ₃	1	90	0
NMe ₃	2	<5	68
NEt ₃	1	82	8
NEt ₃	2	59	16
<i>i</i> -Pr ₂ NEt	1	3	0.2
<i>i</i> -Pr ₂ NMe	1	94	0
<i>i</i> -Pr ₂ NMe	2	85	3
NMM	1	92	0
NMM	2	93	0



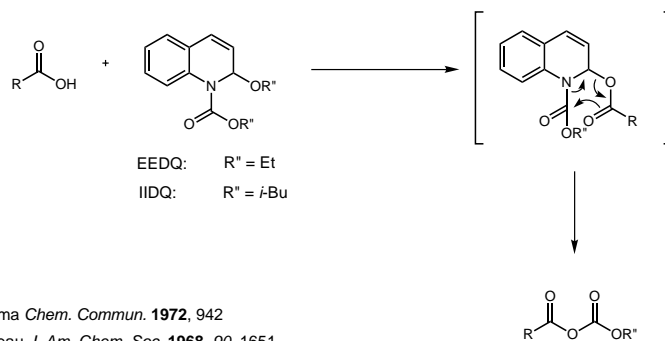
Z-Gly-Phe-Gly-OEt synthesis in THF at -15 °C,
12 min activation with *i*-BuOC(O)Cl

- The solvents of choice are ethyl acetate and THF. Acetonitrile, CH₂Cl₂, NMP and DMF should be avoided.
- Side reactions: disproportionation to symmetrical anhydride, attack at undesired carbonyl

Anderson *J. Am. Chem. Soc.* **1967**, *89*, 5012
Jouin *Tetrahedron* **1989**, *45*, 5039

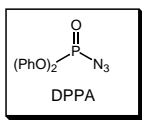
Mixed Anhydrides of Carbon

In situ generation (no preactivation) of mixed anhydrides:



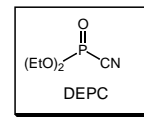
Yajima *Chem. Commun.* **1972**, 942
Belleau *J. Am. Chem. Soc.* **1968**, *90*, 1651

- No additional base is necessary

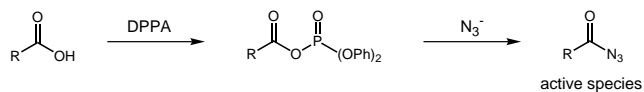


- Tends to be used for macrocyclizations

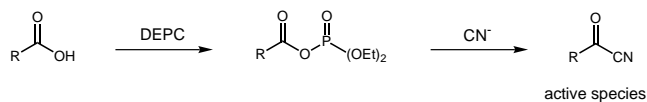
DPPA and DEPC



- Tends to be used for stepwise couplings

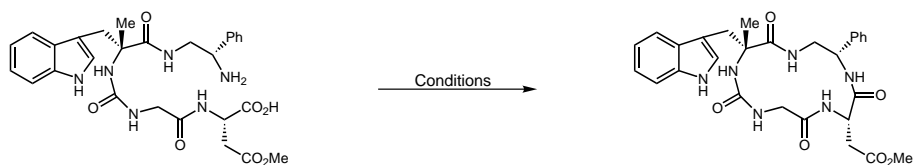


- Curtius rearrangement is slow relative to coupling
- Acyl azide relatively unreactive to non-amine nucleophiles



Yamada *J. Am. Chem. Soc.* **1972**, *94*, 6203
Yamada *Tet. Lett.* **1973**, 1595

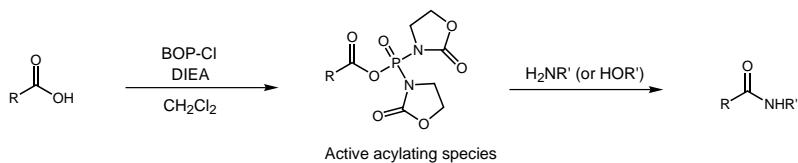
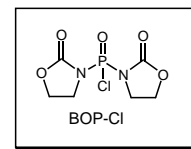
DPPA



Method	% Yield
DPPA	71
BOP	49
DCC, HOBT	31

Pritchard *Tetrahedron* **1992**, *48*, 8471

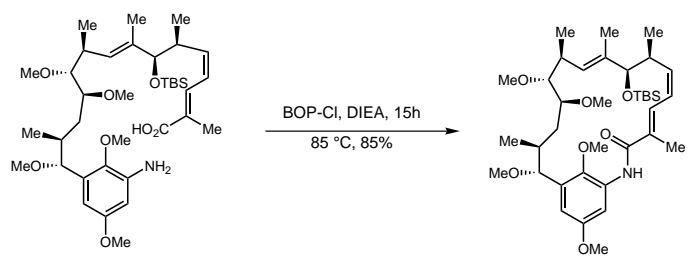
BOP-Cl Activation



- Mixed anhydride intermediate is very reactive and will form both esters and amides
- Less prone to side-reactions than chloroformate-based anhydrides
- Active enough to couple N-alkyl amino acids, but requires long reaction times

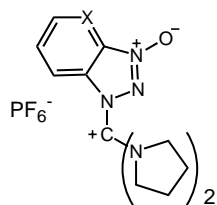
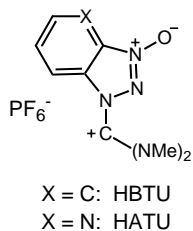
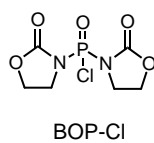
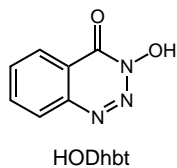
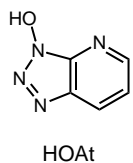
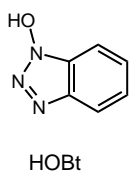
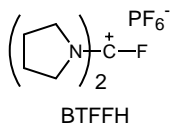
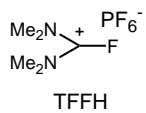
A. Palomo-Coll *Synthesis* **1980**, 547
Daniel Rich *J. Am. Chem. Soc.* **1985**, 107, 4342
Daniel Rich *J. Org. Chem.* **1986**, 51, 3350

BOP-Cl Macrolactamization



DCC, DEPC, DPPA fail

Baker *Chem. Commun.* **1989**, 378



X = C: HBTU
X = N: HATU

