

Chemistry 206

Advanced Organic Chemistry

Handout-36A

***Intramolecular Enone-Olefin
Photocycloadditions Directed Toward Natural
Product Synthesis***

Travis Dunn

Evans Group Seminar, March 31, 2000

DAE Group Friday Afternoon Seminar

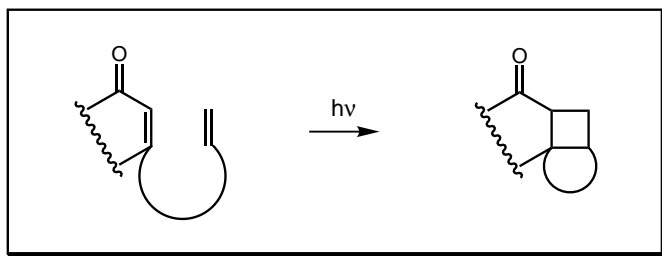
March 31, 2000

D. A. Evans

Monday
September 27, 1999

Intramolecular Enone-Olefin Photocycloadditions Directed Toward Natural Product Synthesis

Travis Dunn
DAE Group Friday Afternoon Seminar
March 31, 2000



Lead References:

- Intramolecular Enone-Olefin Photocycloaddition
Crimmins, M.T. *Chem. Rev.* **1988**, *88*, 1453-1473.
- Cycloaddition/Fragmentation Strategies in Synthesis
Winkler, J.D., *et al.* *Chem. Rev.* **1995**, *95*, 2003-2020.
- Mechanism of Enone-Olefin Photocycloaddition
Schuster, D.I., *et al.* *Chem. Rev.* **1993**, *93*, 3-22.

Introduction and Scope

I) Mechanistic considerations of the enone-olefin photocycloaddition

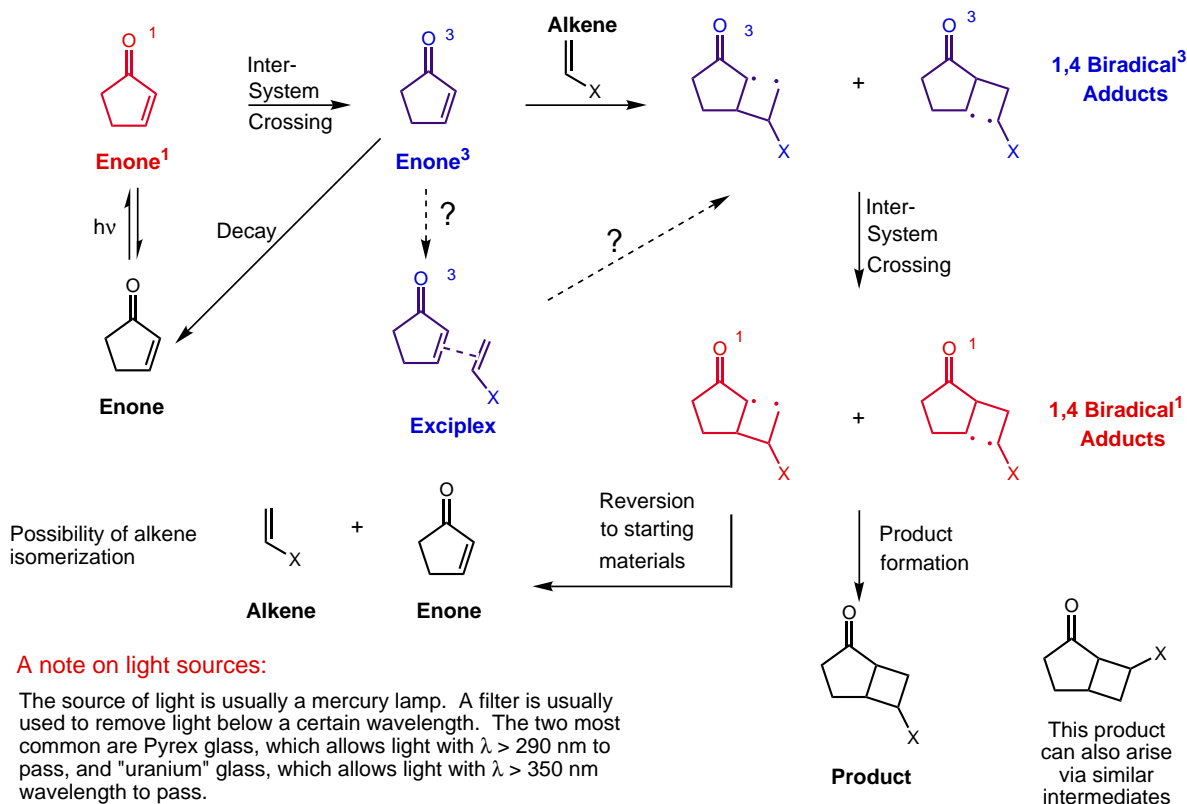
II) Selected synthetic examples:

- A) The Oppolzer syntheses
- B) The Pattenden syntheses
- C) The Pirrung syntheses
- D) One hit wonders
- E) The Crimmins syntheses
- F) The Winkler syntheses

The following will not be covered in this seminar:

- I) Non-target motivated studies of the intramolecular photocycloaddition
- II) Intramolecular Paterno-Büchi reactions
- III) Intramolecular ketene cycloadditions
- IV) Other intramolecular photocycloadditions (e.g. arene-olefin meta cycloadditions)

Mechanistic Considerations of the Enone-Olefin Photocycloaddition

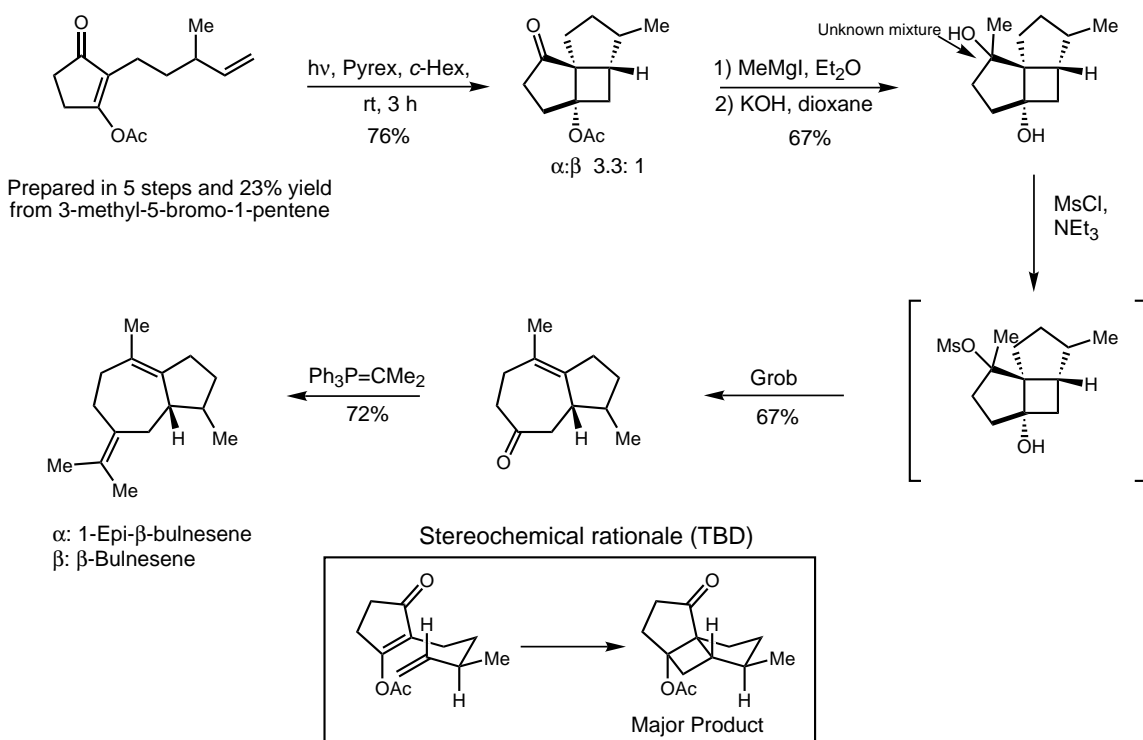


A note on light sources:

The source of light is usually a mercury lamp. A filter is usually used to remove light below a certain wavelength. The two most common are Pyrex glass, which allows light with $\lambda > 290$ nm to pass, and "uranium" glass, which allows light with $\lambda > 350$ nm wavelength to pass.

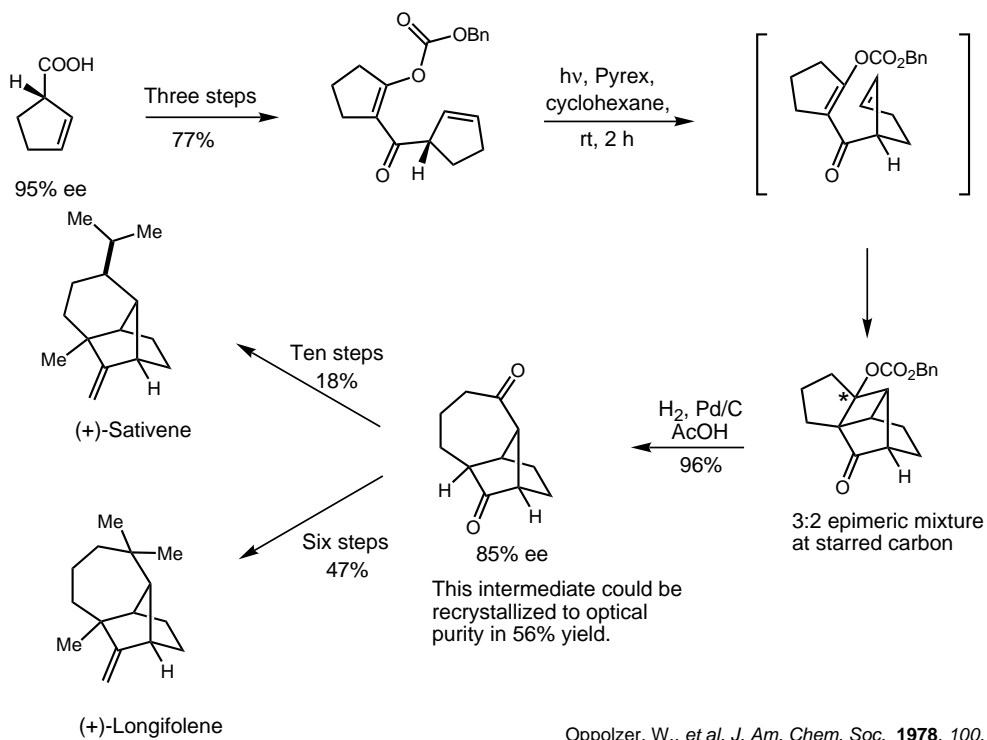
Schuster, D.I., *et al. Chem. Rev.* **1993**, 93, 3.

(±) β-Bulnesene and Epi-β-bulnesene



Oppolzer, W., *et al. Helv. Chim. Acta.*, **1980**, 63, 1198.

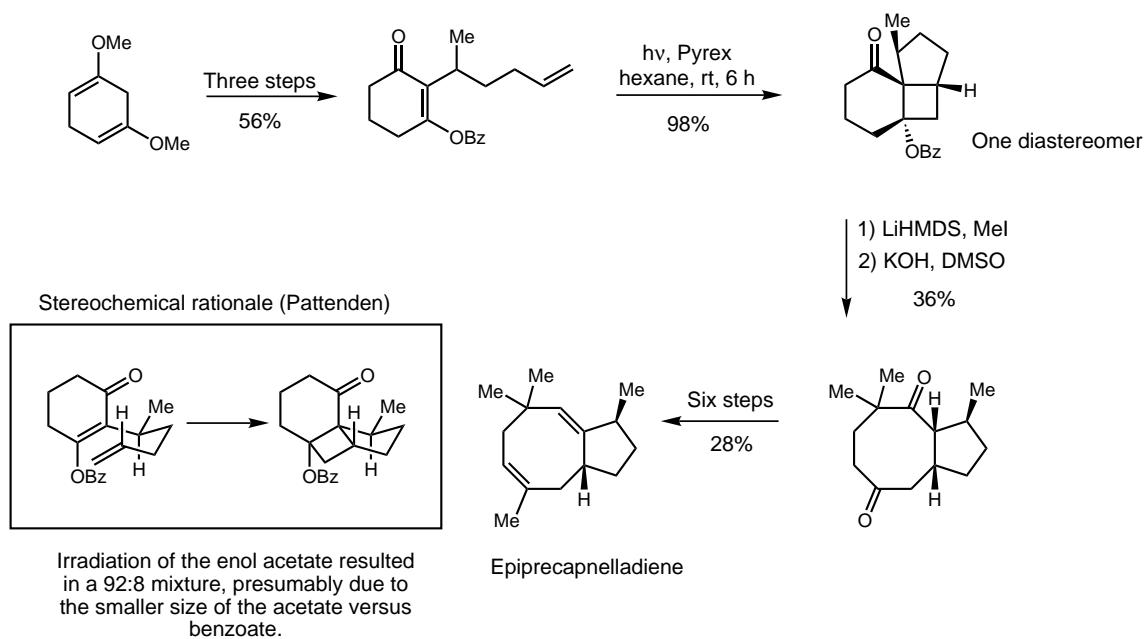
(+)-Longifolene and (+)-Sativene



Oppolzer, W., et al. *J. Am. Chem. Soc.* **1978**, *100*, 2583.

Oppolzer, W., et al. *Helv. Chim. Acta.* **1984**, *67*, 1154.

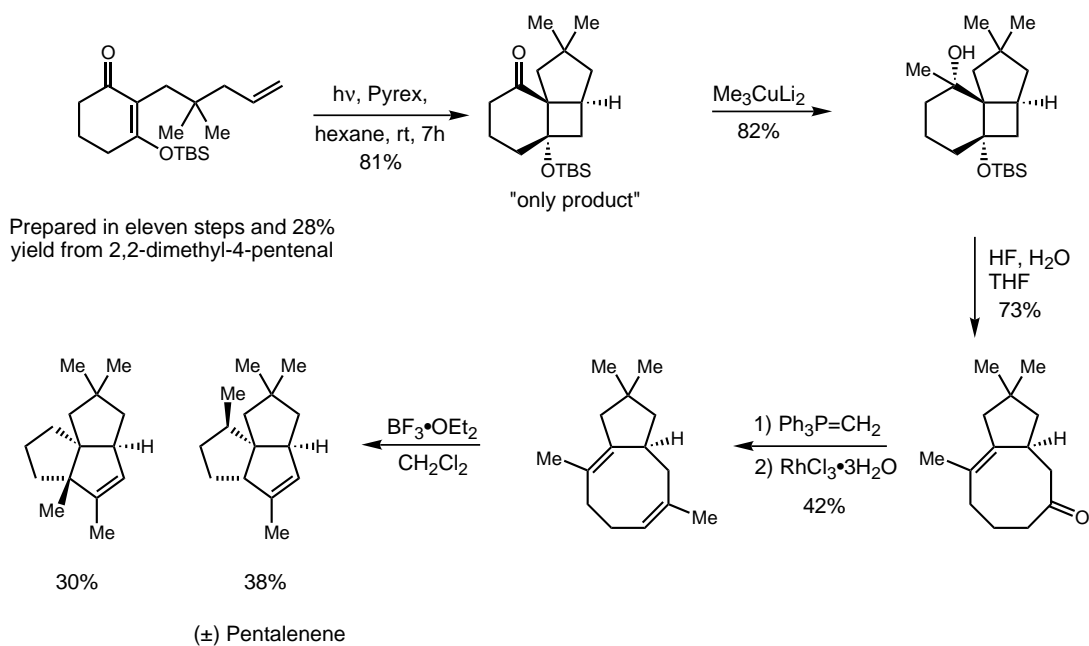
(±)-Epiprecapnelladiene



Pattenden, G., et al. *J. Chem. Soc., Chem. Comm.* **1980**, 1195.

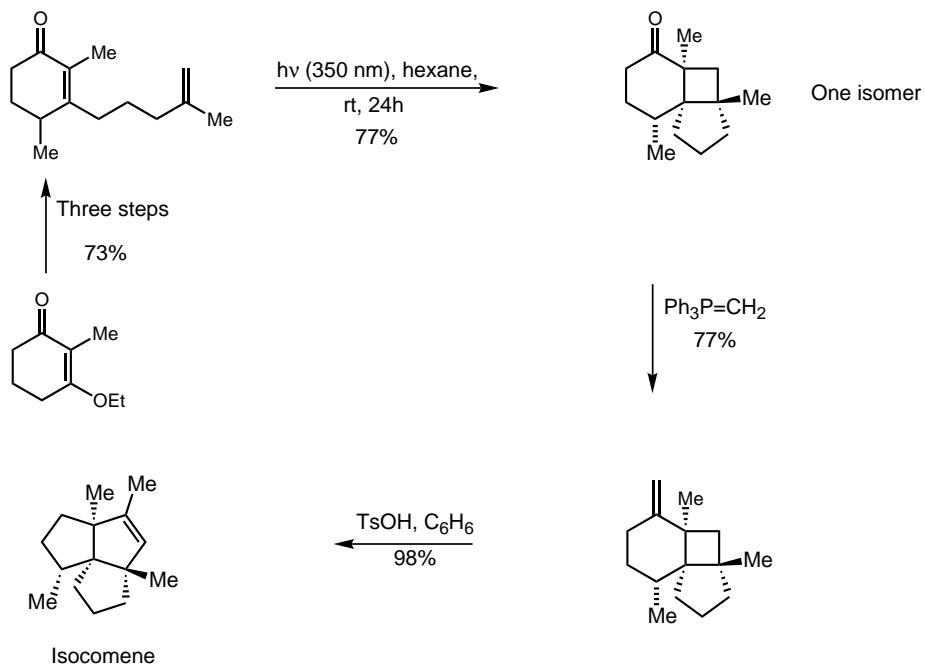
Pattenden, G., et al. *J. Chem. Soc., Perkin Trans. I* **1983**, 1913.

(±)-Pentalene



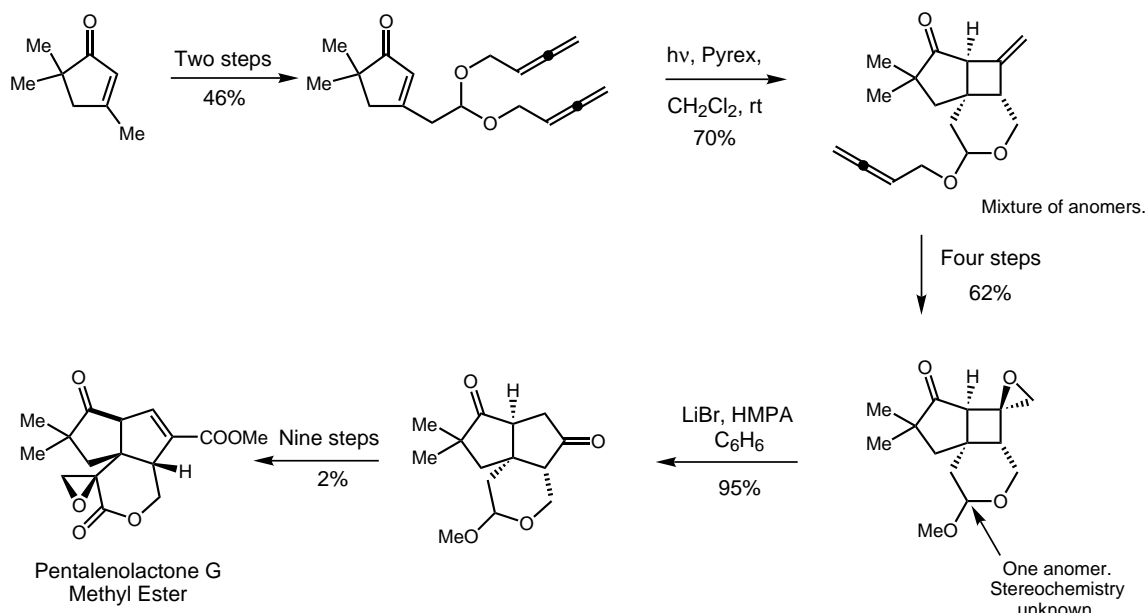
Pattenden, G., et al. *Tetrahedron Lett.* **1984**, 25, 3021.
Pattenden, G., et al. *Tetrahedron* **1987**, 43, 5637.

(±)-Isocomene



Pirrung, M.C. *J. Am. Chem. Soc.* **1979**, 101, 7130.
Pirrung, M.C. *J. Am. Chem. Soc.* **1981**, 103, 82.

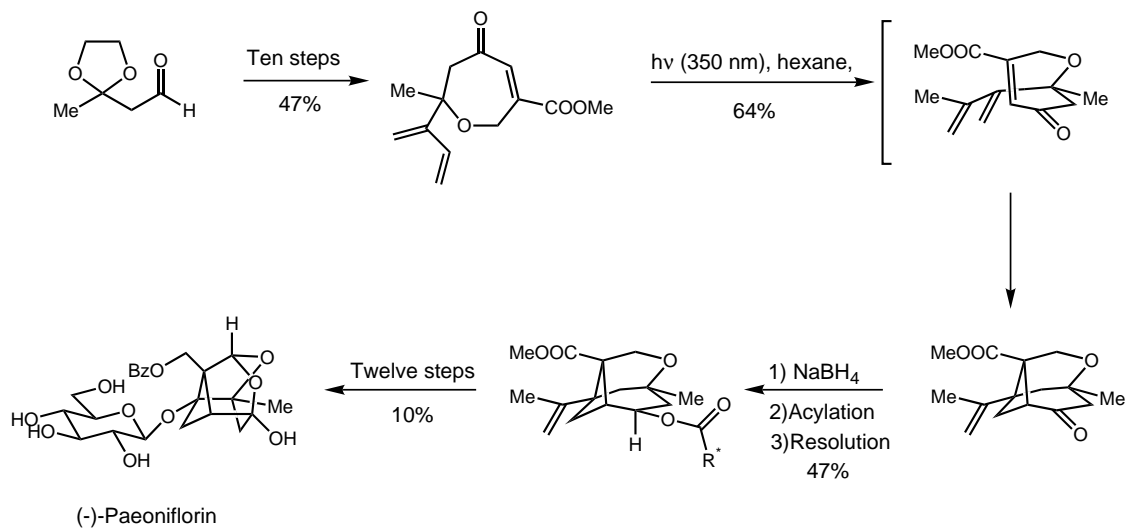
(±)-Pentalenolactone G Methyl Ester



Pirrung, M.C., et al. *Tetrahedron Lett.* **1986**, 27, 2703.

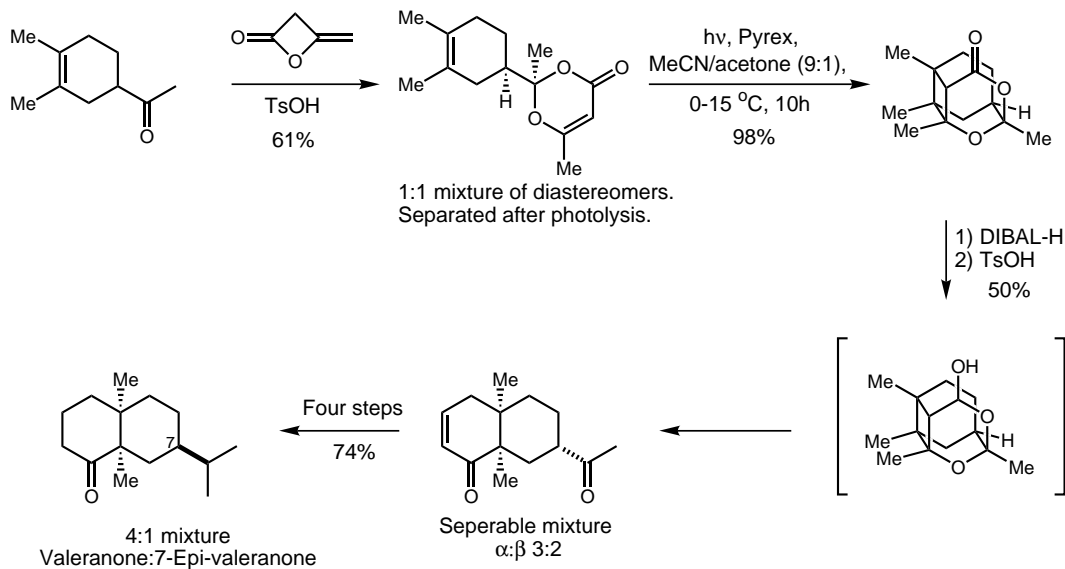
Pirrung, M.C., et al. *J. Org. Chem.* **1988**, 53, 227.

(-)-Paeoniflorin



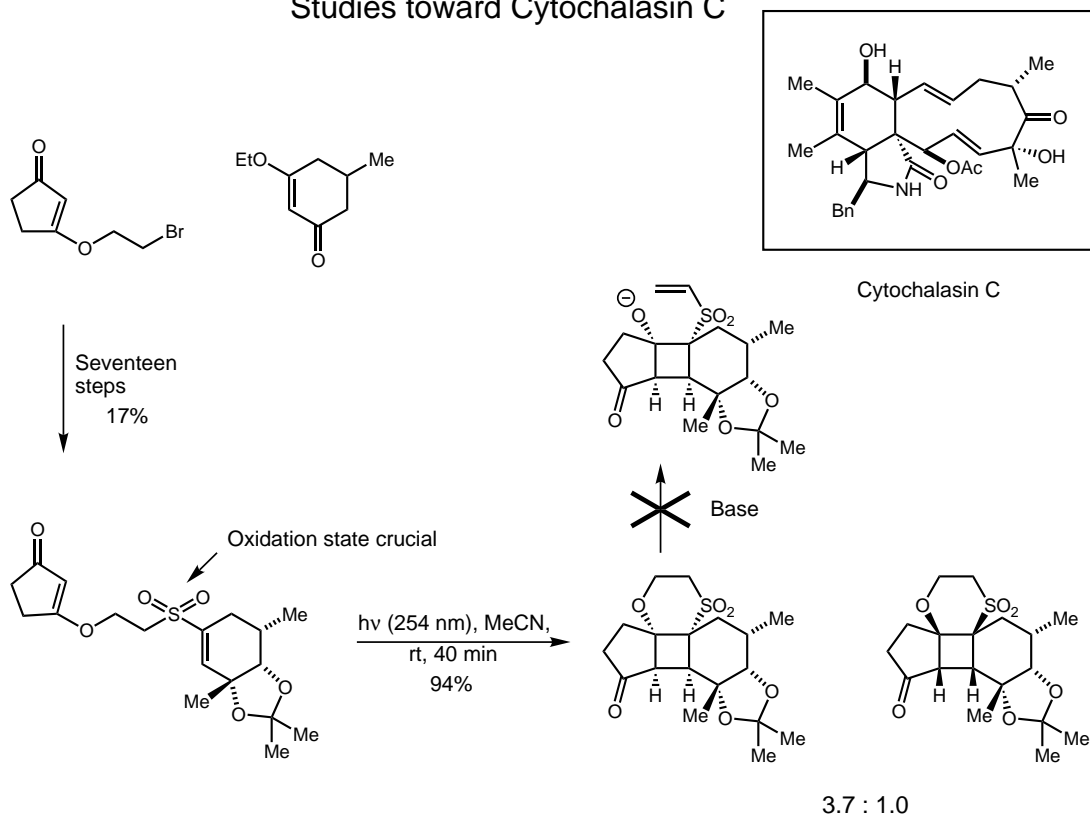
Hatakeyama, S., Takano, S., et al. *J. Am. Chem. Soc.* **1994**, 116, 4081.

(±)-Valeranone



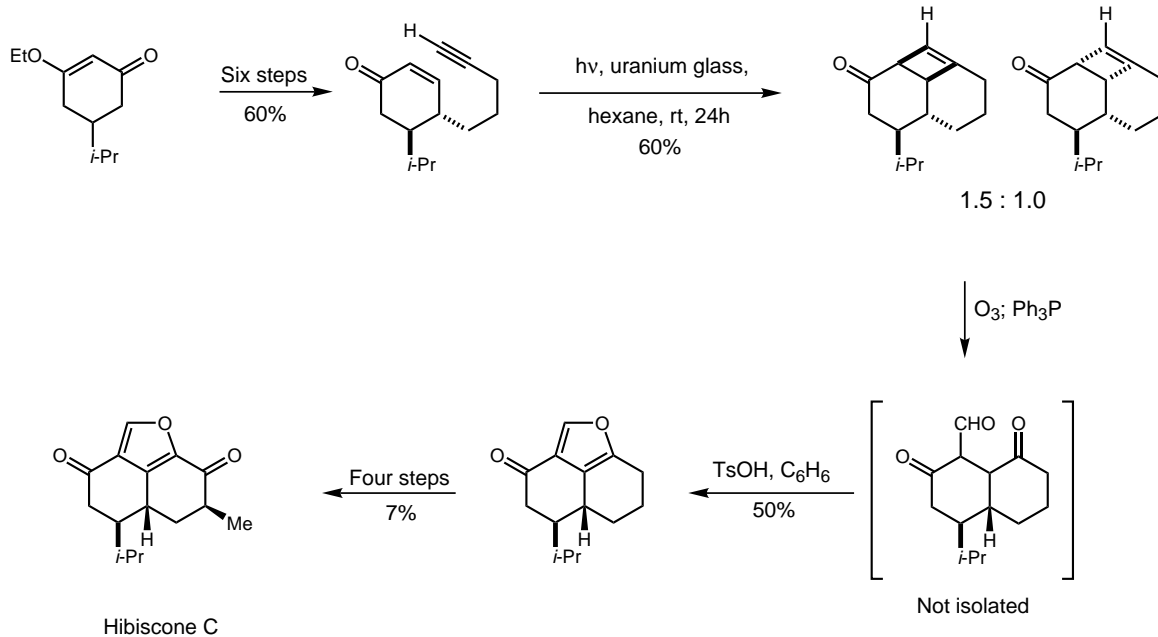
Takeshita, H. *et al. Bull. Chem. Soc. Jpn.* **1993**, 66, 2699.

Studies toward Cytochalasin C



Fuchs, P.L., *et al. J. Org Chem.* **1982**, 47, 3121.

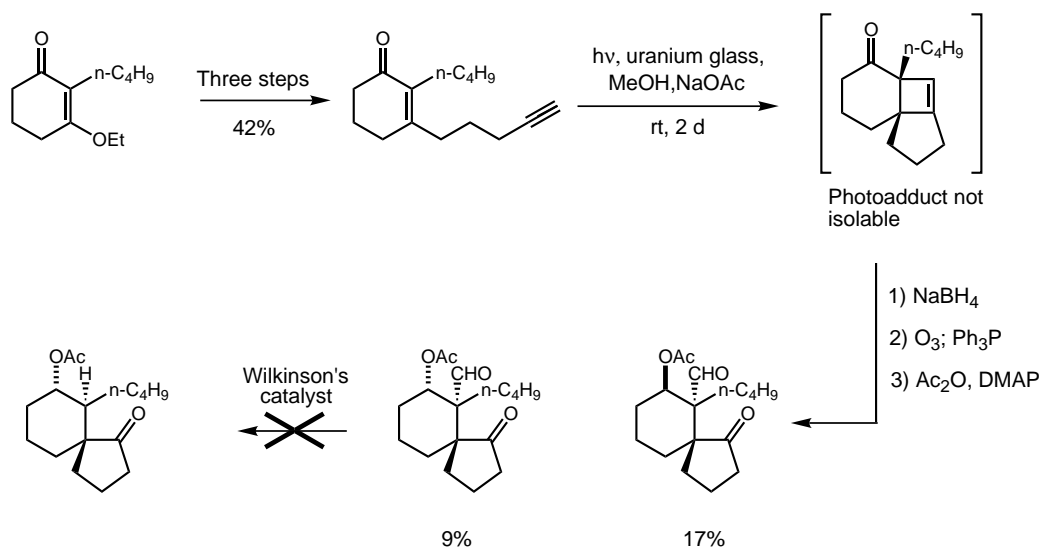
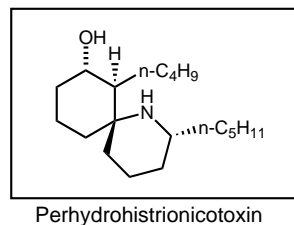
(±)-Hibiscone C



Smith, A.B., III, *et al. J. Am. Chem. Soc.* **1982**, 104, 5568.

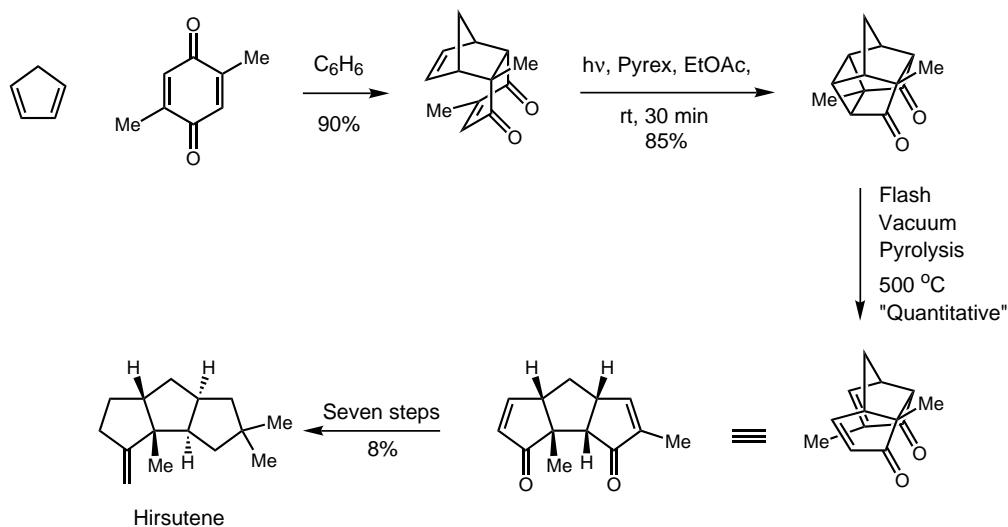
Smith, A.B., III, *et al. J. Am. Chem. Soc.* **1984**, 106, 2115.

Studies toward Perhydrohistrionicotoxin



Smith, A.B., III, *et al. J. Org. Chem.* **1984**, 49, 832.

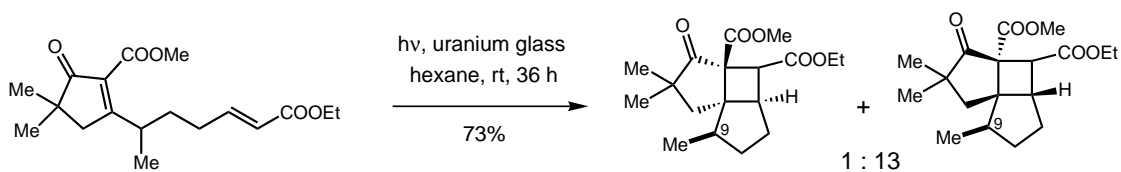
(±)-Hirsutene



Formal syntheses of capnellene and coriolin were reported using a similar photoaddition/fragmentation reaction.

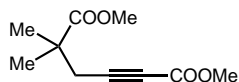
Mehta, G., *et al. J. Am. Chem. Soc.* **1986**, 108, 3443.
 Mehta, G., *et al. J. Chem. Soc., Chem. Comm.* **1981**, 756.

(±)-Pentalenene, (±)-Pentalenic Acid and (±)-Deoxypentalenic Acid from a Common Precursor.



The stereochemistry at the starred carbon was not determined, but the products were isolated as a mixture.

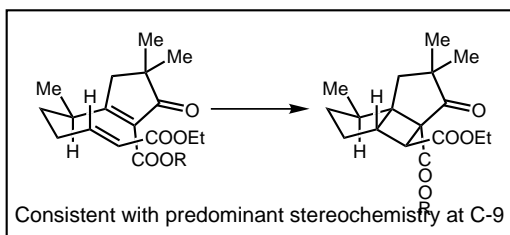
Three steps
44%



Stereochemical rationale (Crimmins)

| R | C-9 dr |
|--------------|--------|
| Me | 13:1 |
| Et | 17:1 |
| <i>i</i> -Pr | >20:1 |

Varying the size of the indicated alkyl group influenced the stereoselectivity



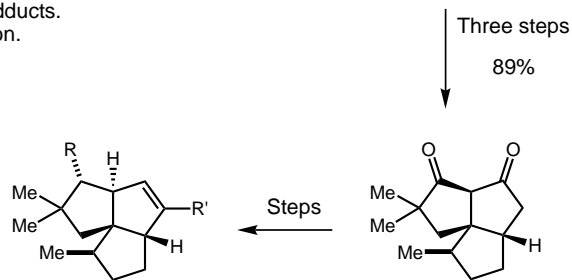
Crimmins, M.T., *et al. J. Org. Chem.* **1984**, 49, 2076.
 Crimmins, M.T., *et al. J. Am. Chem. Soc.* **1986**, 108, 800.

(±)-Pentalenene, (±)-Pentalenic Acid and
(±)-Deoxypentalenic Acid from a Common Precursor.



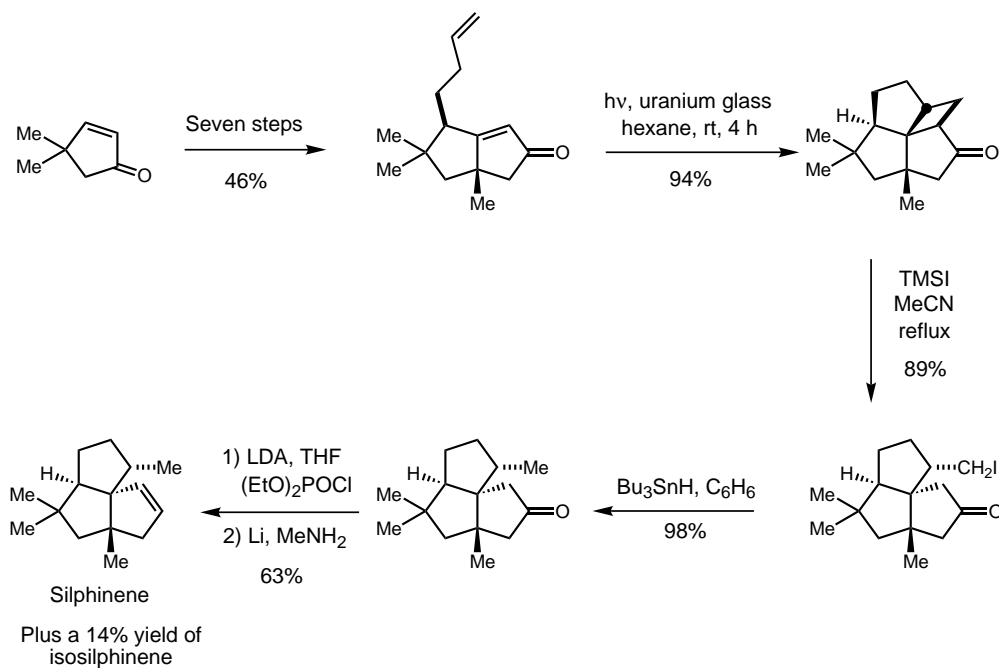
Reduction performed on mixture of photoadducts.
Diastereomers separated after reduction.

R=H, R'=Me, Pentalenene,
nine steps, 27%
R=OH, R'=COOH, Pentalenic acid,
eight steps, 42%
R=H, R'=COOH, Deoxypentalenic acid,
eleven steps, 22%



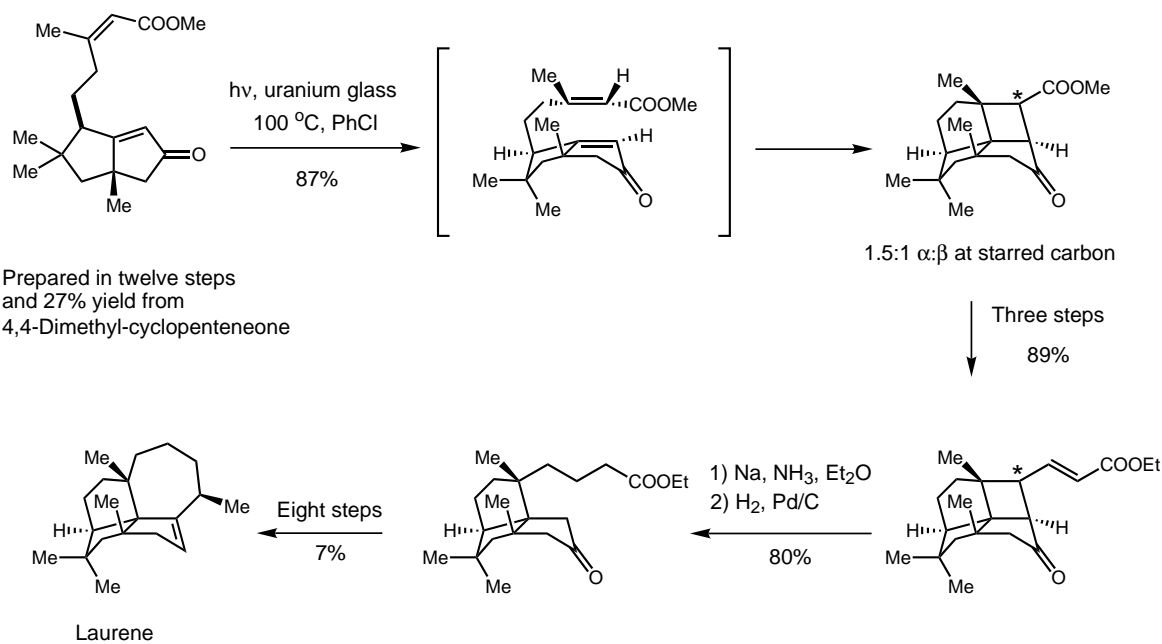
Crimmins, M.T., *et al. J. Org. Chem.* **1984**, *49*, 2076.
Crimmins, M.T., *et al. J. Am. Chem. Soc.* **1986**, *108*, 800.

(±)-Silphinene



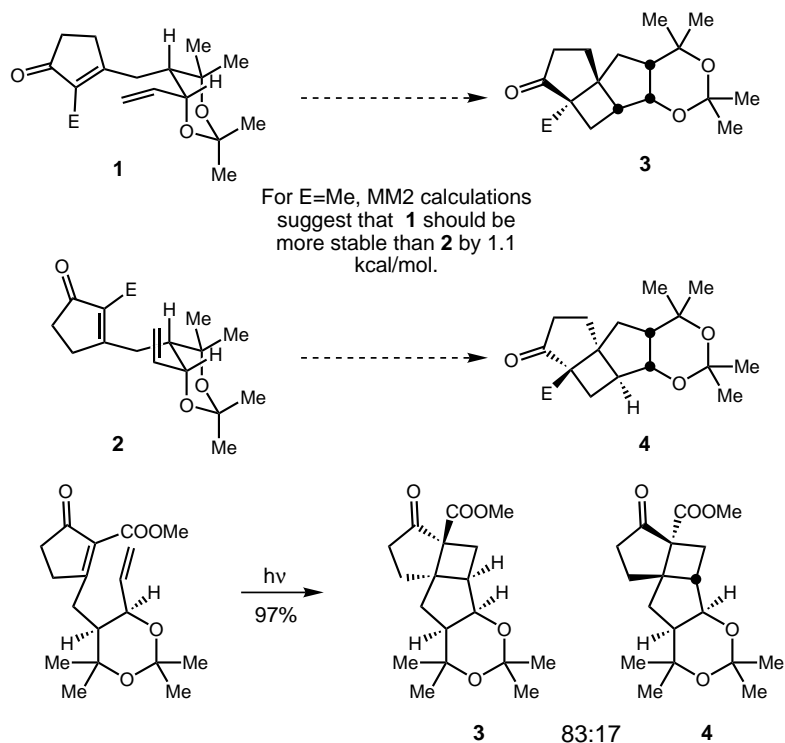
Crimmins, M.T., *et al. J. Am. Chem. Soc.* **1986**, *108*, 3435.
Crimmins, M.T., *et al. Tetrahedron Lett.* **1985**, *26*, 997.

(±)-Laurenene



Crimmins, M.T., *et al. J. Am. Chem. Soc.* **1987**, *109*, 6199.
Crimmins, M.T., *et al. Tetrahedron Lett.* **1985**, *26*, 997.

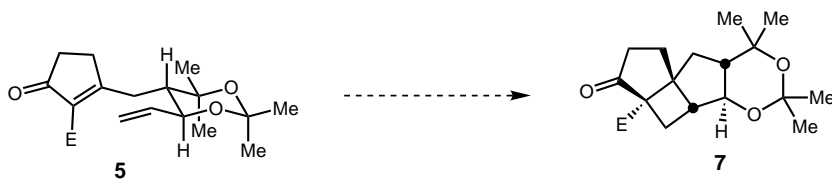
(±)-Lubiminol



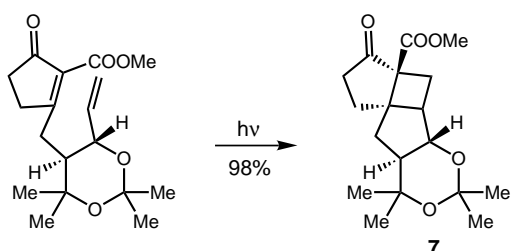
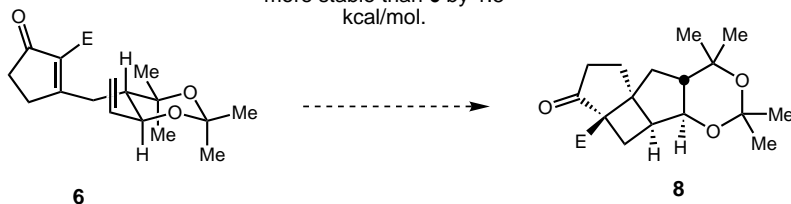
Crimmins, M.T., *et al. Tetrahedron Lett.* **1996**, *37*, 8703.

Crimmins, M.T., *et al. J. Am. Chem. Soc.* **1998**, *120*, 1747.

(±)-Lubiminol



For E=Me, MM2 calculations suggest that **5** should be more stable than **6** by 1.5 kcal/mol.

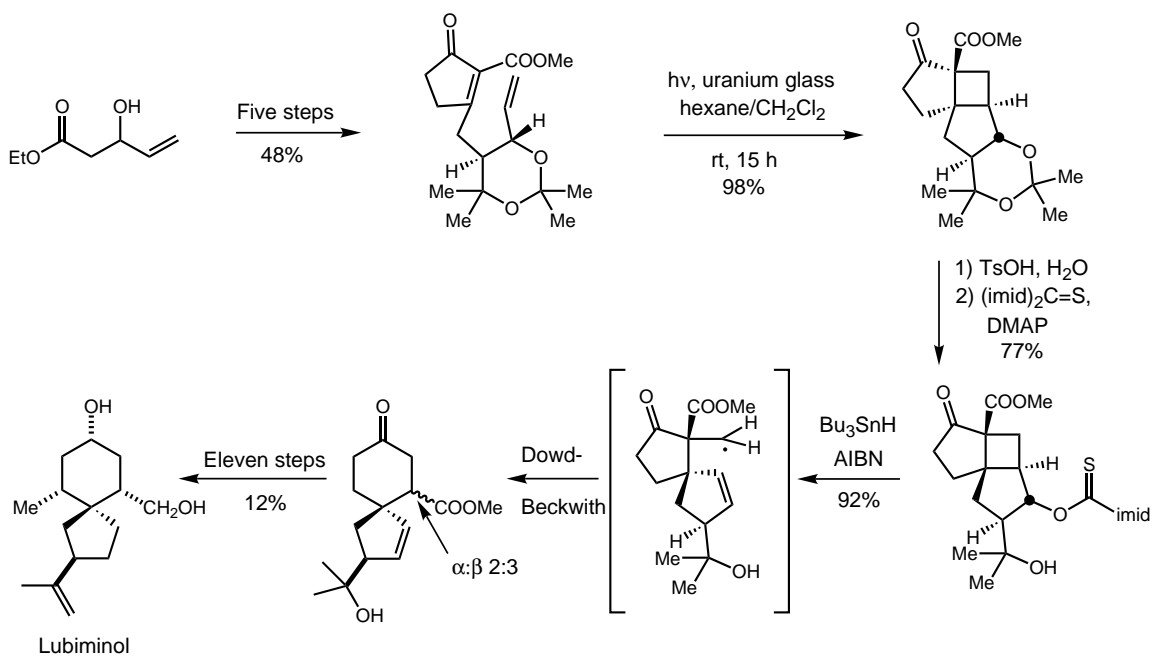


Only isolated product

Crimmins, M.T., *et al. Tetrahedron Lett.* **1996**, 37, 8703.

Crimmins, M.T., *et al. J. Am. Chem. Soc.* **1998**, 120, 1747.

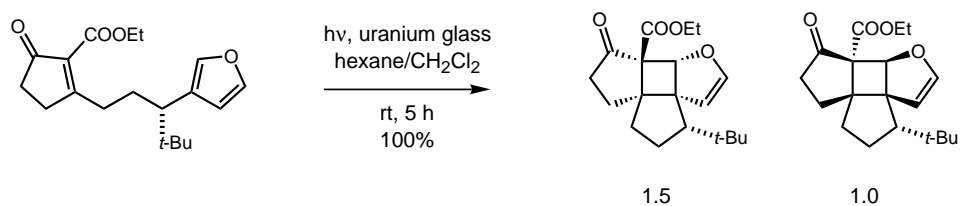
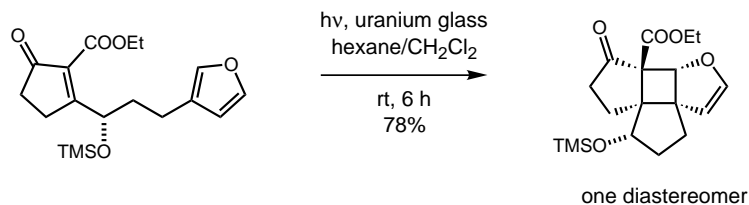
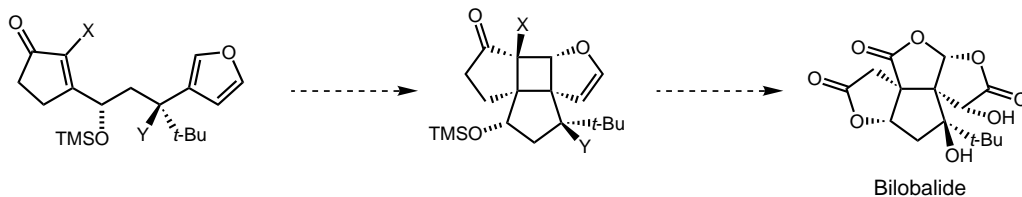
(±)-Lubiminol



Crimmins, M.T., *et al. Tetrahedron Lett.* **1996**, 37, 8703.

Crimmins, M.T., *et al. J. Am. Chem. Soc.* **1998**, 120, 1747.

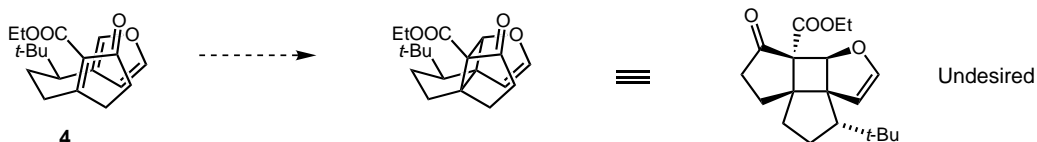
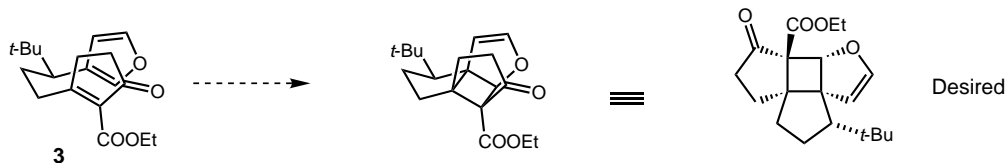
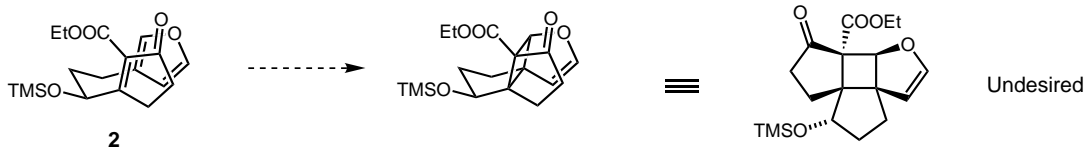
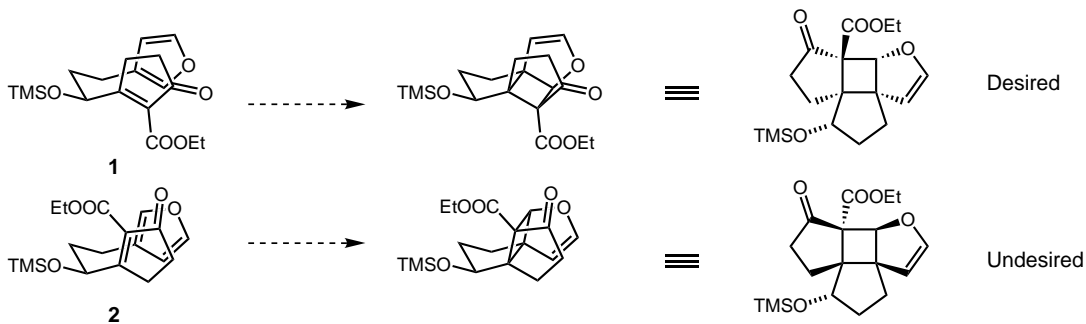
(±)-Bilobalide



Crimmins, M.T., *et al. J. Am. Chem. Soc.* **1992**, *114*, 5445.

Crimmins, M.T., *et al. J. Am. Chem. Soc.* **1993**, *115*, 3146.

(±)-Bilobalide

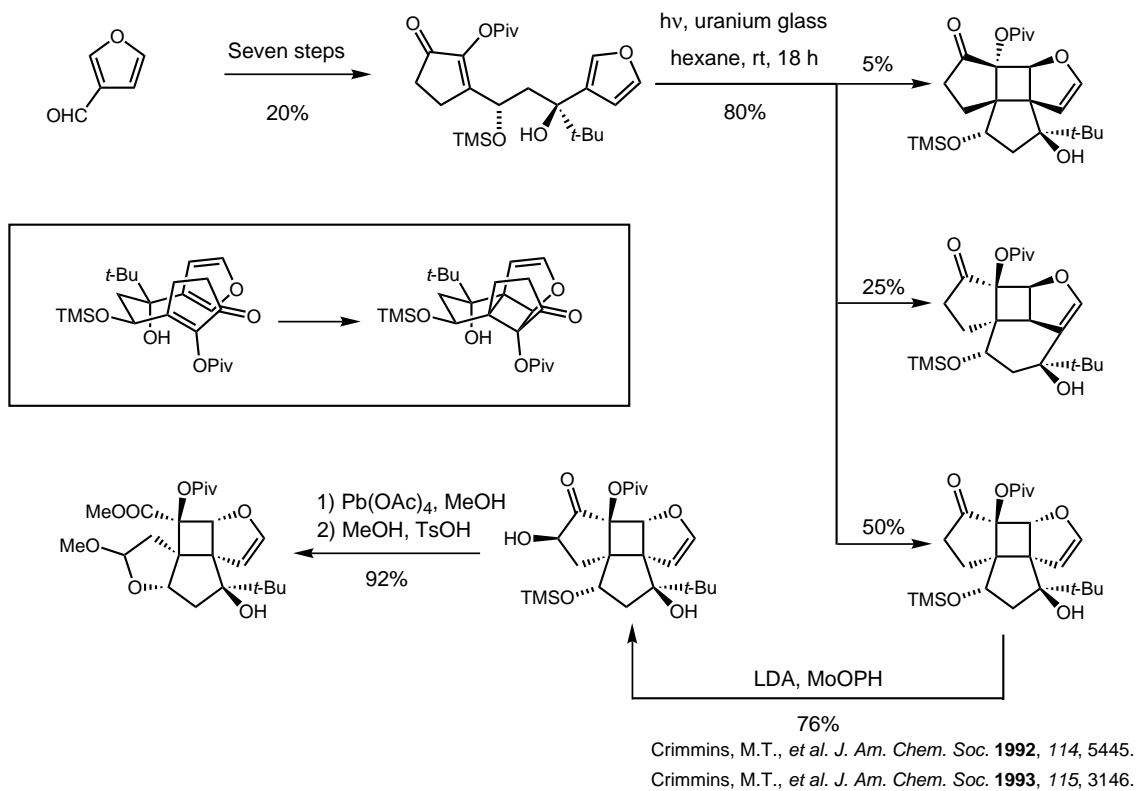


MM2 calculations suggested that **1** was more favored than **2** by approximately 1.6 kcal/mol, while there was almost no difference in energy between **3** and **4** (0.2 kcal/mol).

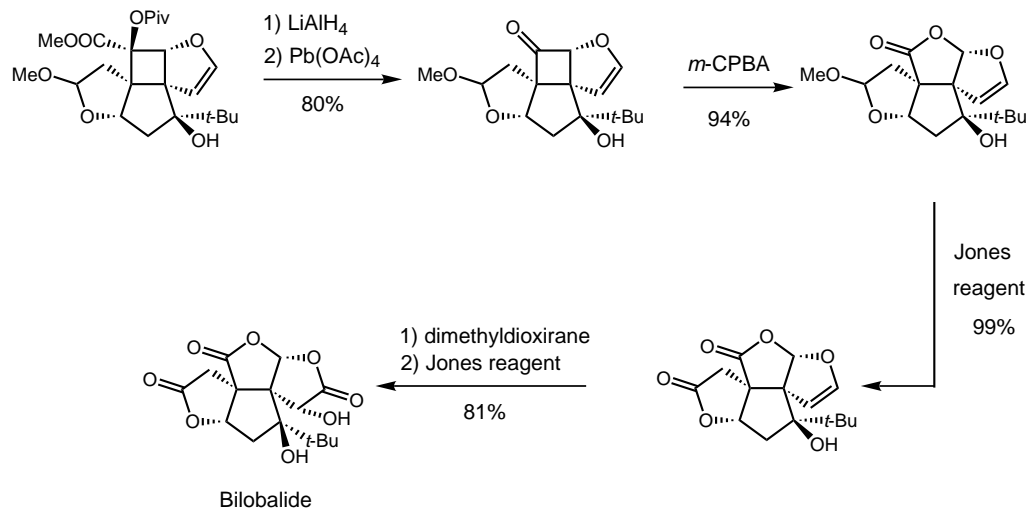
Crimmins, M.T., *et al. J. Am. Chem. Soc.* **1992**, *114*, 5445.

Crimmins, M.T., *et al. J. Am. Chem. Soc.* **1993**, *115*, 3146.

(±)-Bilobalide

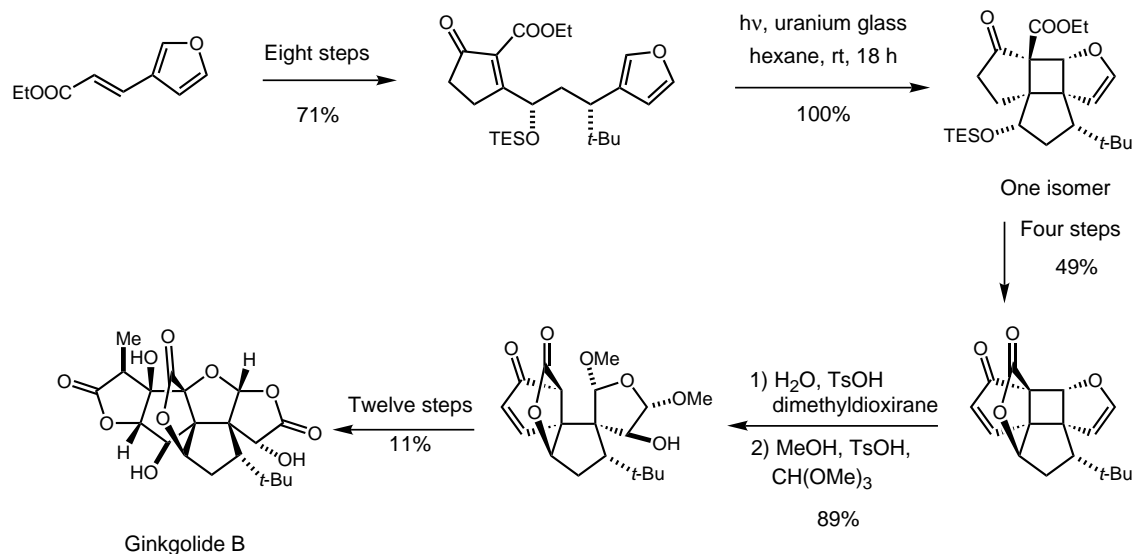


(±)-Bilobalide



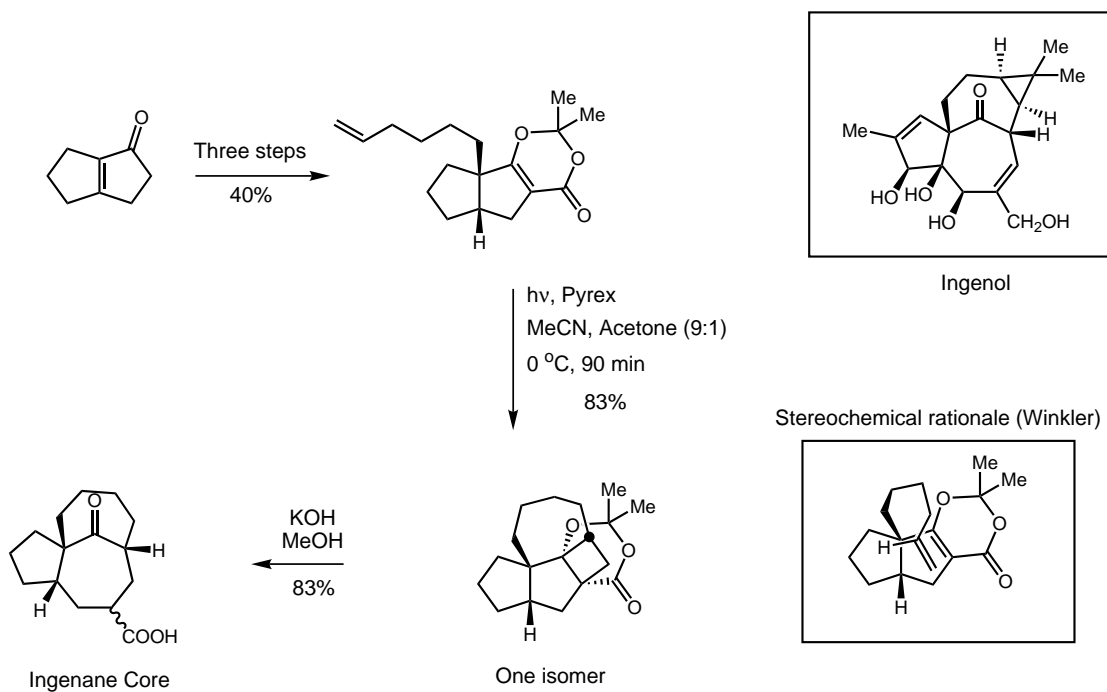
Crimmins, M.T., et al. *J. Am. Chem. Soc.* **1992**, 114, 5445.
Crimmins, M.T., et al. *J. Am. Chem. Soc.* **1993**, 115, 3146.

(±)-Ginkgolide B



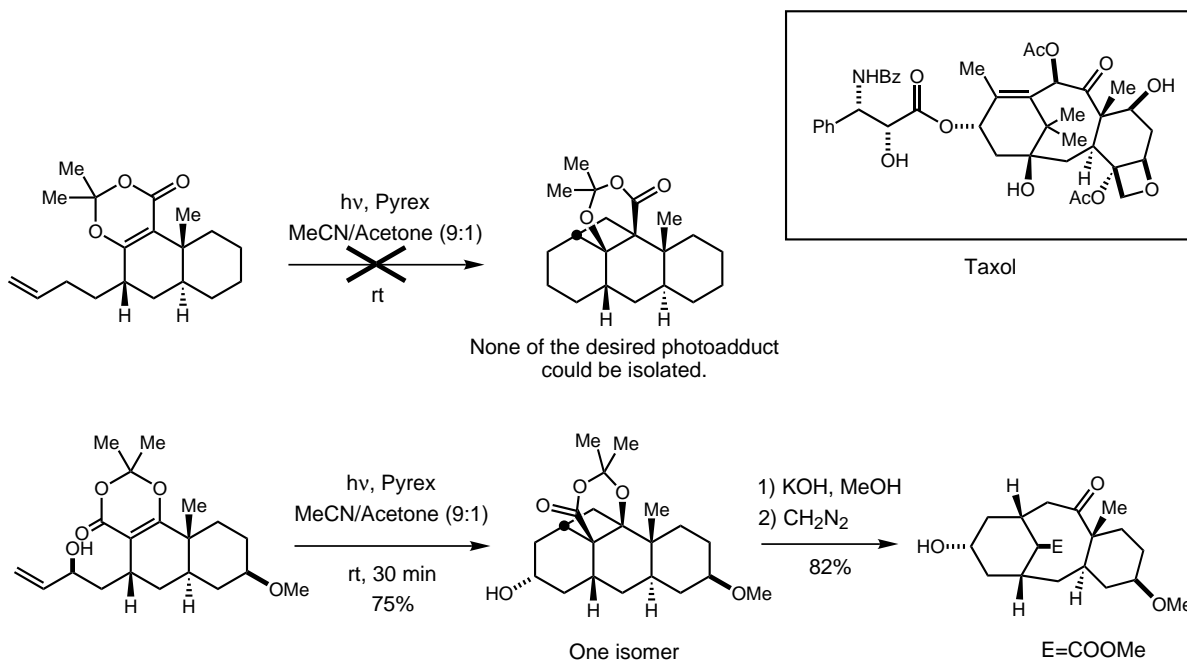
Crimmins, M.T. *et al.* *J. Am. Chem. Soc.* **1999**, *121*, 10249.
 Crimmins, M.T. *et al.* *Tetrahedron Lett.* **1989**, *30*, 5997.

Synthesis of the Core of the Ingenane Diterpenes



Winkler, J.D., *et al.* *J. Am. Chem. Soc.* **1987**, *109*, 2850.

Synthetic Studies Toward the Taxane Core

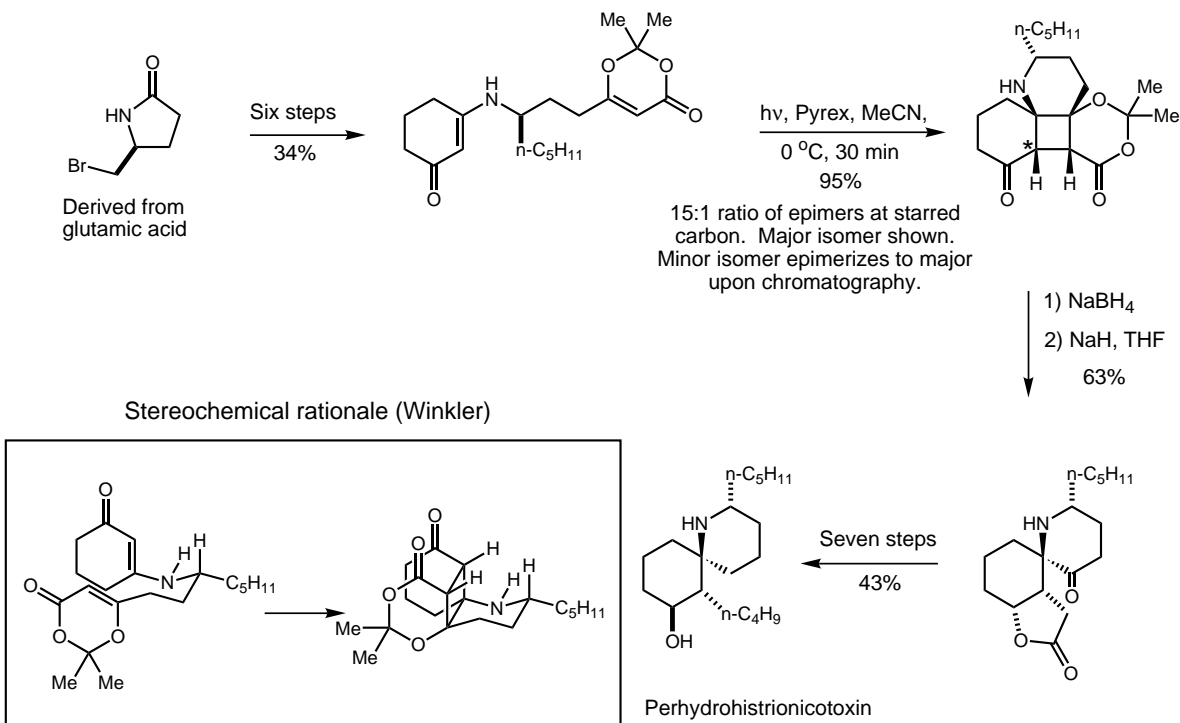


Winkler, J.D., *et al. Tetrahedron Lett.* **1986**, 27, 5959.

Winkler, J.D., *et al. J. Org. Chem.* **1989**, 54, 4491.

Winkler, J.D., *et al. Tetrahedron* **1992**, 48, 7049.

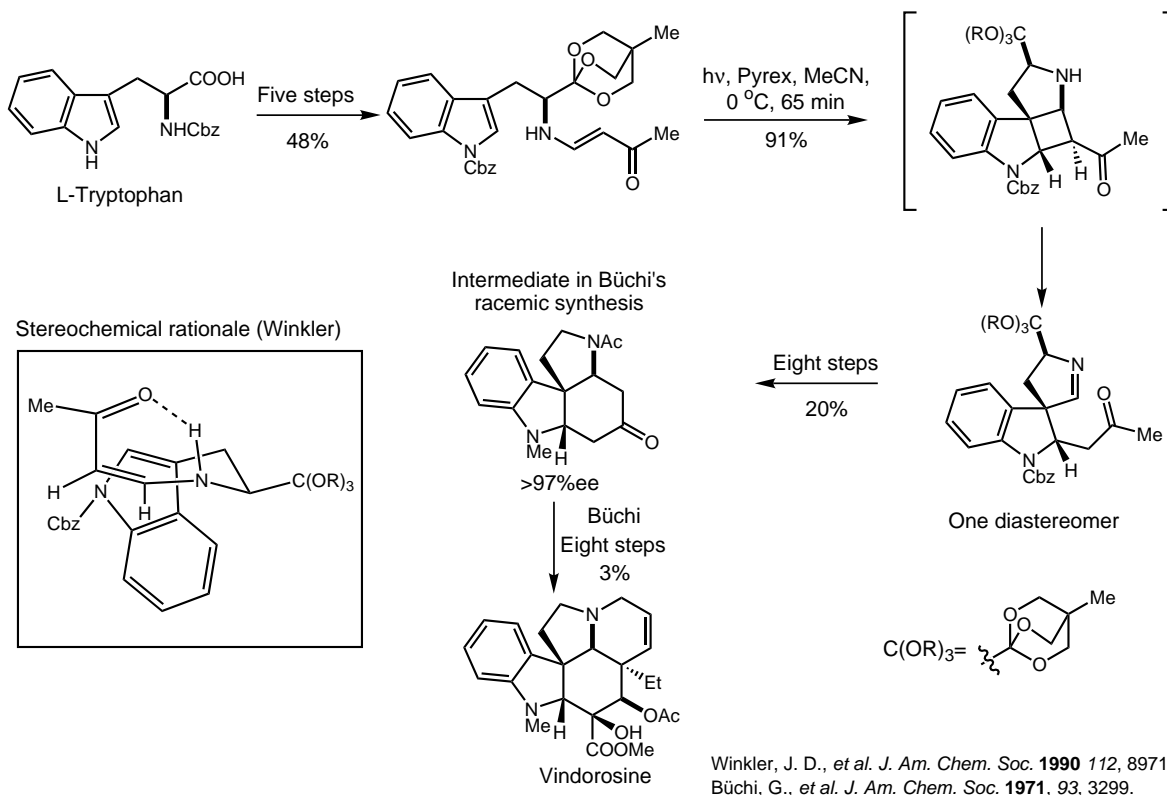
(-)-Perhydrohistrionicotoxin



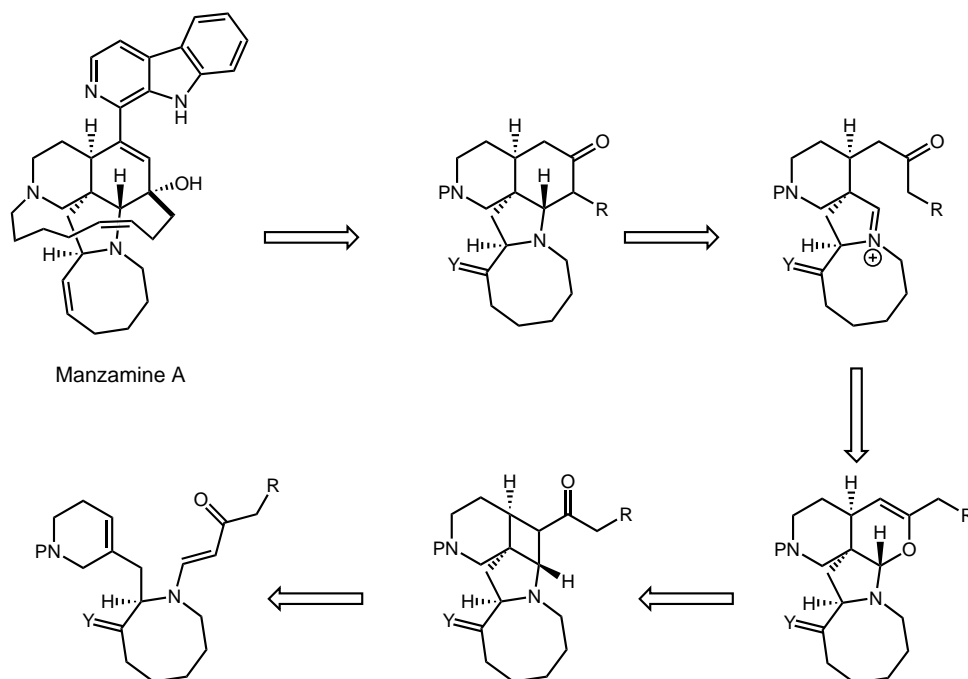
Winkler, J.D., *et al. Tetrahedron Lett.* **1986**, 27, 5177.

Winkler, J.D., *et al. J. Am. Chem. Soc.* **1989**, 111, 4852.

Formal Synthesis of Vindorosine



(±)-Manzamine A

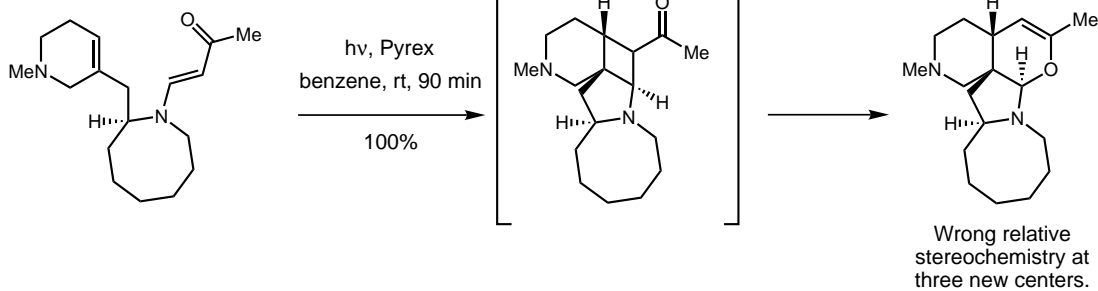


For an analysis of the total synthesis of Manzamine A, see the Evans Group Seminar "Approaches to the Total Synthesis of the Manzamine Alkaloids," Hemaka Rajapakse, Jan. 21, 2000.

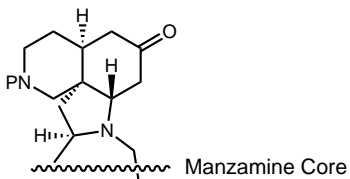
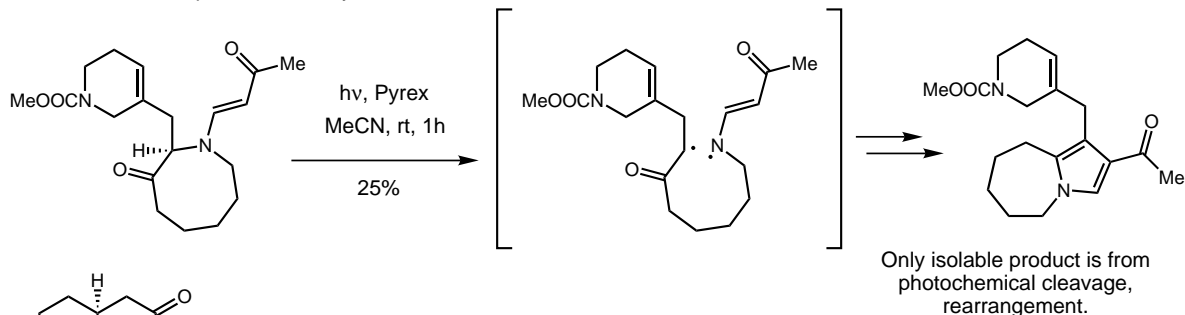
Winkler, J.D., *et al. Tetrahedron Lett.* **1993**, 34, 6509.
 Winkler, J.D., *et al. Isr. J. Chem.* **1997**, 37, 47.
 Winkler, J.D., *et al. Tetrahedron* **1998**, 54, 7045.
 Winkler, J.D., *et al. J. Am. Chem. Soc.* **1998**, 120, 6425.

(±)-Manzamine A

Model system: Saturated ring



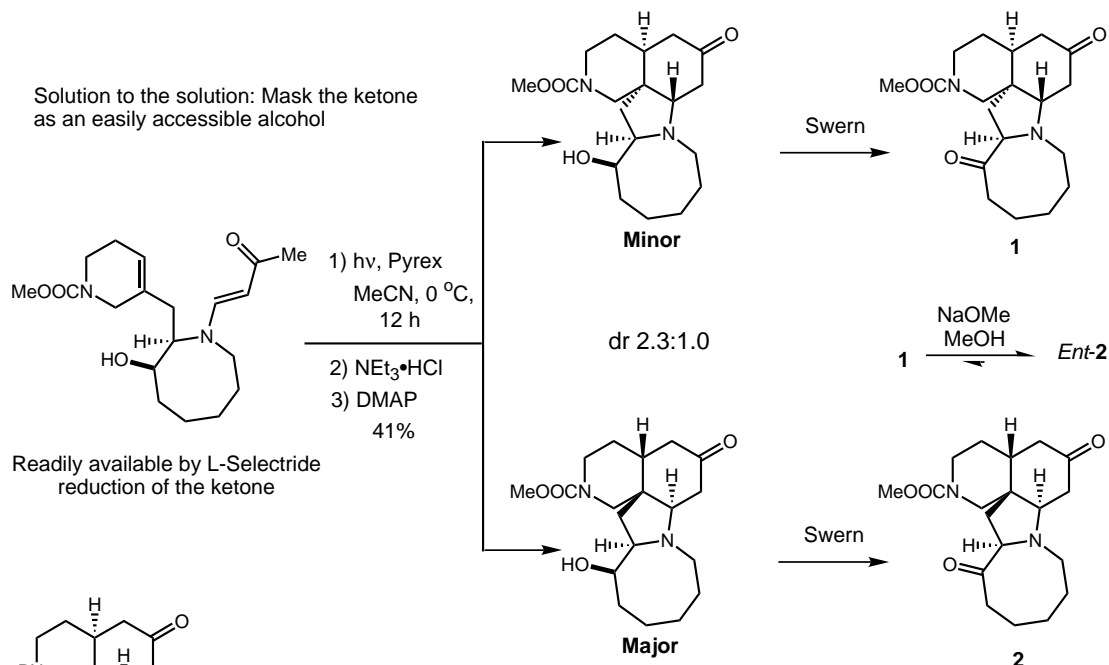
Possible solution: Epimerize after cyclization



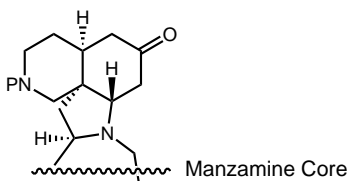
Winkler, J.D., et al. *Tetrahedron Lett.* **1993**, 34, 6509.
Winkler, J.D., et al. *Isr. J. Chem.* **1997**, 37, 47.
Winkler, J.D., et al. *Tetrahedron* **1998**, 54, 7045.
Winkler, J.D., et al. *J. Am. Chem. Soc.* **1998**, 120, 6425.

(±)-Manzamine A

Solution to the solution: Mask the ketone as an easily accessible alcohol



Readily available by L-Selectride reduction of the ketone

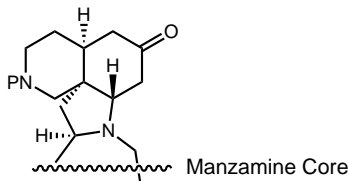
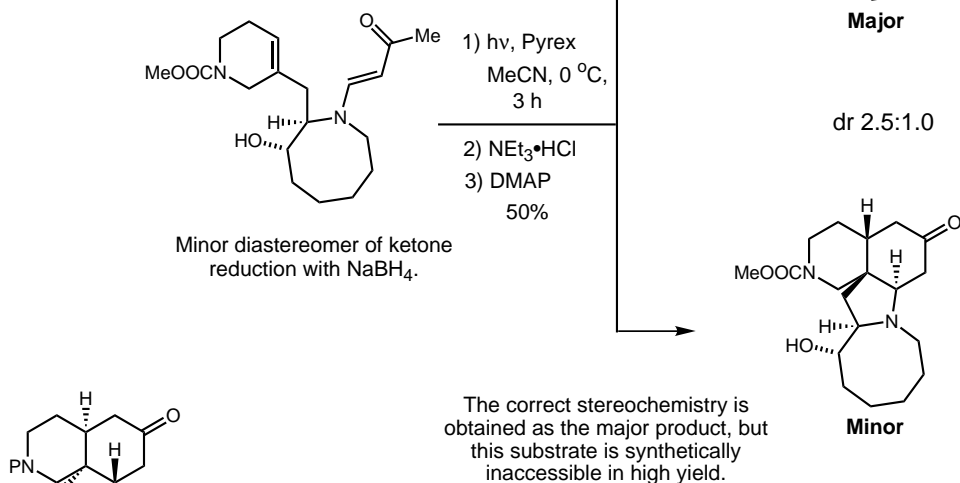


Again, the predominant product has the wrong stereochemistry and cannot be isomerized to the correct stereochemistry.

Winkler, J.D., et al. *Tetrahedron Lett.* **1993**, 34, 6509.
Winkler, J.D., et al. *Isr. J. Chem.* **1997**, 37, 47.
Winkler, J.D., et al. *Tetrahedron* **1998**, 54, 7045.
Winkler, J.D., et al. *J. Am. Chem. Soc.* **1998**, 120, 6425.

(±)-Manzamine A

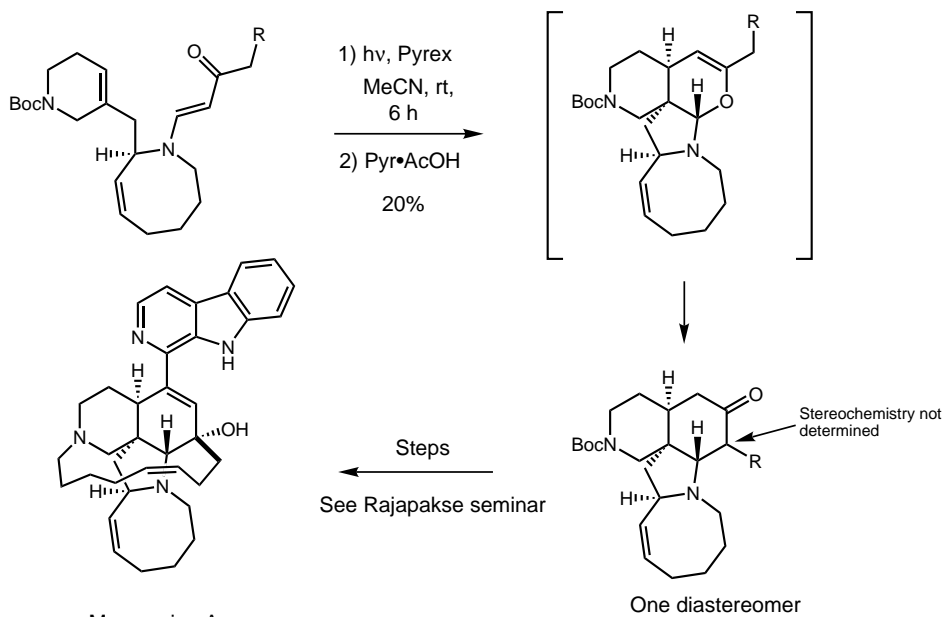
Partial solution: The opposite diastereomer provides the desired stereochemistry



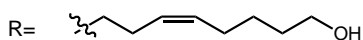
Winkler, J.D., *et al. Tetrahedron Lett.* **1993**, 34, 6509.
Winkler, J.D., *et al. Isr. J. Chem.* **1997**, 37, 47.
Winkler, J.D., *et al. Tetrahedron* **1998**, 54, 7045.
Winkler, J.D., *et al. J. Am. Chem. Soc.* **1998**, 120, 6425.

(±)-Manzamine A

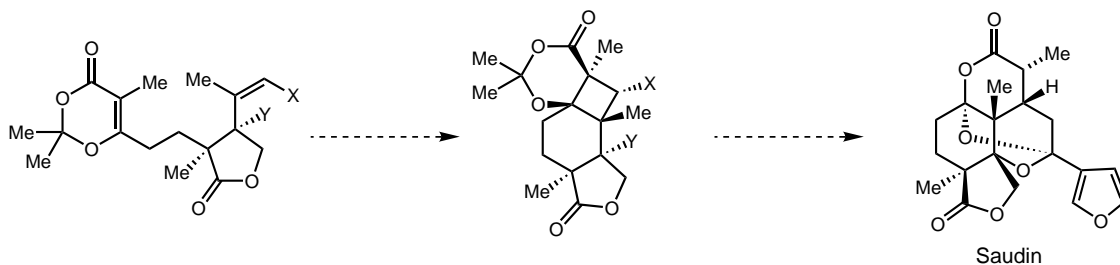
Ultimate solution: The olefinic linkage in the natural product provides for the correct stereoinduction



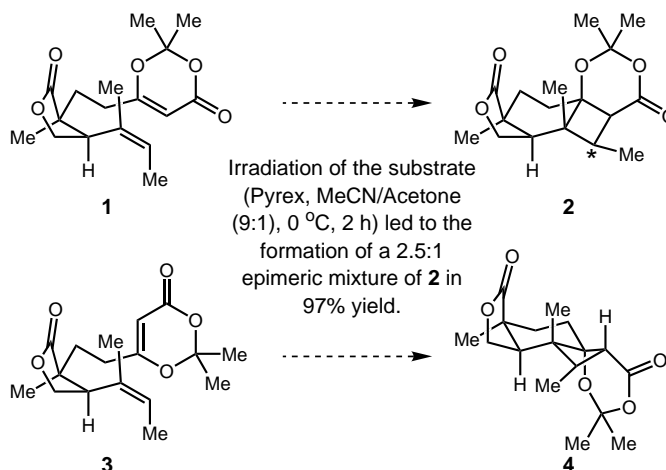
Winkler, J.D., *et al. Tetrahedron Lett.* **1993**, 34, 6509.
Winkler, J.D., *et al. Isr. J. Chem.* **1997**, 37, 47.
Winkler, J.D., *et al. Tetrahedron* **1998**, 54, 7045.
Winkler, J.D., *et al. J. Am. Chem. Soc.* **1998**, 120, 6425.



(±)-Saudin



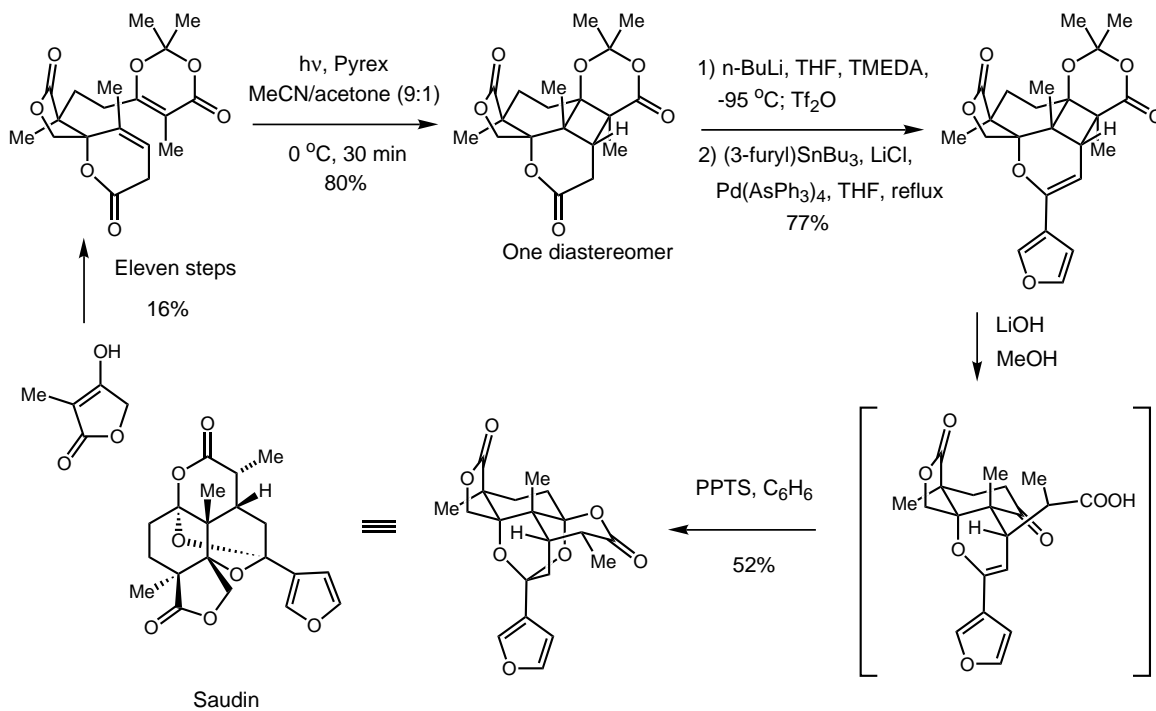
MM2 calculations suggest that **1** should be approximately 1.6 kcal/mol more stable than **3**.



Irradiation of the substrate (Pyrex, MeCN/Acetone (9:1), 0 °C, 2 h) led to the formation of a 2.5:1 epimeric mixture of **2** in 97% yield.

Winkler, J.D., et al. *J. Am. Chem. Soc.* **1999**, 121, 7425.
Winkler, J.D., et al. *Tetrahedron Lett.* **1998**, 39, 2253.

(±)-Saudin



Winkler, J.D., et al. *J. Am. Chem. Soc.* **1999**, 121, 7425.
Winkler, J.D., et al. *Tetrahedron Lett.* **1998**, 39, 2253.

Conclusions

- A lack of complete mechanistic understanding has not prevented the successful application of the enone-olefin photocycloaddition to a great many synthetic challenges.
 - Intramolecular enone-olefin photocycloaddition is useful for the stereoselective construction of carbocycles, especially five membered rings.
 - Multiple quaternary and congested centers can be constructed in a single operation.
 - Great creativity is possible in the fragmentation of the derived cyclobutane to access useful synthetic intermediates.
 - Stereinduction from existing stereocenters is usually explicable in a rational and satisfying manner.
 - Molecular modeling simulations have proven useful in predicting the stereoselectivity of several photocycloadditions.
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