

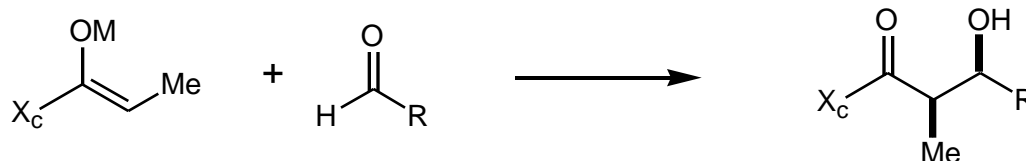
Denmark's Base Catalyzed Aldol/Allylation

Evans' Group Seminar
November 21th, 2003
Jimmy Wu

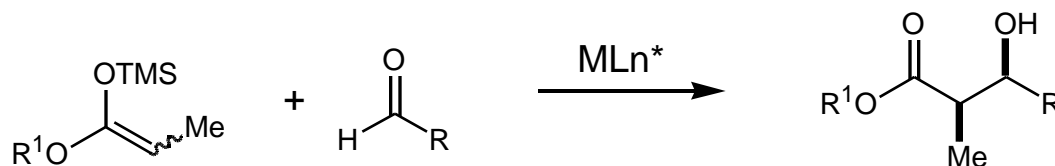
Lead References:

Denmark, S. E. *Acc. Chem. Res.*, **2000**, 33, 432
Denmark, S. E. *Chem. Comm.* **2003**, 167
Denmark, S. E. *Chem. Rev.* **2003**, 103, 2763
Denmark, S. E. *JOC*, **1998**, 63, 9517

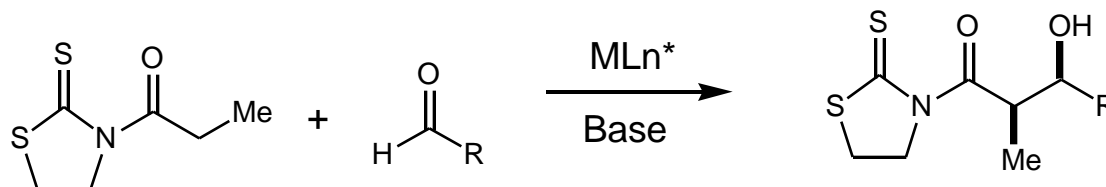
Background



- Requires stoichiometric amounts of covalently bound auxiliaries

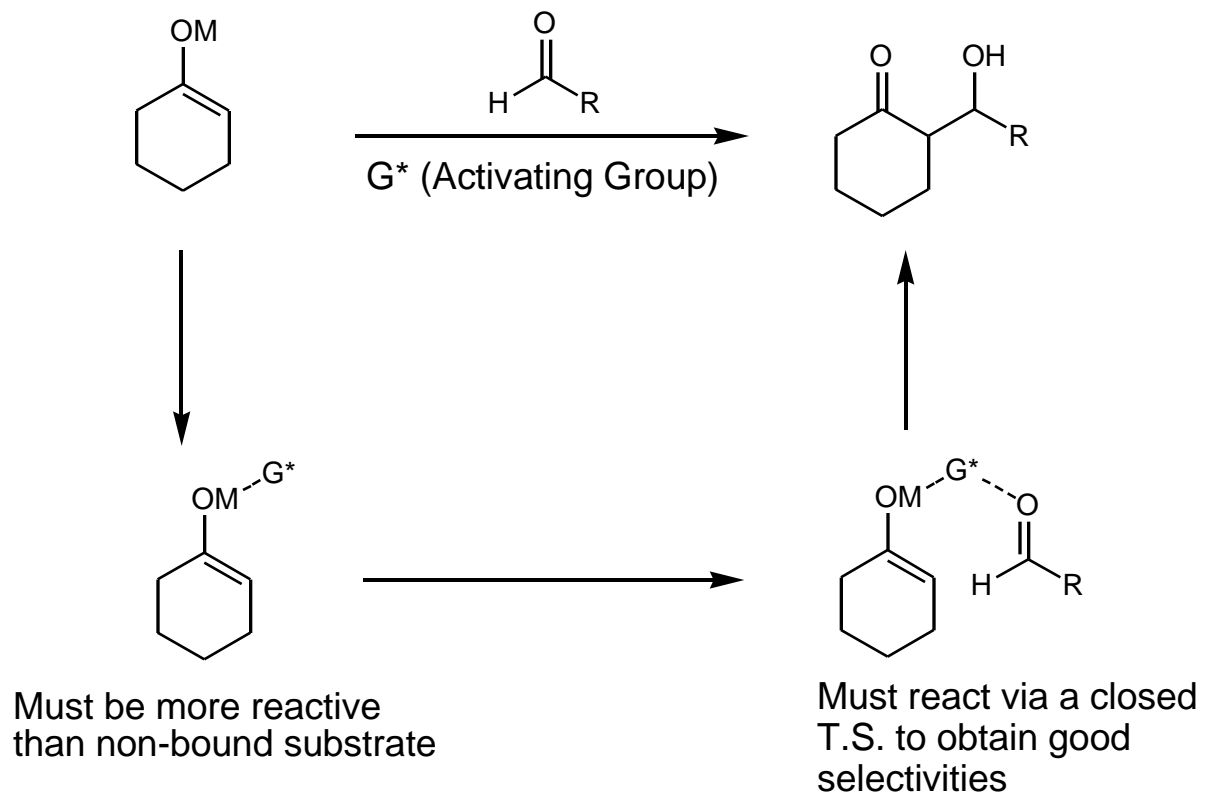


- Ligand is often deactivating
- Diastereoselectivity can be high but are often variable

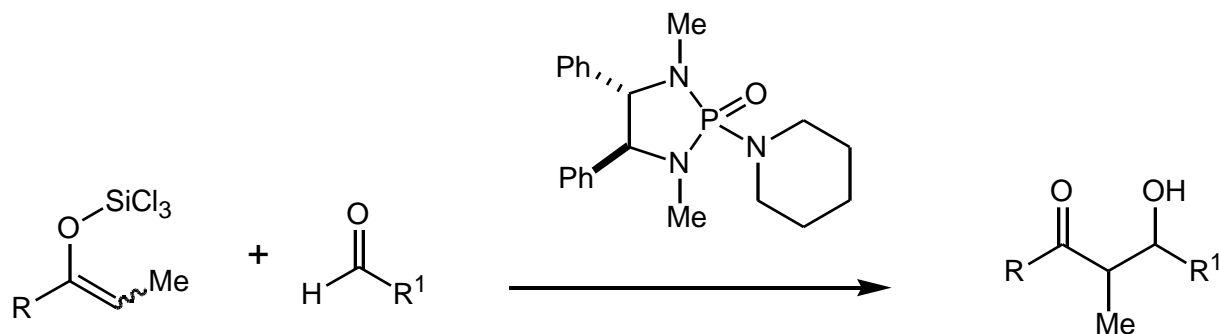


- Number of literature examples is still limited.

Lewis Base Activated Aldol



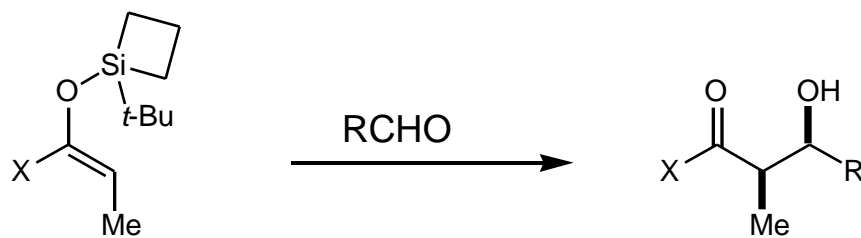
General Reaction Overview



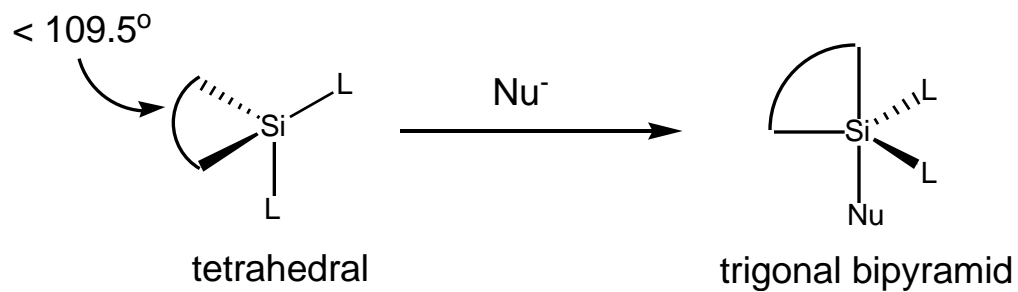
- *E* enolsilane ---> anti product
- Highly diastereoselective (up to 99:1 anti/syn)
- Highly enantioselective (up to 98% ee)

- *Z* enolsilane ---> syn product
- Highly diastereoselective (up to 18:1 syn/anti)
- Highly enantioselective (up to 98% ee)

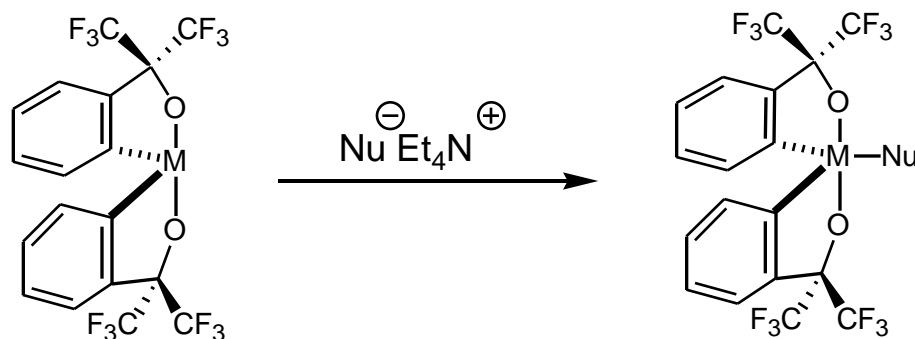
How It All Started



- Wanted to exploit the concept of “strain-release Lewis acidity”



Literature Examples

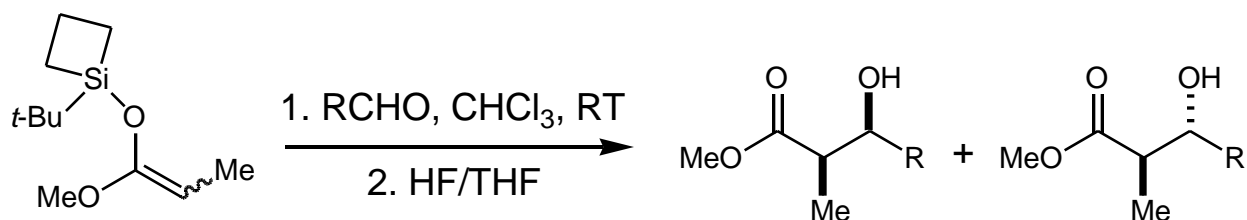


M = Si and Ge
Ligand C-O(Ge) -2.6°

- Germanium compound is able to catalyze the ene reaction while silicon is not.

Martin, J. C. *JOC*, **1981**, *46*, 1049
Denmark, S. E. *Organometallics*, **1990**, *9*, 3015

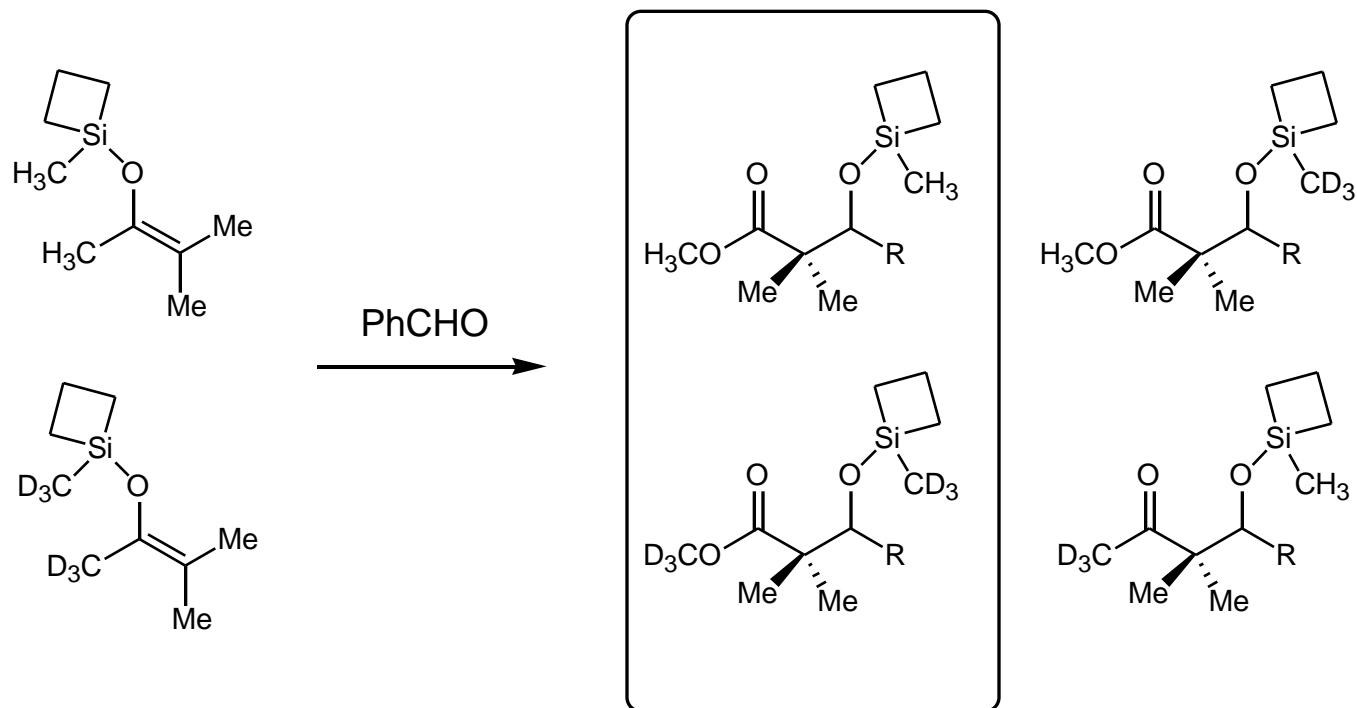
Strain-Release Catalyzed Aldol



entry	R	<i>E/Z</i>	yield	syn/anti
1	Ph	0/100	80	42/58
2	Ph	95/5	94	95/5
3	cinnamyl	89/11	95	93/7
4	<i>n</i> -pentyl	89/11	91	93/7
5	cyclohexyl	89/11	85	>99/1

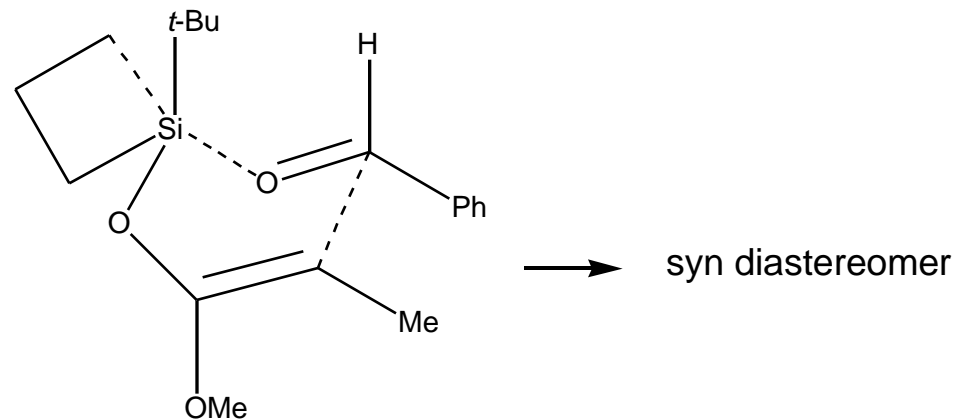
- Also worked for thiosilylketene acetals but both *E* and *Z* gave *syn* selectivity.
- Amide derived enolsilanes were unselective.

Mechanistic Studies



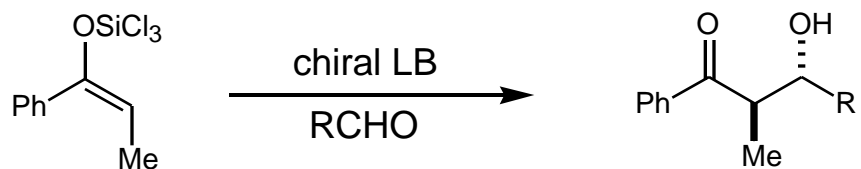
- Must go through a trigonal bipyramidal T.S. with internal transfer of silicon
- Analogous studies with KO*t*-Bu as a catalyst revealed complete deuterium scrambling. Cannot go through an octahedral T.S.

T.S. Models



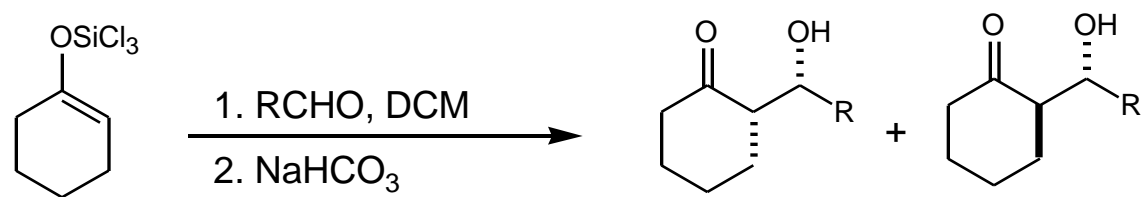
- Closed boat-like T.S.
- Pentacoordinate silicon
- Evans proposes similar T.S. in Zr based aldol (Evans, D. A. *TL*, **1980**, 21, 3975).

2nd Generation of LB Catalyzed Aldol



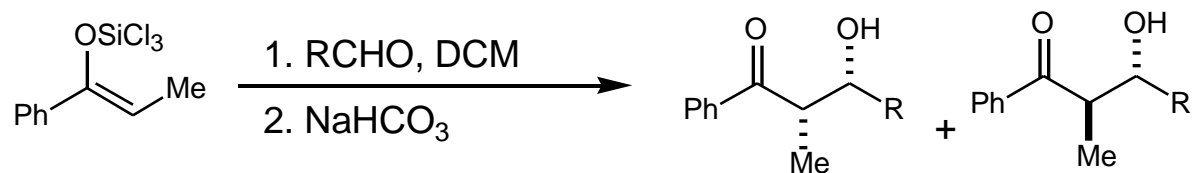
- Trichlorosilane required for sufficient Lewis acidity
- Silicon can expand valency by 2
- Simultaneous activation of Nu⁻ and E1⁺ via closed T.S.
- Facial selectivity comes from chiral base

Uncatalyzed Reactions: *E* Enolsilanes



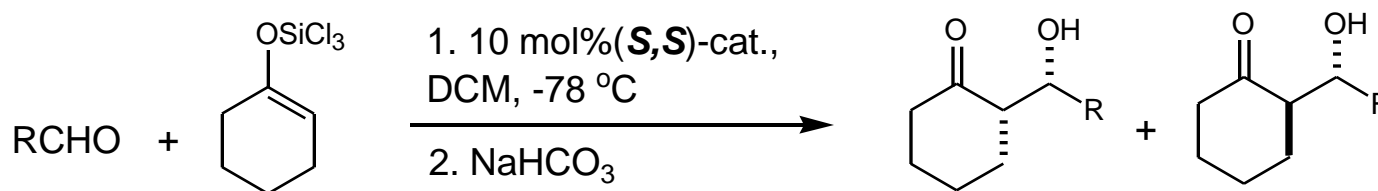
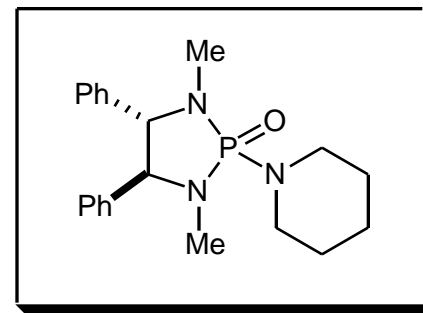
entry	R	time, hr	syn/anti	yield
1	Ph-	6	49/1	92
2	1-Naphthyl-	8	16/1	90
3	cinnamyl-	1	49/1	83
4	-methylcinnamyl-	11	5.7/1	86
5	Phenylpropargyl-	2	36/1	91
6	dihydrocinnamyl-	12	5.3/1	82
7	cyclohexyl-	36	1/1	92

Uncatalyzed Reaction: *Z* Enolsilanes



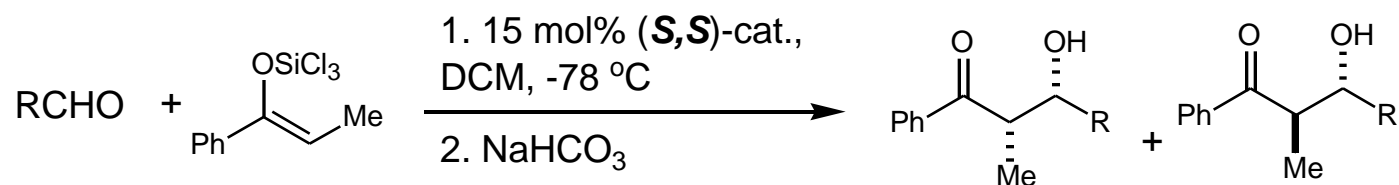
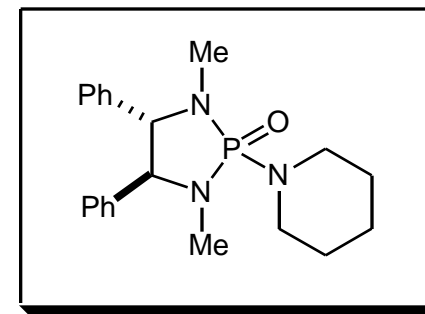
entry	R	time, hr	syn/anti	yield
1	Ph-	10	1/2.3	97
2	4-BrPh-	10	1/2.9	93
3	1-Naphtyl-	16	1/1.3	95
4	cinnamyl-	10	1/1.9	95
5	-methylcinnamyl-	12	1/2.2	64
6	crotyl-	16	1/1.9	89
7	Phenylpropargyl-	11	1/2.2	89

Catalyzed Aldol: *E* Enolsilanes



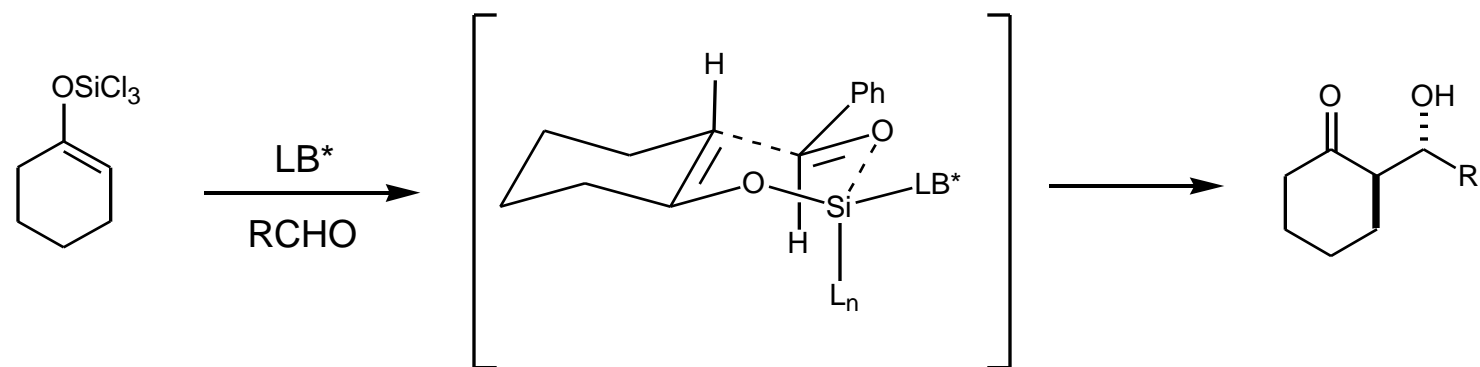
entry	R	syn/anti	anti ee%	yield, %
1	Ph-	1/61	93	95
2	1-Naphthyl-	<1/99	97	94
3	cinnamyl-	<1/99	88	94
4	-methylcinnamyl-	<1/99	92	98
5	Phenylpropargyl-	1/5.3	82	90

Catalyzed Aldol: *Z* Enolsilanes

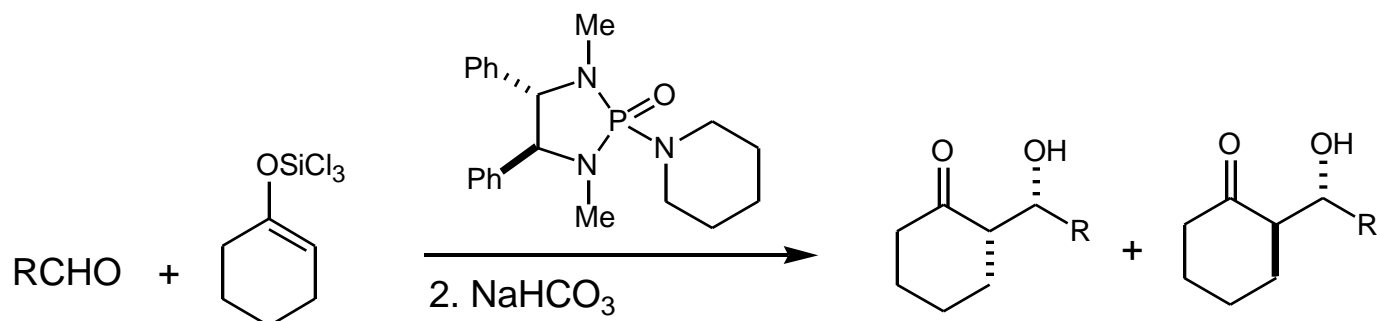


entry	R	syn/anti	syn ee%	yield, %
1	Ph-	18/1	95	95
2	4-BrPh-	12/1	96	89
3	1-Napthyl-	3/1	84	96
4	cinnamyl-	9.4/1	92	97
5	crotyl-	7/1	91	94
6	Phenylpropargyl-	1/3.5	58	92

Current T.S. Model - Too Simplistic



Mechanistic Clues



entry	loading	conc.	syn/anti	yield
1	10%	0.5	1/14	94
2	5%	0.5	1/10	90
3	2%	0.5	1/2.4	84

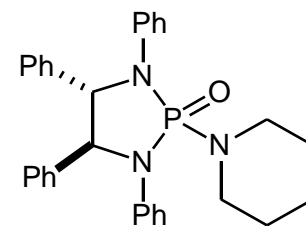
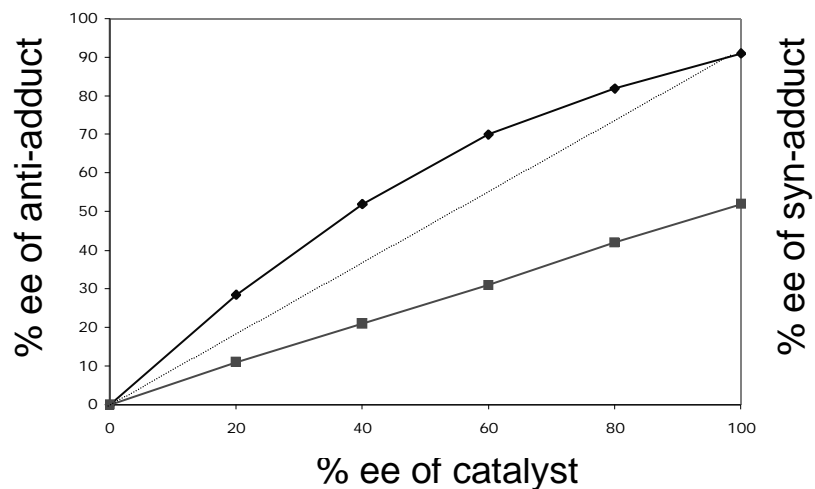
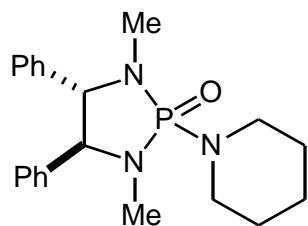
- Syn/anti ratio is highly dependent on catalyst loading
- Simple change in silicon valency cannot account for rate acceleration
- Bulkier catalysts lowers and even switches diastereoselectivities

Denmark, S. E. *JACS*, **1998**, *120*, 12990

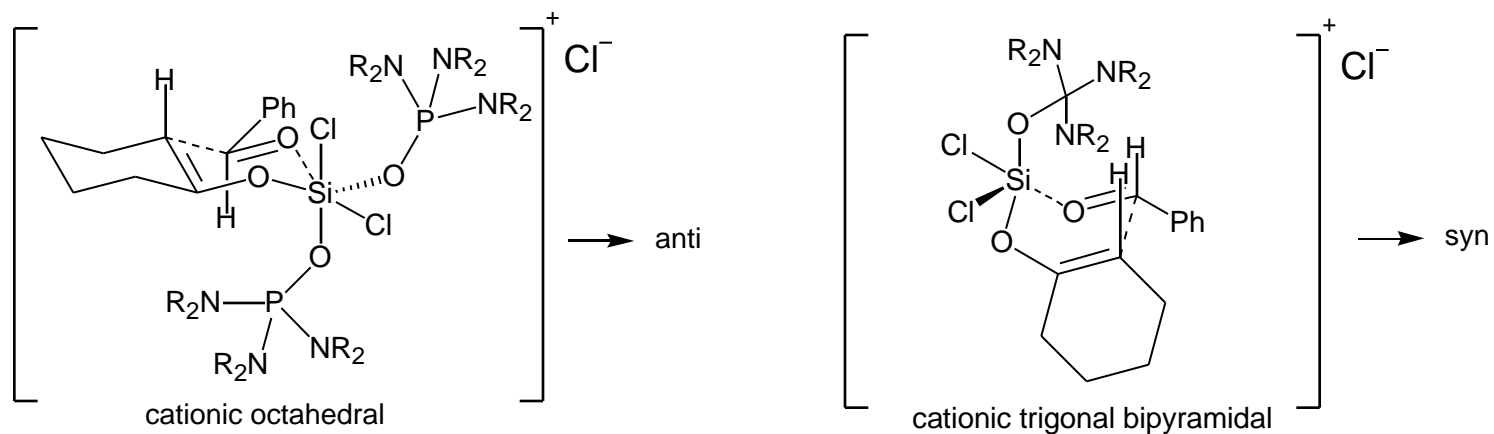
Denmark, S. E. *JACS*, **1999**, *121*, 4982

Mechanistic Proposal

- Two competitive pathways, one for the formation of each diastereomer
- Anti diastereomer dominates with less bulky ligands and higher conc.
- Syn diastereomer dominates with more bulky ligands and lower conc.

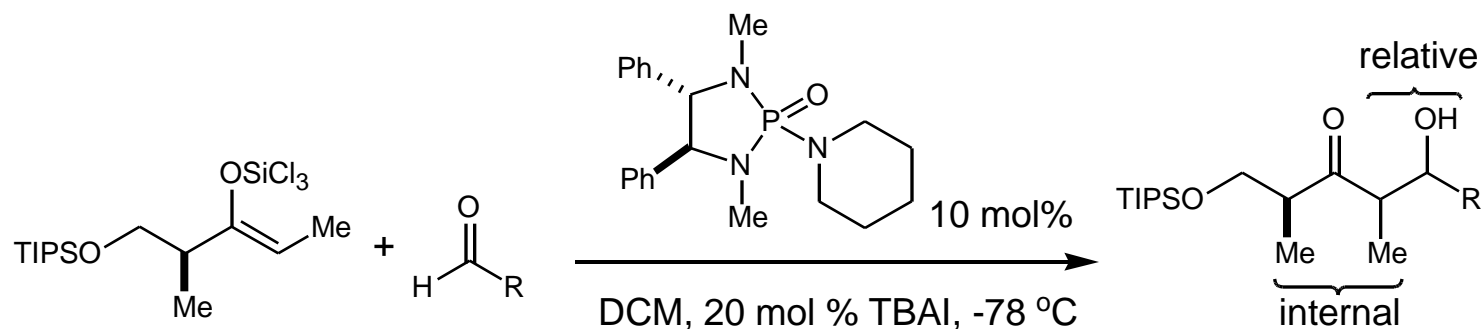


Unified T.S. Model



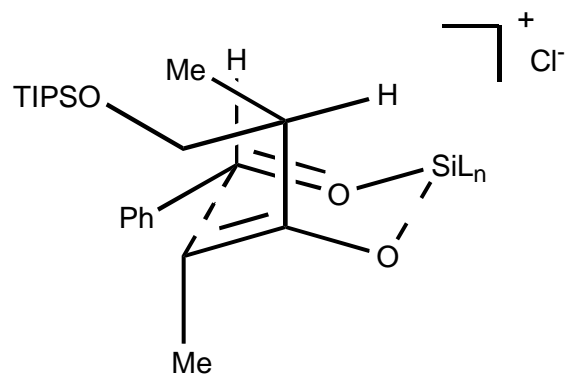
- Ligand binding causes Cl^- to dissociate:
- Bu_4NCl inhibits reaction because of the common ion effect
- Bu_4NOTf and TBAI accelerate reaction rates by increasing ionic strength

Double Stereodifferentiating Reactions I

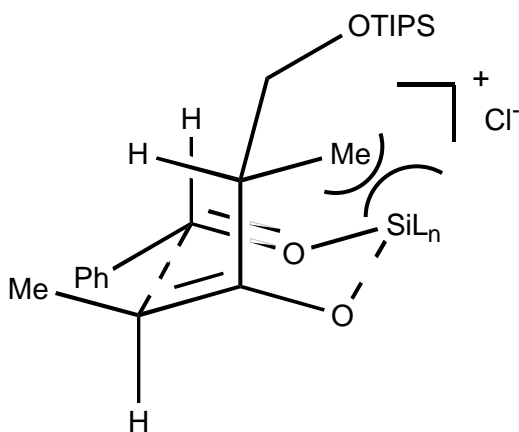


entry	R	catalyst	relative dr (syn/anti)	internal dr (syn/anti)
1	Ph-	(<i>R,R</i>)	53/1	24/1
2	Ph-	(<i>S,S</i>)	32/1	1/8
3	1-Naphthyl-	(<i>R,R</i>)	14/1	89/1
4	1-Naphthyl-	(<i>S,S</i>)	14/1	1/17
5	cinnamyl-	(<i>R,R</i>)	9/1	14/1
6	cinnamyl-	(<i>S,S</i>)	15/1	1/6
7	crotyl-	(<i>R,R</i>)	>50/1	15/1
8	crotyl-	(<i>S,S</i>)	>50/1	1/5
9	tiglyl-	(<i>R,R</i>)	13/1	13/1
10	tiglyl-	(<i>S,S</i>)	19/1	1/5
11	Ph	achiral	27/1	5/1

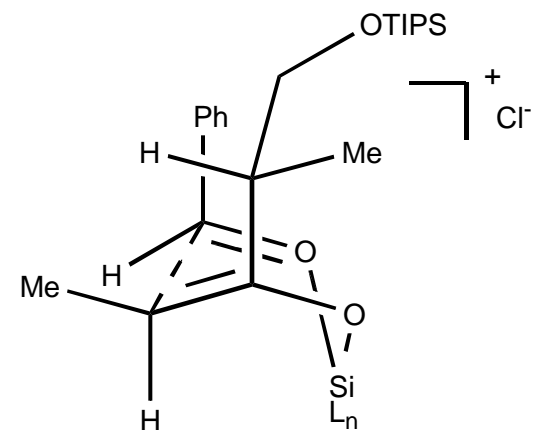
T.S. Models



- Minimizes steric interactions between substituents on enolate and bulky ligands on silicon
- Anti diastereomer may be formed via a boat-like T.S.

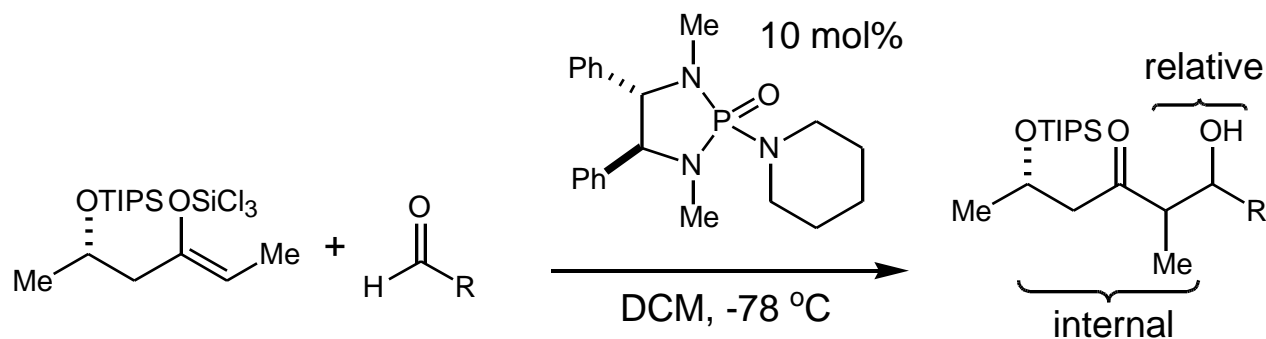


- $A_{1,3}$ strain is minimized but severe non-bonding interactions exist between substrate and ligands on silicon.



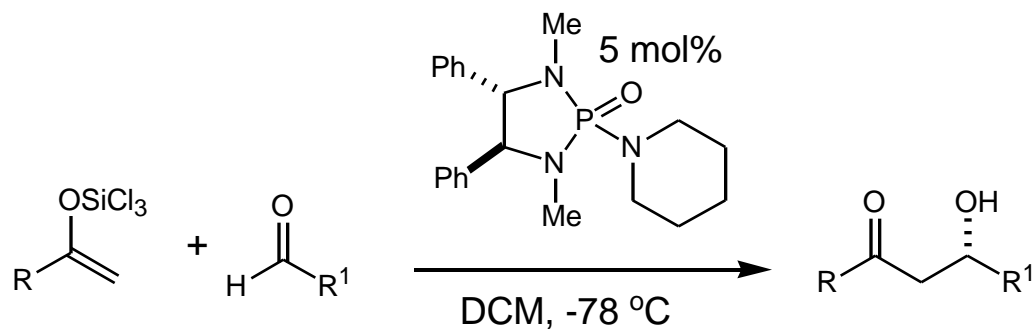
- Non-bonding interactions arising from chair T.S. can give rise to boat-like T.S.

Double Stereodifferentiating Reactions II



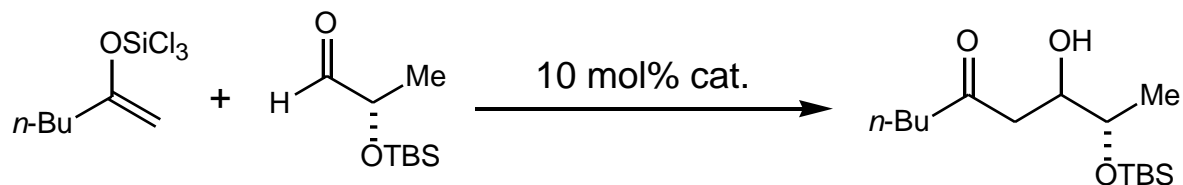
entry	R	catalyst	relative dr (syn/anti)	internal dr (syn/anti)
1	Ph-	(<i>R,R</i>)	30/1	16/1
2	Ph-	(<i>S,S</i>)	26/1	1/10
3	1-Naphthyl-	(<i>R,R</i>)	17/1	30/1
4	1-Naphthyl-	(<i>S,S</i>)	18/1	1/10
5	cinnamyl-	(<i>R,R</i>)	>50/1	10/1
6	cinnamyl-	(<i>S,S</i>)	>50/1	1/8
7	crotyl-	(<i>R,R</i>)	28/1	7/1
8	crotyl-	(<i>S,S</i>)	37/1	1/6
9	tiglyl-	(<i>R,R</i>)	>50/1	3/1
10	tiglyl-	(<i>S,S</i>)	>50/1	1/3
11	Ph	achiral	29/1	1.4/1

Methyl Ketones as Nucleophiles

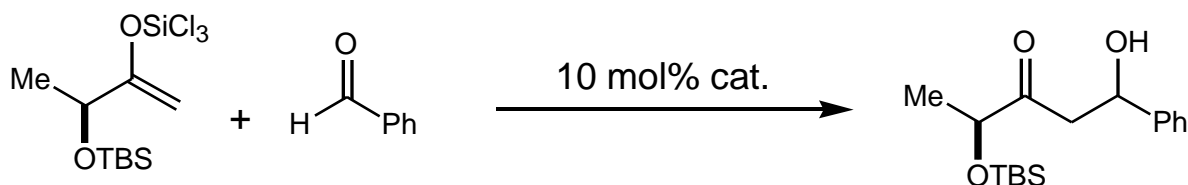


entry	R	R ¹	ee %	yield %
1	Me-	Ph-	87	98
2	<i>n</i> -Bu-	Ph-	85	98
3	<i>i</i> -Pr-	Ph-	81	97
4	Ph-	Ph-	49	93
5	TBSOCH ₂ -	Ph-	86	94
6	<i>n</i> -Bu-	cinnamyl-	84	94
7	<i>n</i> -Bu-	-methylcinnamyl-	91	95
8	<i>n</i> -Bu-	1-naphthyl-	86	92
9	<i>n</i> -Bu-	cyclohexyl-	89	79
10	<i>n</i> -Bu-	<i>t</i> -Bu-	92	81

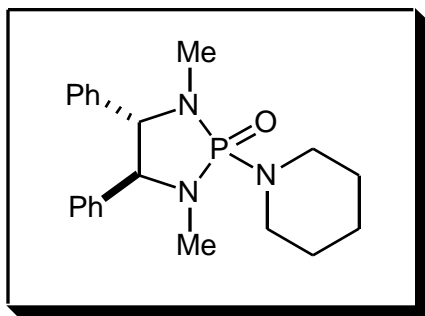
Double Stereodifferentiating Reactions



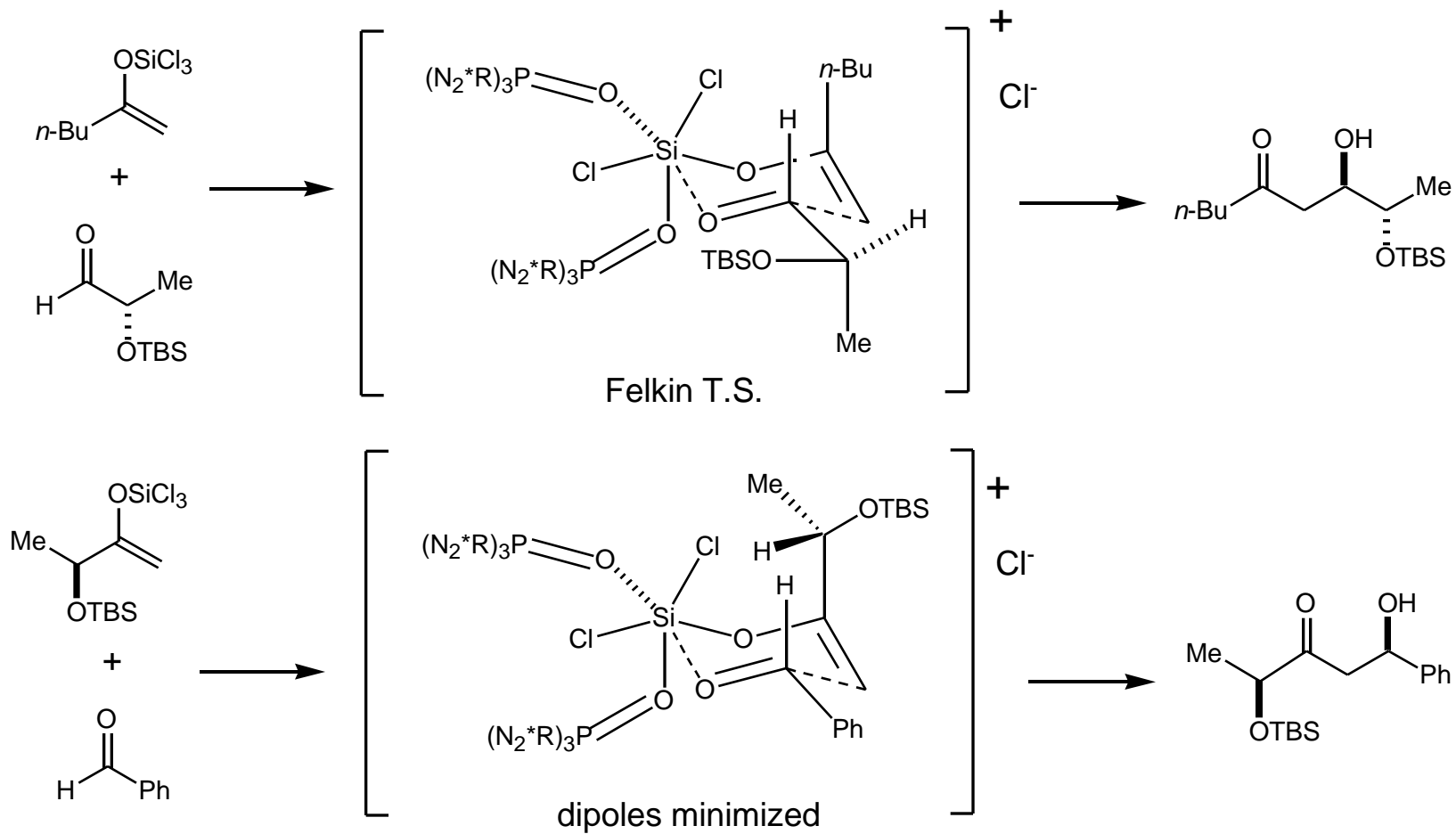
(*R,R*) - 1/15.6 syn/anti, 56% yield
(*S,S*) - 2.7/1 syn/anti, 47% yield
achiral - 1/1.3 syn/anti, 41% yield



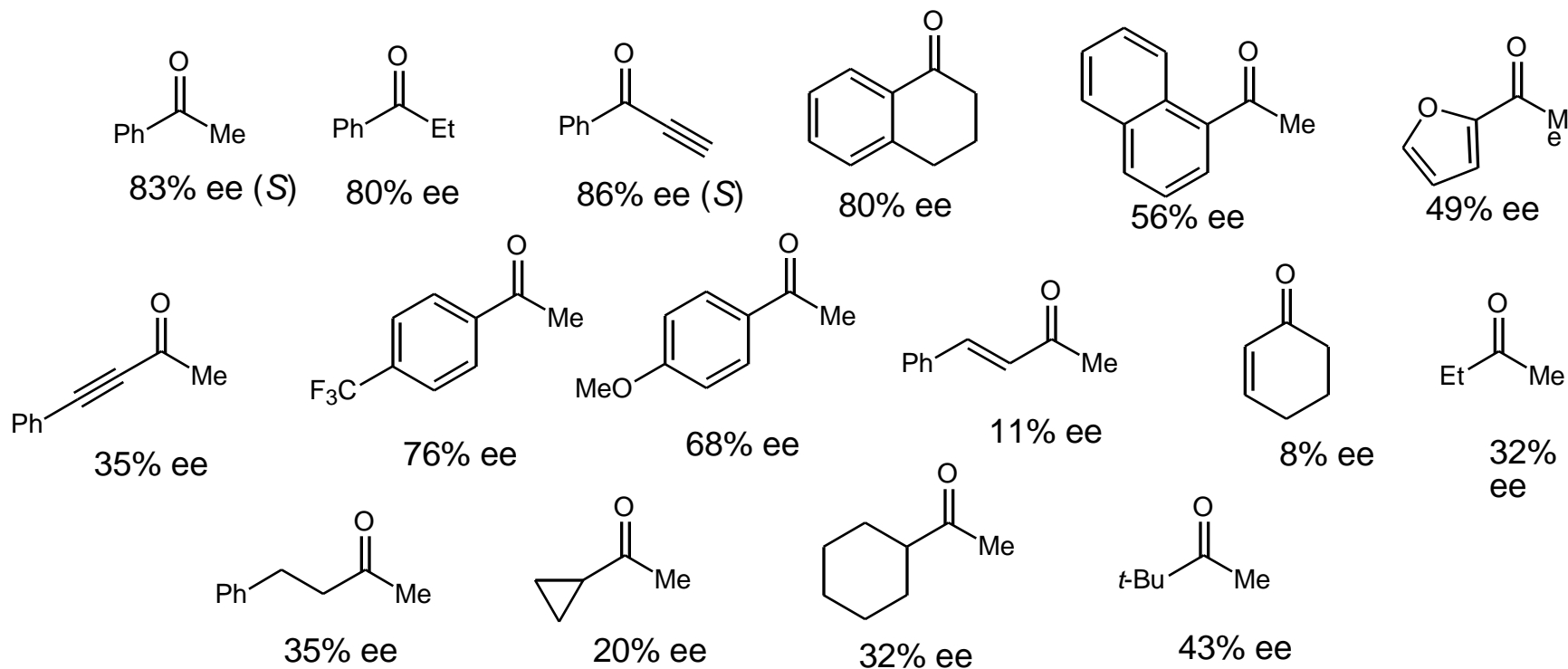
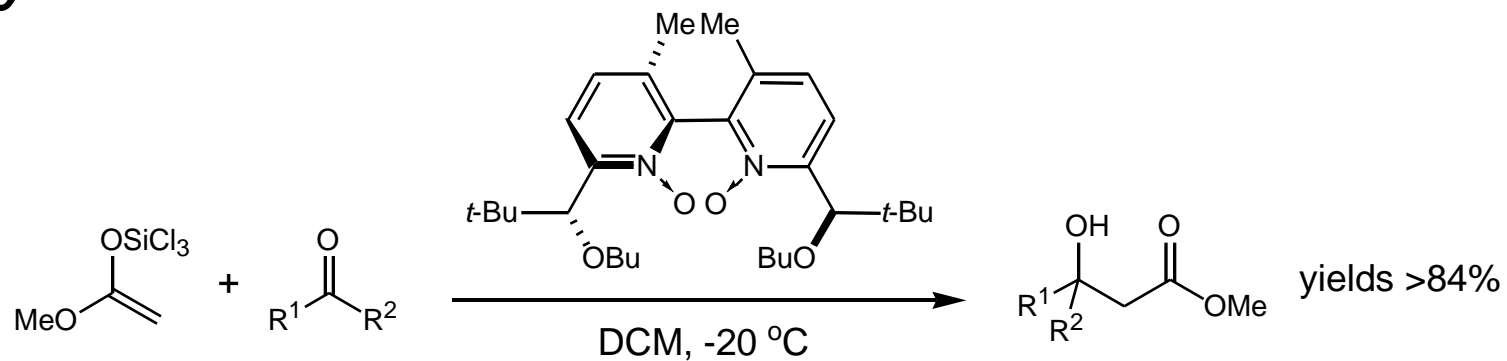
(*R,R*) - 73/1 syn/anti, 85% yield
(*S,S*) - 1.5/1 syn/anti, 85% yield
achiral - 1.2/1 syn/anti, 81% yield



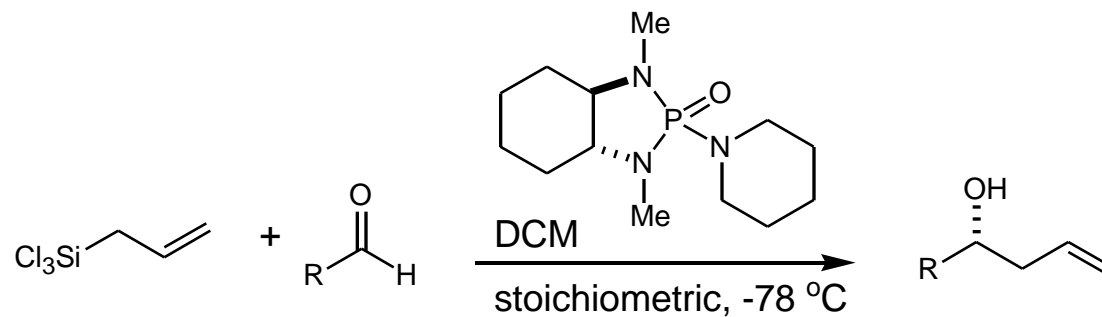
T.S. Models



Silyl Ketene Acetals

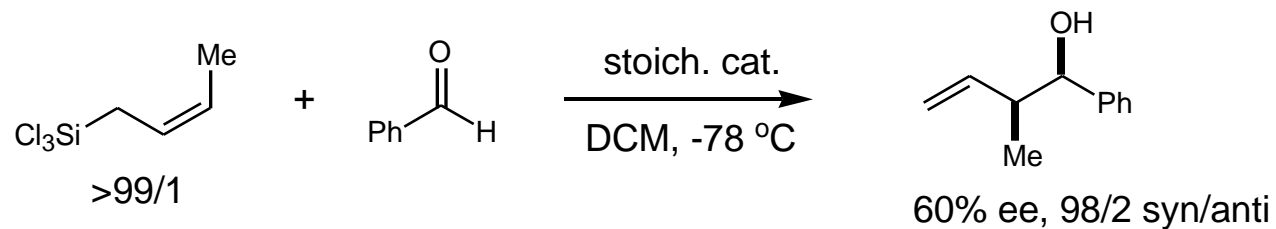
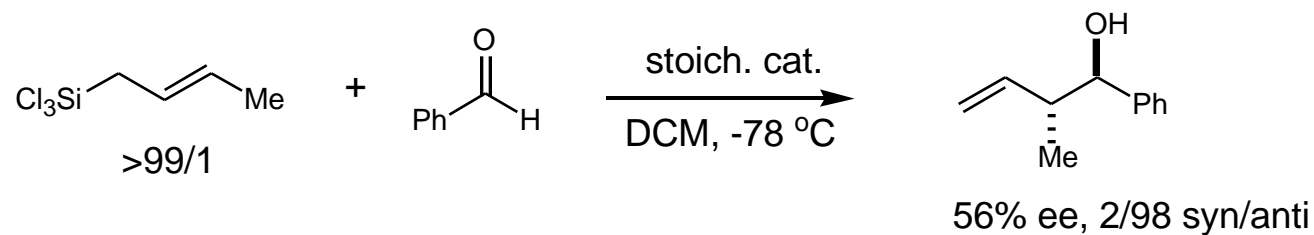
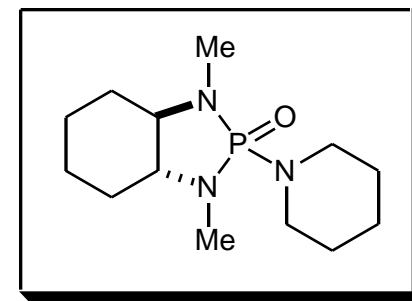


Asymmetric Allylation

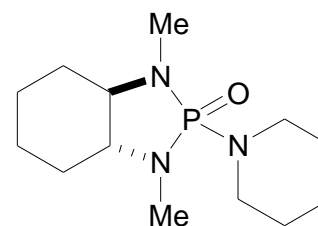
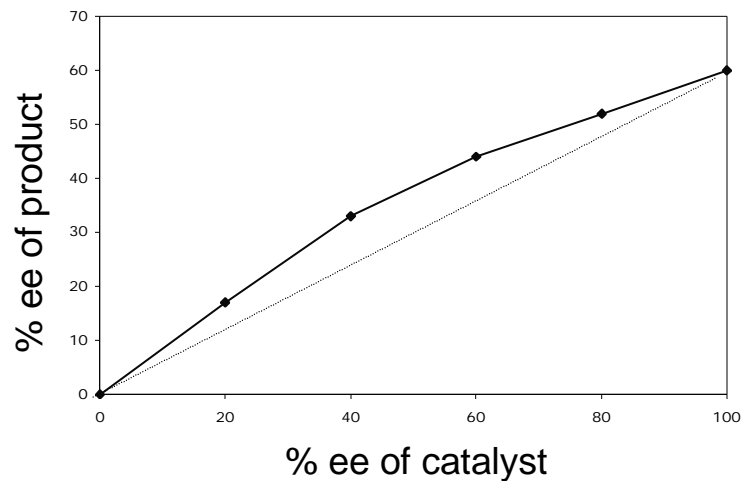


entry	R	ee, %	yield
1	Ph-	60	81
2	<i>o</i> -tol-	65	81
3	4-NO ₂ Ph-	21	76
4	4-MeOPh-	50	80
5	4-NMe ₂ Ph-	33	69
6	cinnamyl-	38	67

Diastereoselective Allylations

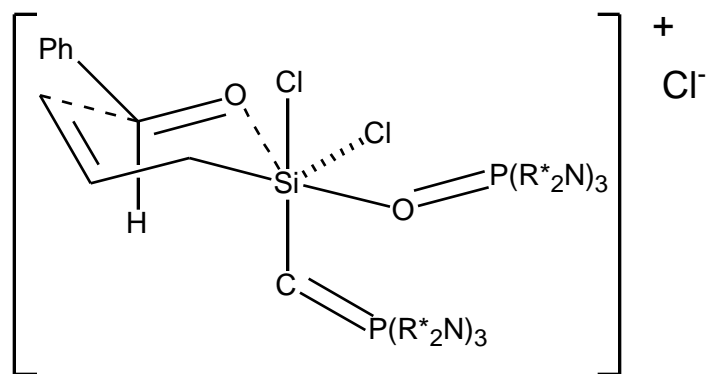


Mechanistic Studies

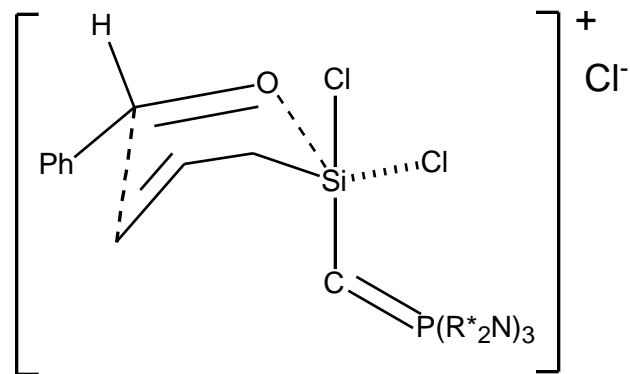


- 1st order in aldehyde
- 1st order in allylsilane
- 1.77th order in catalyst due to competing pathways involving 1 or 2 phosphoramidates bound to silicon.

T.S. Models



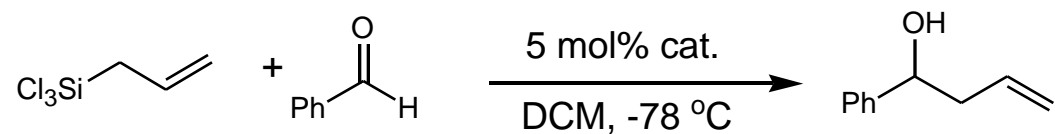
cationic octahedral



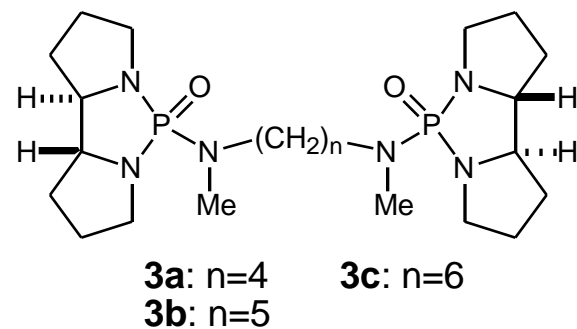
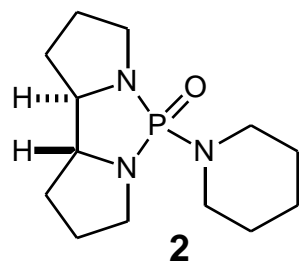
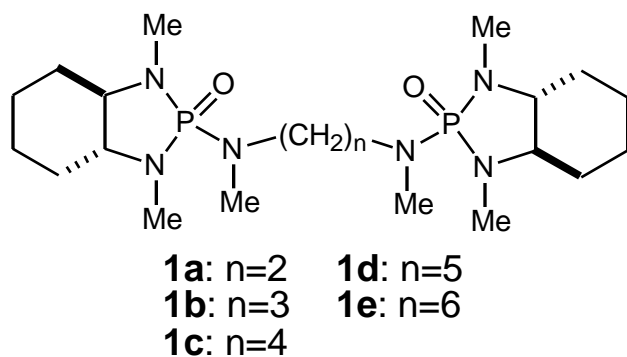
cationic trigonal bipyramidal

- Facial selectivity is lower because the cationic octahedral T.S. gives the opposite facial selectivity as the cationic trigonal bipyramidal T.S.

Bisphosphoramides

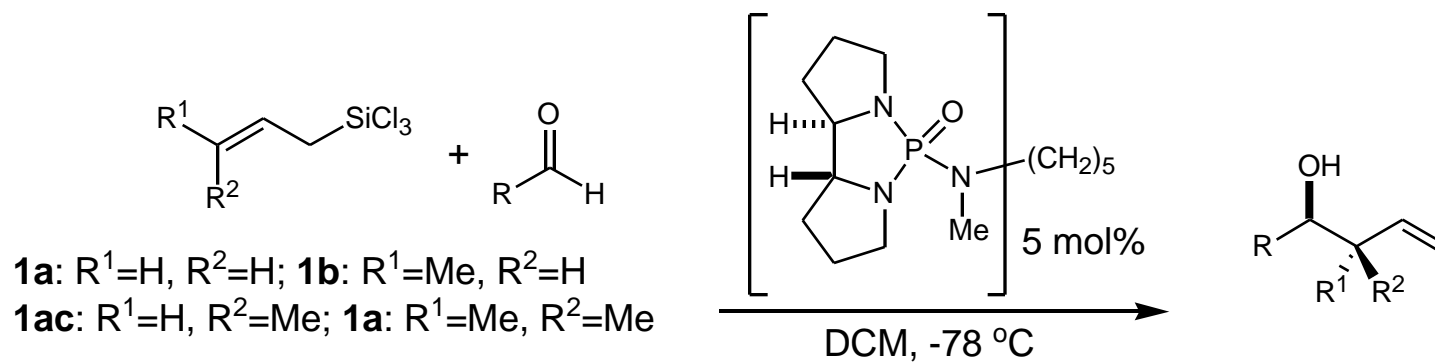


entry	cat.	ee %	config.	entry	cat.	ee%	config.
1	1a	0	<i>R</i>	6	2	56	<i>S</i>
2	1b	35	<i>R</i>	7	3a	18	<i>S</i>
3	1c	17	<i>R</i>	8	3b	87	<i>S</i>
4	1d	65	<i>R</i>	9	3c	67	<i>S</i>
5	1e	46	<i>R</i>				



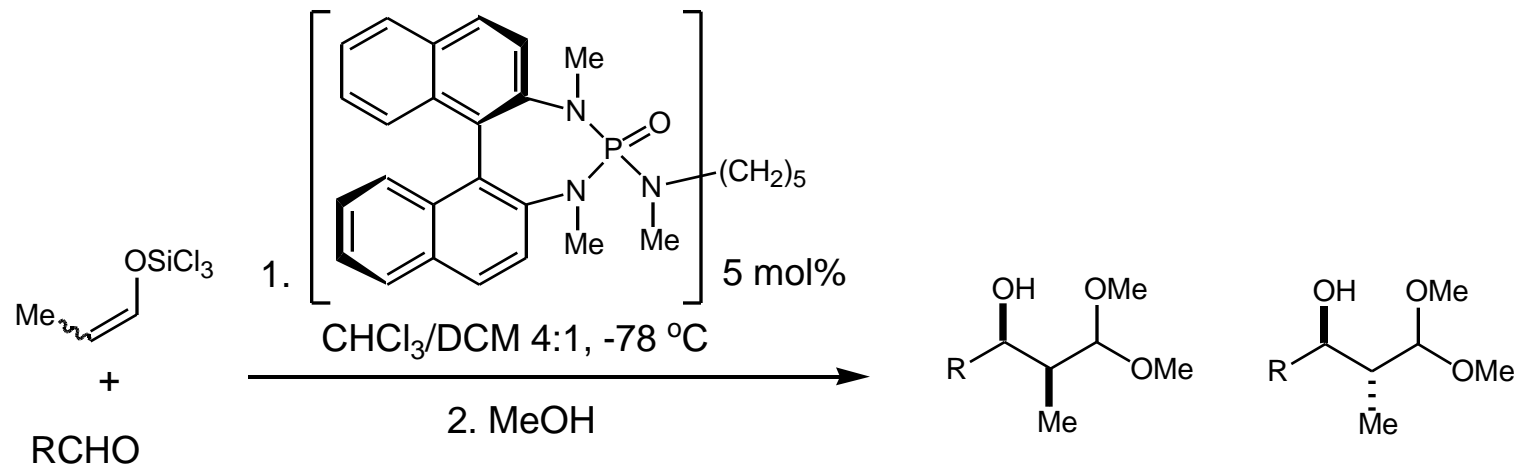
Denmark, S. E. *JACS*, **2000**, 122, 12021
 Denmark, S. E. *JACS*, **2001**, 123, 9488

Allylations Revisited



entry	silanes	R	syn/anti	ee %	yield %
1	1a	Ph-		87	85
2	1a	2-naphthyl		87	92
3	1a	cinnamyl-		81	86
4	1b	Ph-	1/99	86	82
5	1b	cinnamyl-	1/99	80	57
6	1c	Ph-	99/1	94	89
7	1c	cinnamyl-	99/1	88	78
8	1d	Ph-		96	89
9	1d	cinnamyl-		88	70

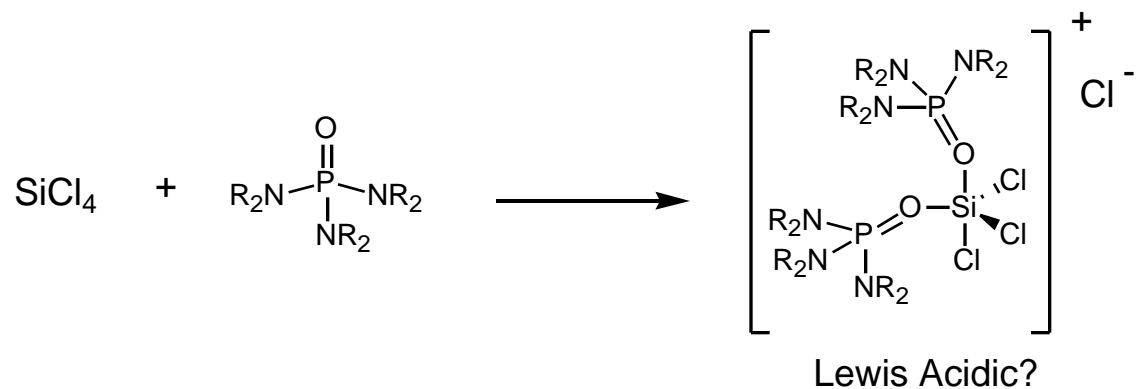
Crossed-Aldol Reactions of Aldehydes



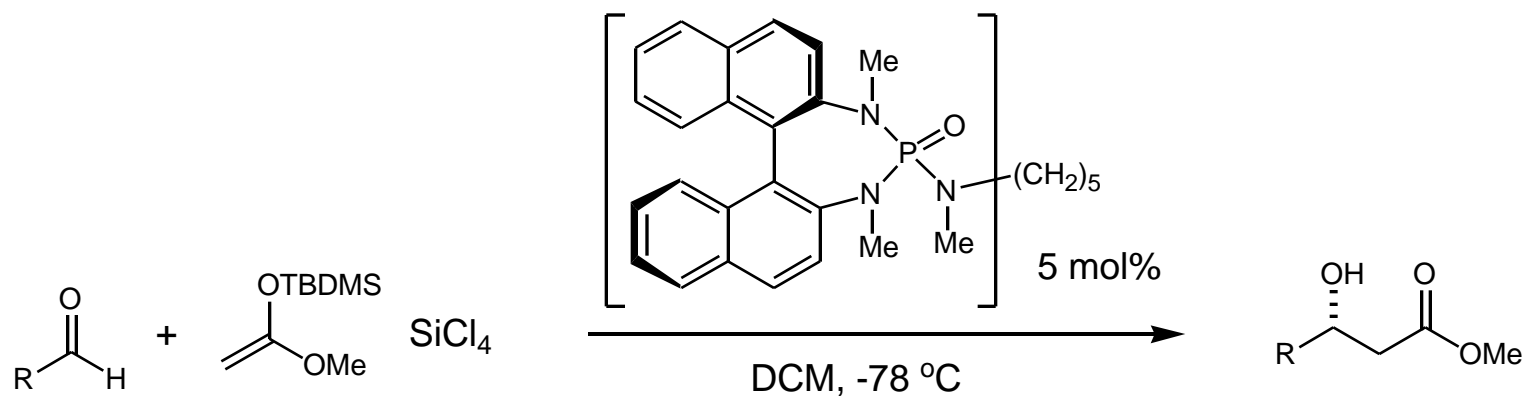
entry	enolate	R	syn/anti	ee %	yield
1	<i>Z</i>	Ph-	98/2	81	95
2	<i>E</i>	Ph-	1/99	59	97
3	<i>Z</i>	cinnamyl-	99/1	42	86
4	<i>E</i>	cinnamyl-	1/99	26	88
5	<i>Z</i>	crotyl-	99/1	5	85
6	<i>E</i>	crotyl-	2/98	52	91
7	<i>Z</i>	phenylpropargyl-	98/2	7	98
8	<i>E</i>	phenylpropargyl-	2/98	76	99
9	<i>Z</i>	dihydrocinnamyl-	95/5	8	47
10	<i>E</i>	dihydrocinnamyl-	1/99	66	79

3rd Generation of LB Catalyzed Aldol

- Limitation of current methodology is the ability to prepare the required trichlorosilanes.

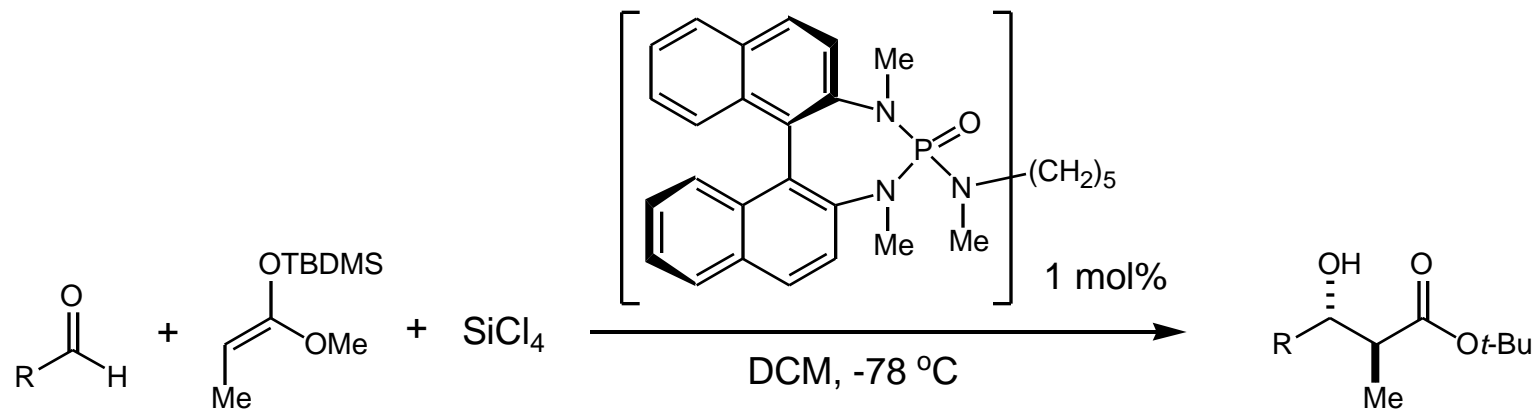


Acetate Aldols



entry	R	ee	yield	entry	R	ee	yield
1	Ph-	93	97	6	4-CF ₃ Ph-	91	97
2	1-naphthyl-	80	98	7	cinnamyl-	94	95
3	2-naphthyl-	94	98	8	2-furyl-	87	94
4	4-MePh-	94	97	9	cyclohexyl-	88	86
5	4-MeOPh-	97	97	10	dihydrocinnamyl-	81	72

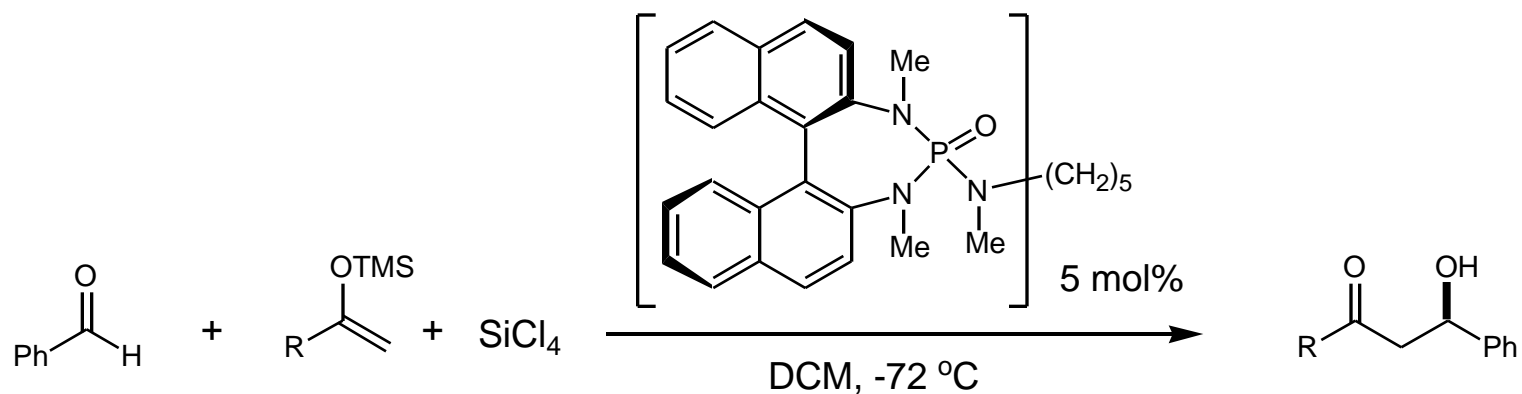
Propanate Aldols



entry	R	dr	ee	yield
1	Ph-	99/1	98	93
2	1-naphthyl-	96/4	94	98
3	2-naphthyl-	>99/1	98	95
4	4-MeOPh-	>99/1	98	88
5	4-CF ₃ Ph-	>99/1	92	93
6	cinnamyl-	>99/1	98	98
7	phenylpropargyl-	96/4	68	92
8	dihydrocinnamyl ^a -	93/7	89	55

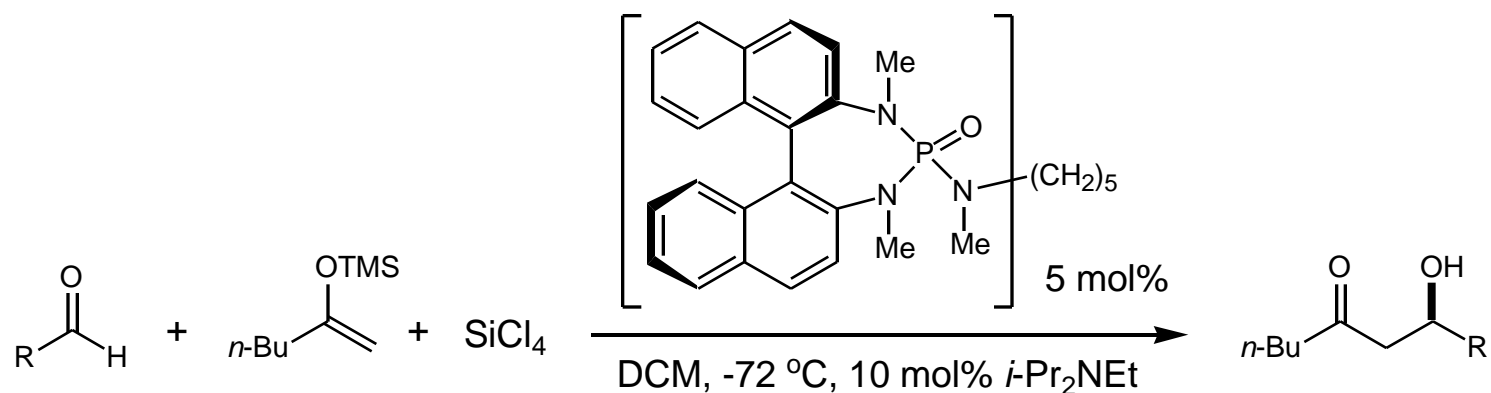
^aReaction run with 0.1 equiv of TBAI

Aldol Additions with Methyl Ketones I



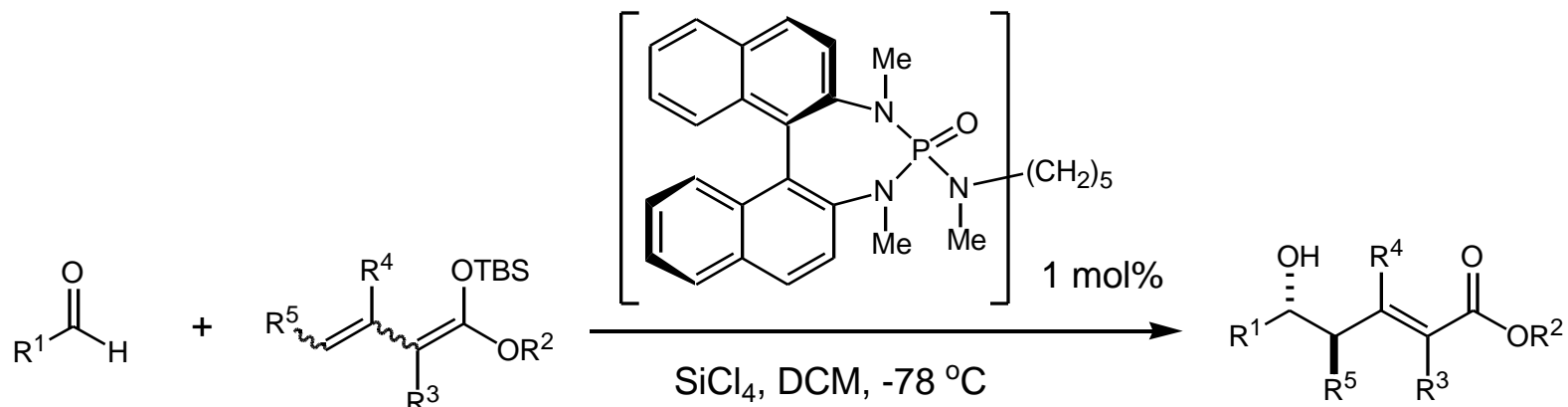
entry	R	ee %	yield
1	<i>n</i> -Bu-	98	81
2	<i>sec</i> -Bu-	98	70
3	<i>t</i> -Bu-	—	—
4	<i>i</i> -Pr-	99	72
5	Ph-	94	76

Aldol Additions with Methyl Ketones II



entry	R	ee %	yield
1	Ph-	99	99
2	cinnamyl-	99	98
3	1-naphthyl-	92	95
4	2-naphthyl-	99	92
5	4-MeOPh-	99	98
6	4-CF ₃ Ph-	99	96
7	2-furyl-	90	88
8	2-thiophenyl-	89	79
9	dihydrocinnamyl-	—	—

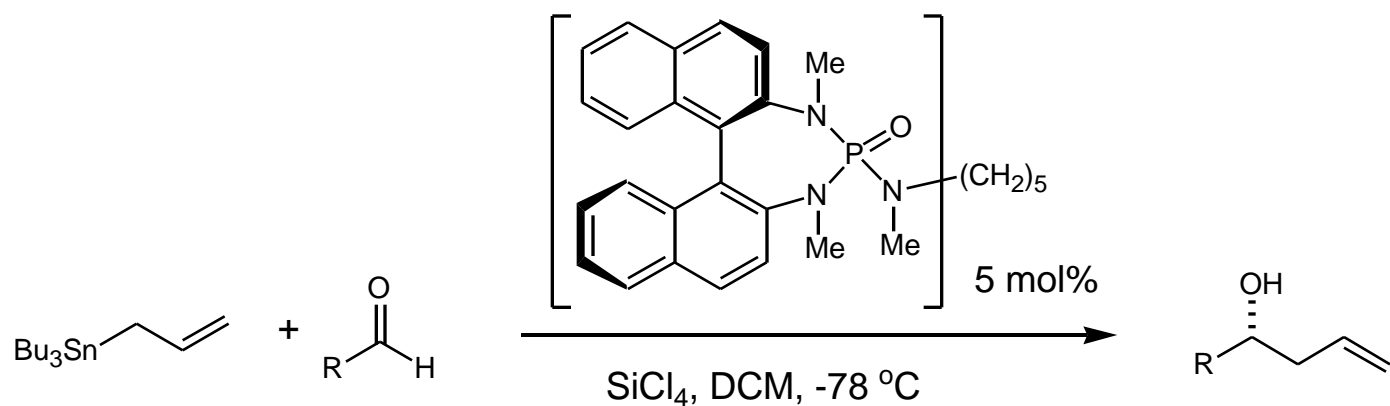
Vinylogous Aldol



entry	R ¹	R ²	R ³	R ⁴	R ⁵	ee %	yield
1	Ph-	Et-	H-	H-	H-	98	89
2	cinnamyl-	Et-	H-	H-	H-	96	84
3	dihydrocinnamyl-	Et-	H-	H-	H-	90	68
4	Ph-	Me-	Me-	H-	H-	99	93
5	cinnamyl-	Me-	Me-	H-	H-	99	88
6	Ph	Et-	H-	Me-	H-	92	91
7	cinnamyl-	Et-	H-	Me-	H-	88	97
8	dihydrocinnamyl-	Et-	H-	Me-	H-	95	73
9	Ph-	<i>t</i> -Bu-	H-	H-	Me ^a -	89	92
10	cinnamyl-	<i>t</i> -Bu-	H-	H-	Me ^a -	82	71

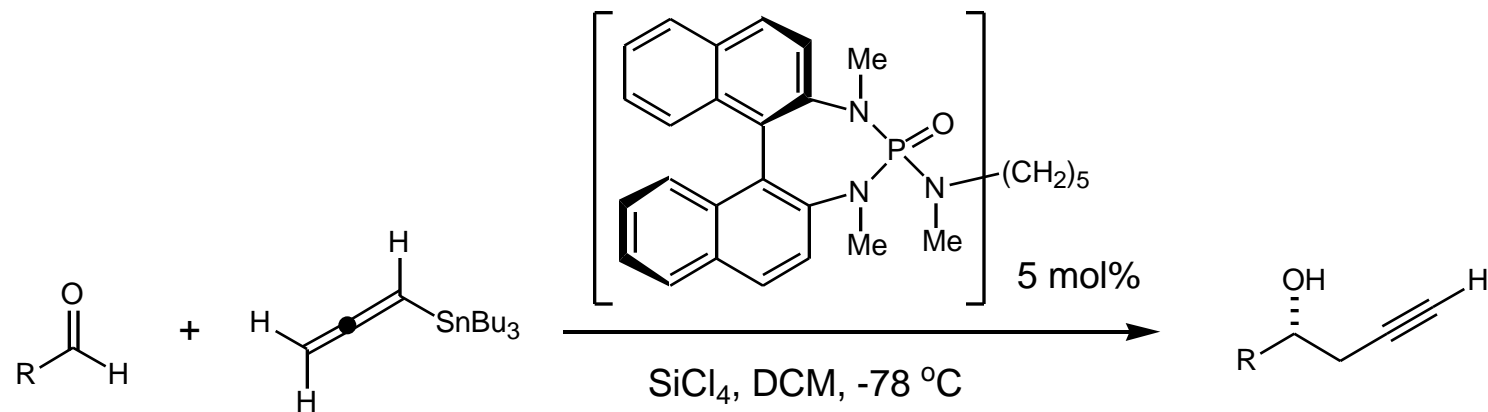
^aDiastereoselectivity >99/1

Allylation Revisited, Again



entry	R	ee %	yield %
1	Ph-	94	91
2	4-NO ₂ Ph-	83	90
3	cinnamyl-	65	91
4	phenylpropargyl-	22	92
5	1-naphthyl-	94	94
6	2-naphthyl-	93	92
7	2-furyl-	62	65

Propargylation



R = Ph	81% (97% ee (<i>R</i>))
R = cinnamyl	90% (87% ee (<i>R</i>))
R = 2-naphthyl	95% (93% ee (<i>R</i>))